

Autonomous Workflows for an Accelerated Design of 1D Nanostructures and Battery Electrodes

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Section for Autonomous Materials Discovery

Department of Energy Conversion and Storage (DTU Energy)

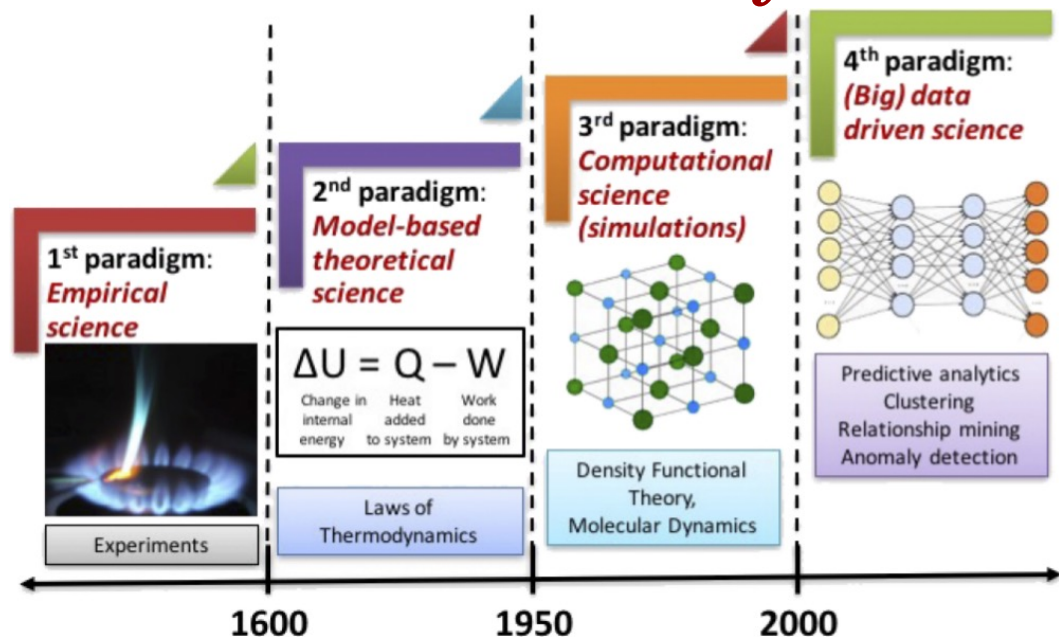
Technical University of Denmark

Challenges for Sustainable Development



To solve many of these issues, we need to discover novel materials

Materials Discovery – A Bit of History



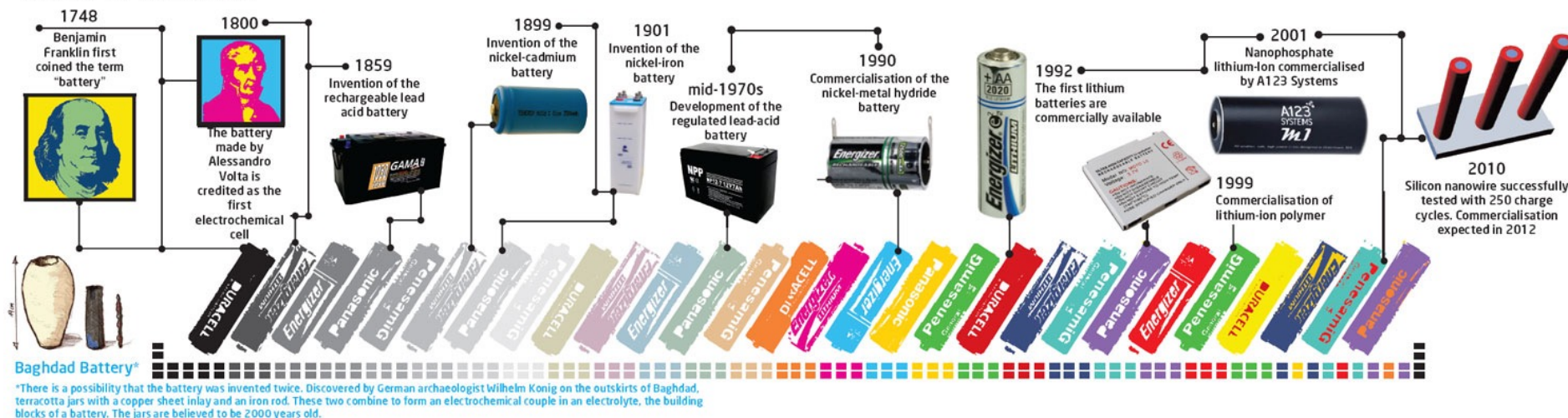
Angrawal and Choudhary, APL Materials 4, 053208 (2016)

Jose and Ramakrishna, Applied Materials Today 10, 127 (2018)

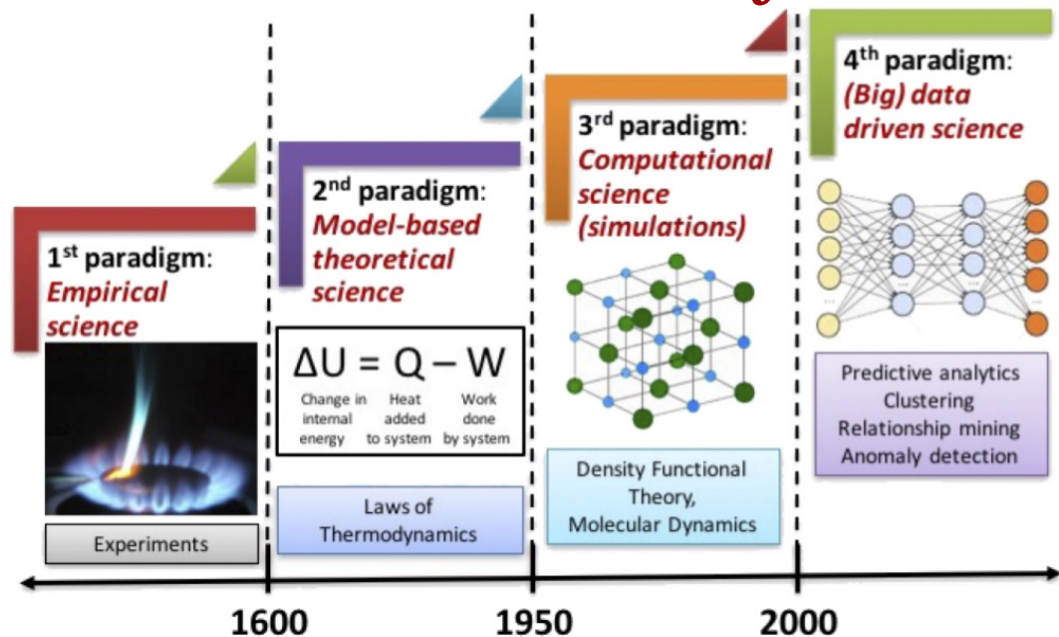
<https://www.upsbatterycenter.com>

Pancorbo, DOI: 10.13140/RG.2.2.14706.04806 (2018)

HISTORY OF THE BATTERY



Materials Discovery – A Bit of History



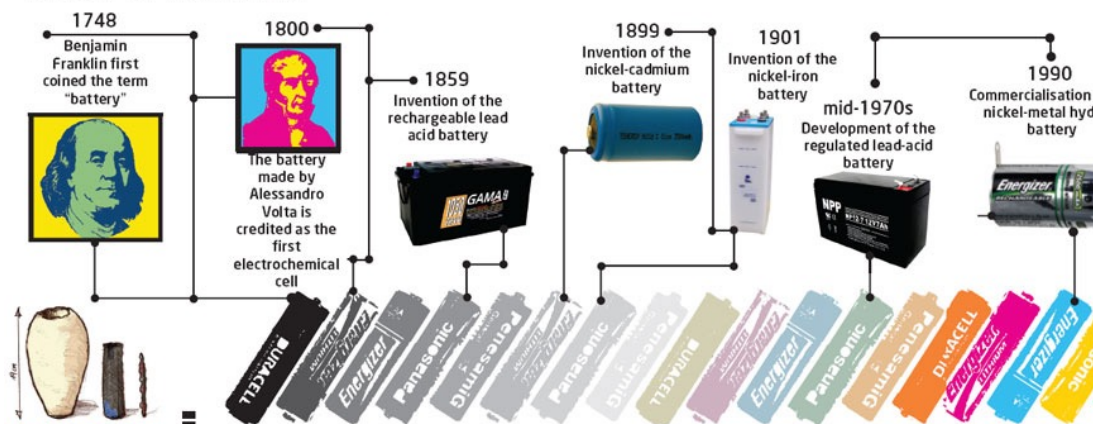
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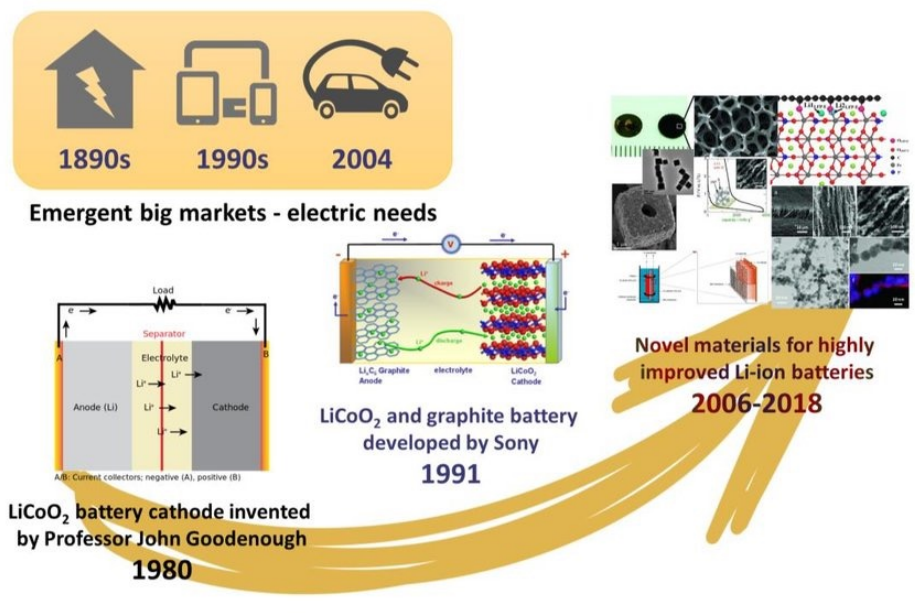
Pancorbo, DOI: 10.13140/RG.2.2.14706.04806 (2018)

HISTORY OF THE BATTERY



Baghdad Battery*

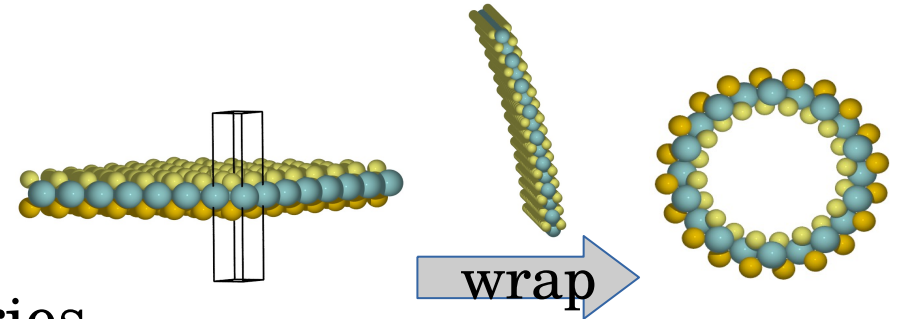
*There is a possibility that the battery was invented twice. Discovered by German archaeologist Wilhelm König on the outskirts of Baghdad, terracotta jars with a copper sheet inlay and an iron rod. These two combine to form an electrochemical couple in an electrolyte, the building blocks of a battery. The jars are believed to be 2000 years old.



Autonomous Workflows for Materials Discovery – Outline

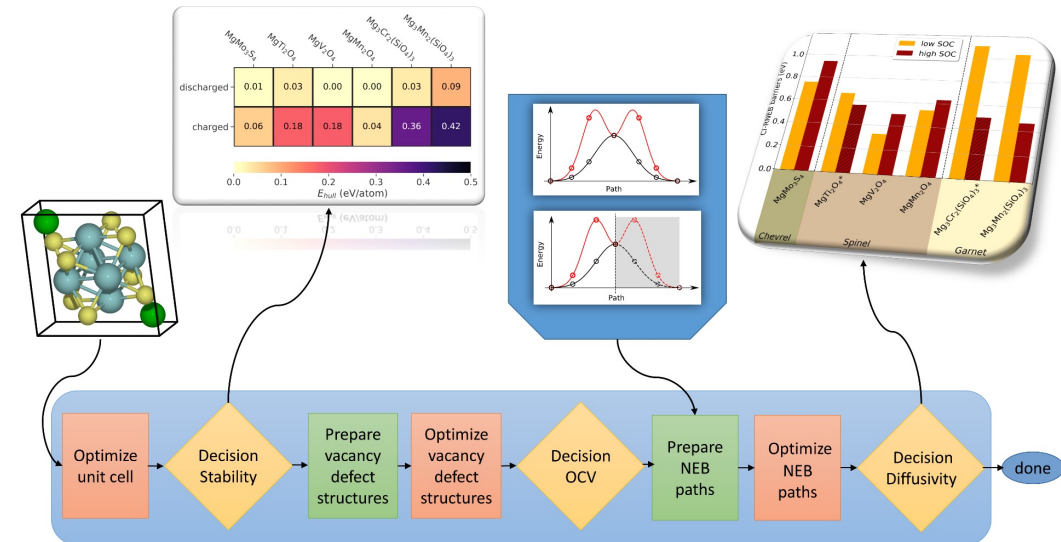
1) From 2D sheets to 1D inorganic nanotubes

Vision: use shape as a new degree of freedom to discover new materials

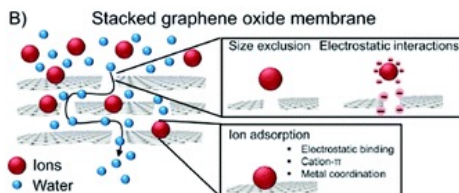
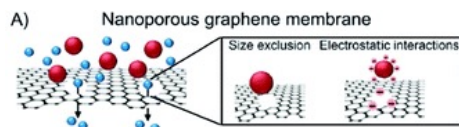


2) New electrodes for intercalation secondary batteries

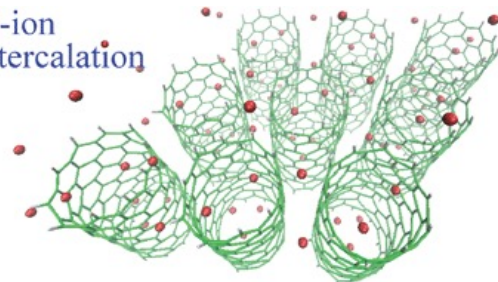
3) Workflows for interfaces – is it a dream?



From 2D Databases to Inorganic 1D Nanotubes



Li-ion
Intercalation



Possible applications:

- water filtration/membranes
- anode materials for batteries
- catalysts
- health (drug delivery)
- nanofriction
- electronics
- magnetic effects

Computational 2D Materials Database (C2DB)

If you are using data from this database in your research, please cite the following paper:

The Computational 2D Materials Database: High-Throughput Modeling and Discovery of Atomically Thin Crystals

Sten Hastrup, Mikkel Strange, Mohnish Pandey, Thorsten Deilmann, Per S. Schmidt, Nicki F. Hinsche, Morten N. Gjerding, Daniele Torelli, Peter M. Larsen, Anders C. Riis-Jensen, Jakob Gath, Karsten W. Jacobsen, Jens Jørgen Mortensen, Thomas Olsen, Kristian S. Thygesen

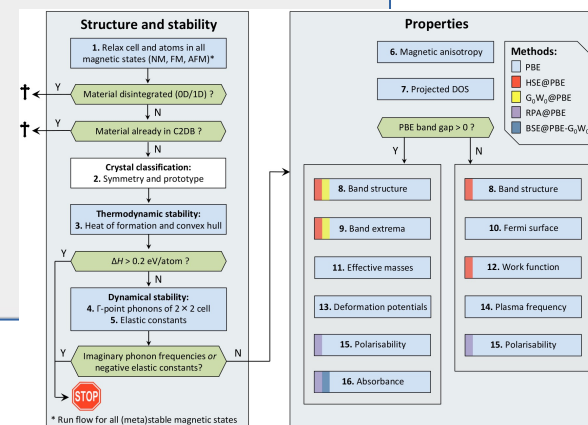
2D Materials 5, 042002 (2018)

The data can be downloaded or browsed online:

- Browse data
- Download data: [c2db.db](https://cmr.fysik.dtu.dk/c2db/c2db.html)

Contents

- Computational 2D Materials Database (C2DB)
 - Brief description
 - Overview of methods and parameters used
 - Versions
 - Key-value pairs
 - Example
 - Tools for creating the "Computational 2D materials" database
 - Requirements
 - Installation
 - Workflow
 - Dependencies



Dervin, Dionysiou and Pillai, *Nanoscale* **8**, 15115 (2016)

Santhosh, Nivetha, Kollu, Srivastava, Sillanpää, Grace and Bhatnagar, *Scientific Reports* **7**, 14107 (2017)

Song, Yang, Zhao and Fang, *Energy & Environmental Science* **4**, 1379 (2011)

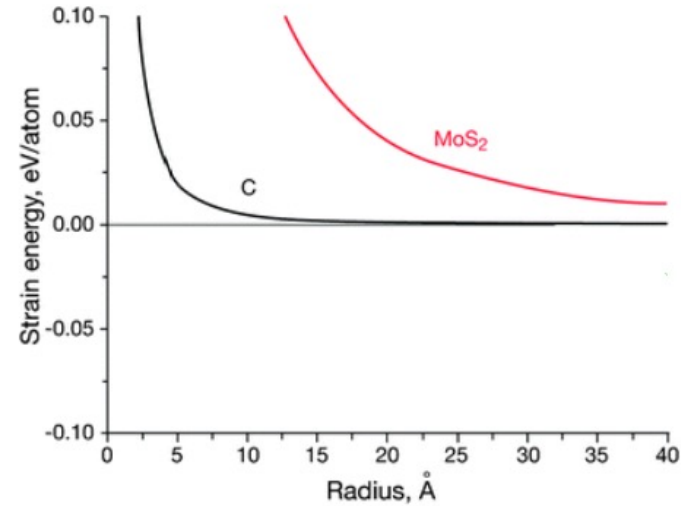
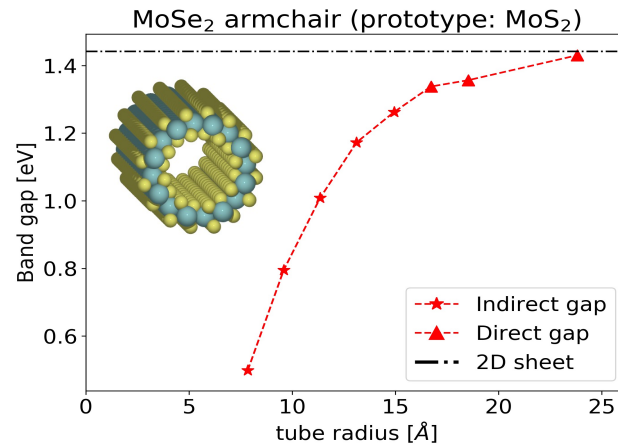
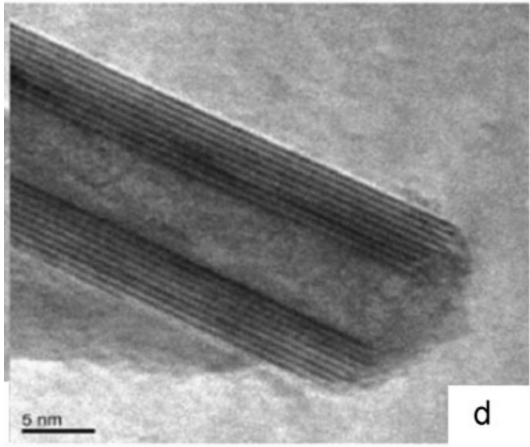
Panchakarla and Tenne, *Nanotechnology for Energy Sustainability* 745 (2017)

Hastrup, ..., and Thygesen, *2D Materials* **5**, 042002 (2018)

<https://cmr.fysik.dtu.dk/c2db/c2db.html>

Most inorganic nanotubes are multiwalled \rightarrow no preferred size, difficult to tune properties

Inorganic single walled exists, **but** have large radii ($> 35 \text{ \AA}$) with properties close to the 2D limit

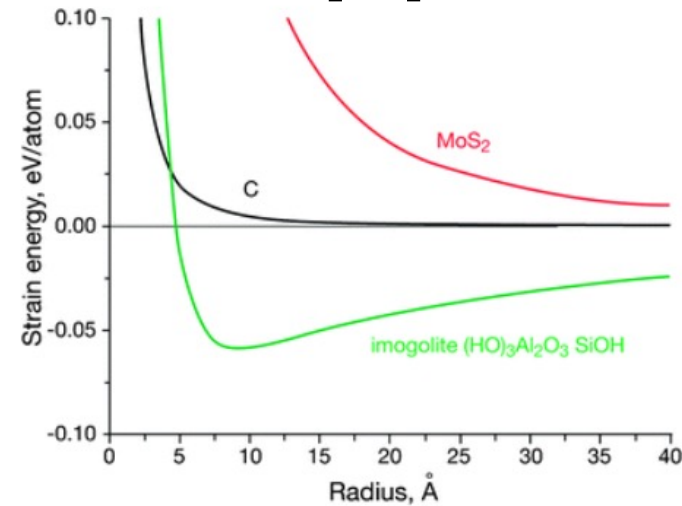
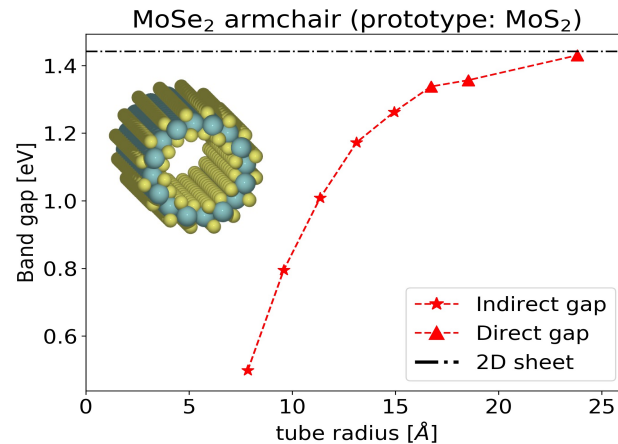
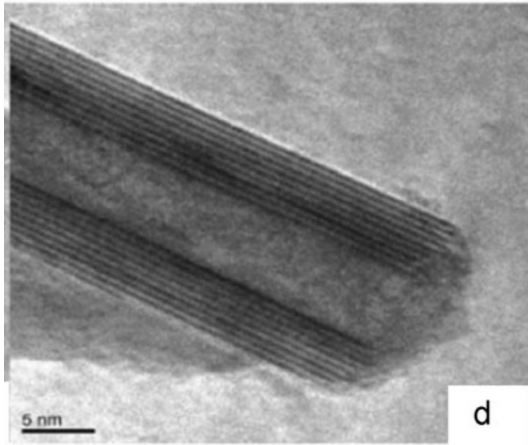


Seifert, Köhler, and Tenne, *J. Phys. Chem. B* **106**, 2497 (2002)

Inorganic Janus Nanotubes

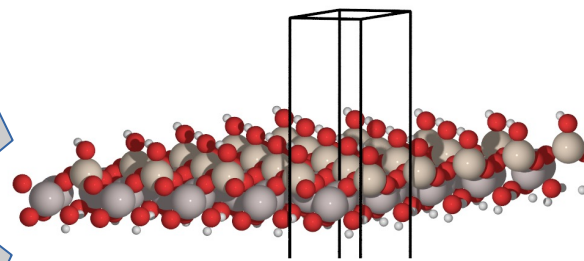
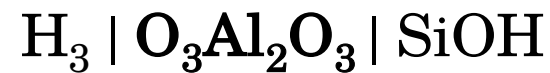
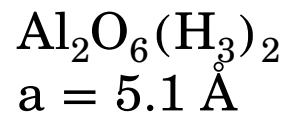
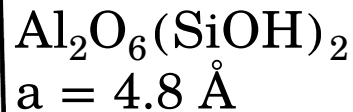
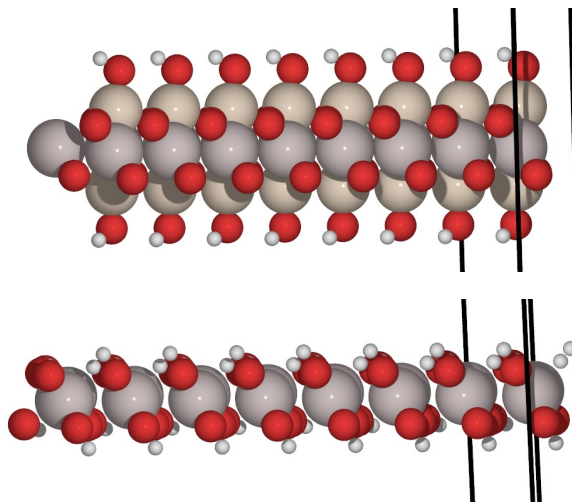
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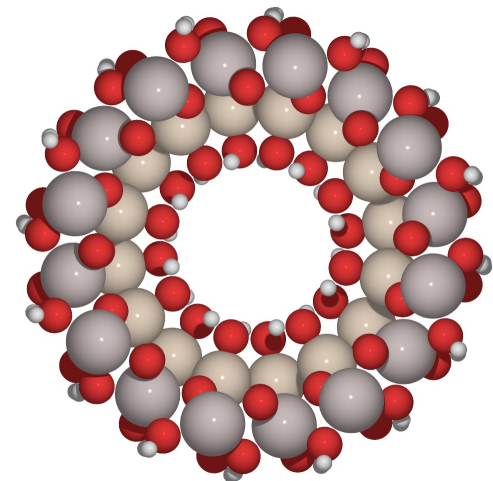


Seifert, Köhler, and Tenne, *J. Phys. Chem. B* **106**, 2497 (2002)

1st Janus Nanotube: Imogolite



wrap



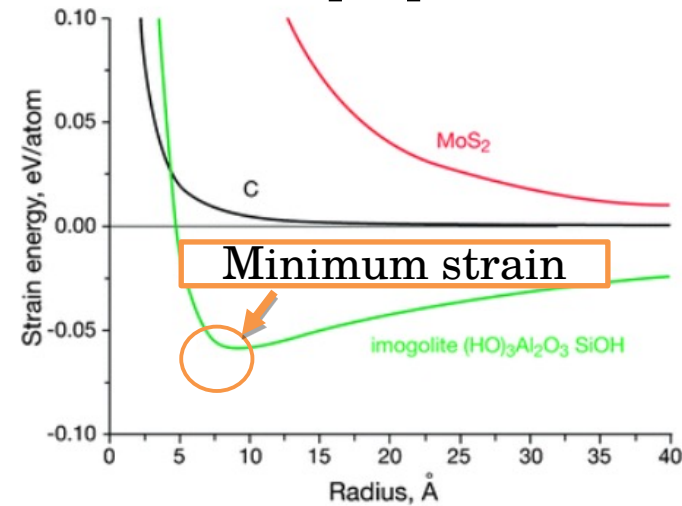
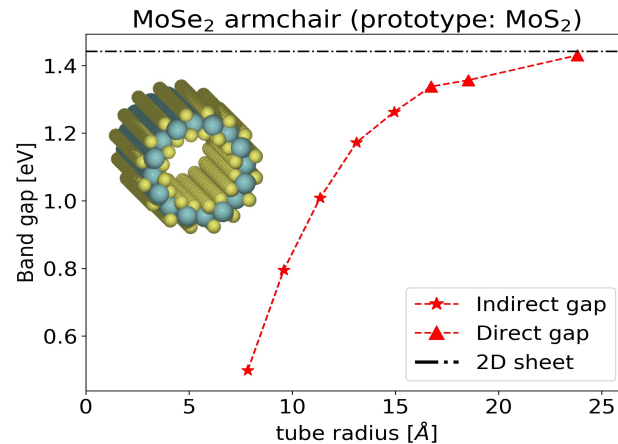
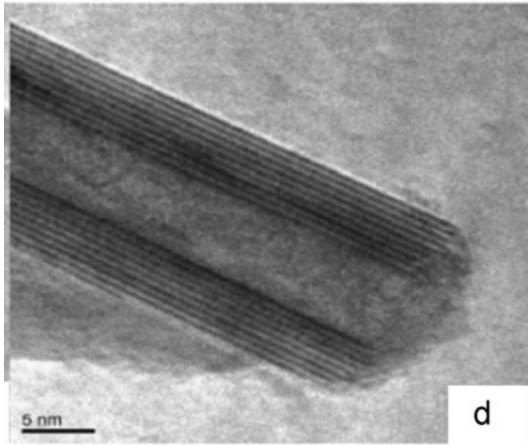
Cradwick *et al.*, *Nat. Phys. Sci.* **240**, 187 (1972)

Monet *et al.*, *Nat. Comm.* **9**, 2033 (2018)

Inorganic Janus Nanotubes

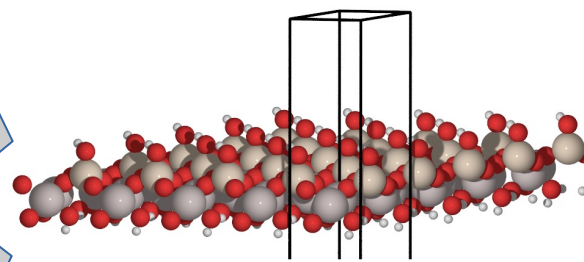
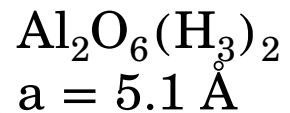
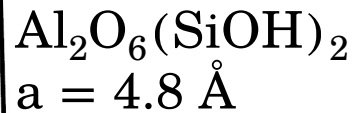
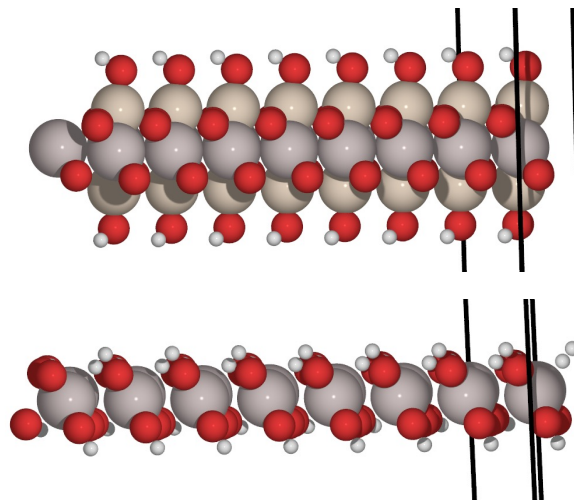
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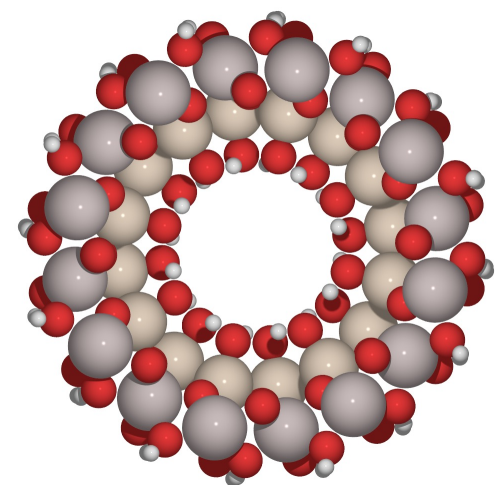


Seifert, Köhler, and Tenne, *J. Phys. Chem. B* **106**, 2497 (2002)

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wrap



Cradwick *et al.*, *Nat. Phys. Sci.* **240**, 187 (1972)

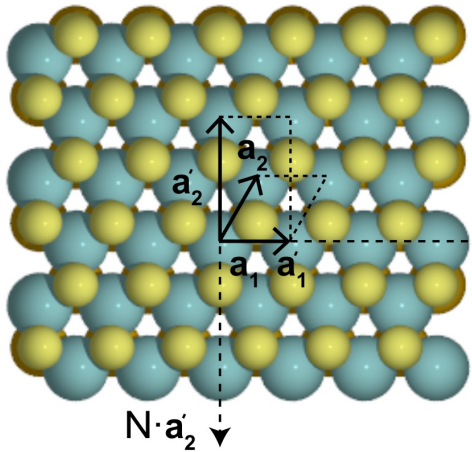
Monet *et al.*, *Nat. Comm.* **9**, 2033 (2018)

Inorganic Janus Nanotubes

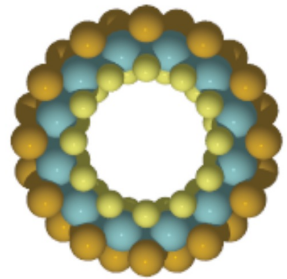
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(a) 2D MoSTe



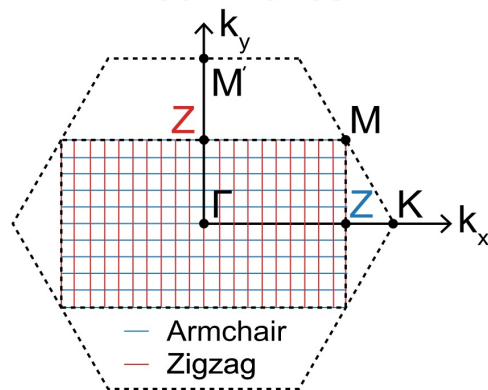
Zigzag nanotube



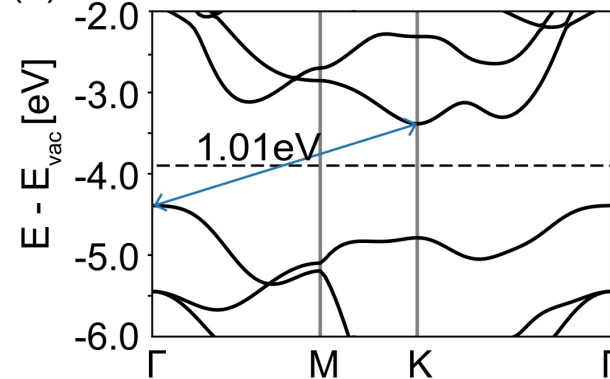
Armchair nanotube



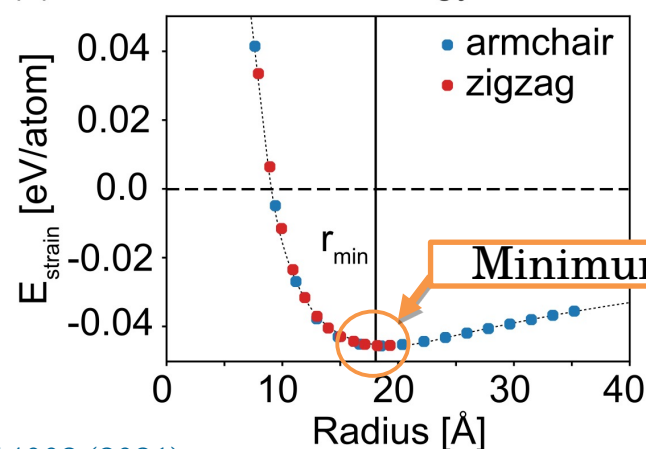
Brillouin zones



(b) 2D MoSTe band structure

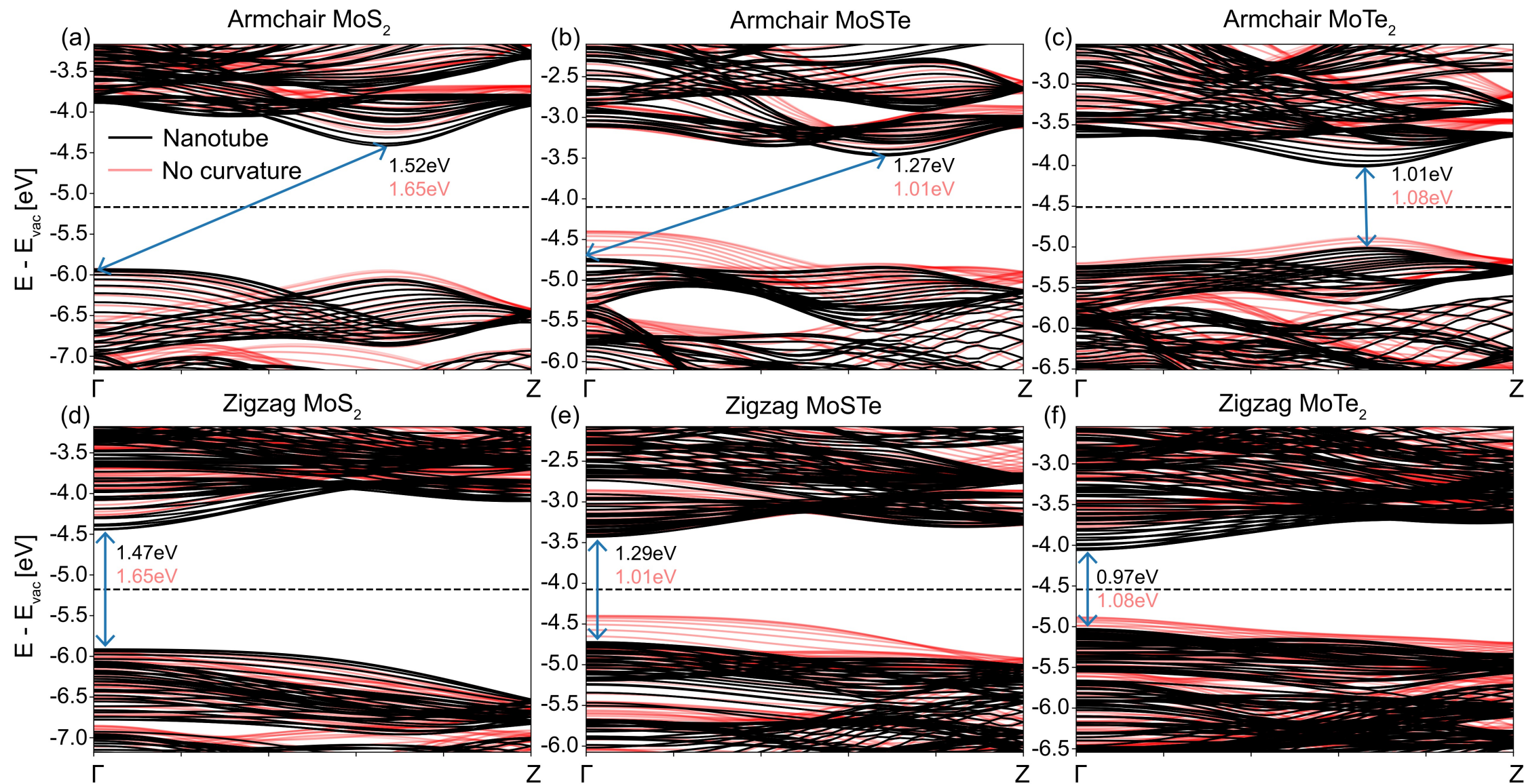


(c) Strain energy curve



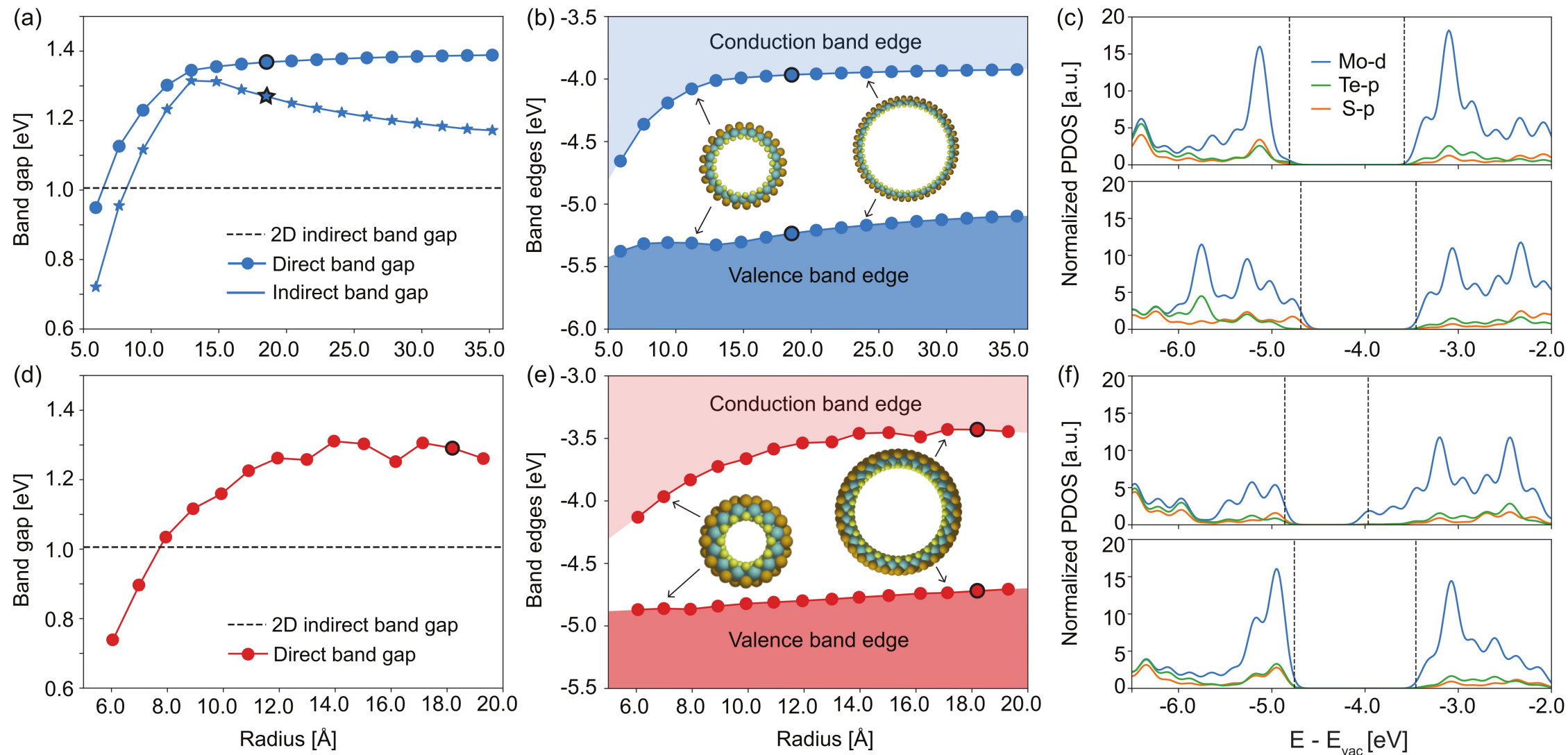
$$E_{strain} = \frac{E_{tube}}{N_{tube}} - \frac{E_{\infty-sheet}}{N_{\infty-sheet}}$$

Janus Nanotube VS Parent Sheets

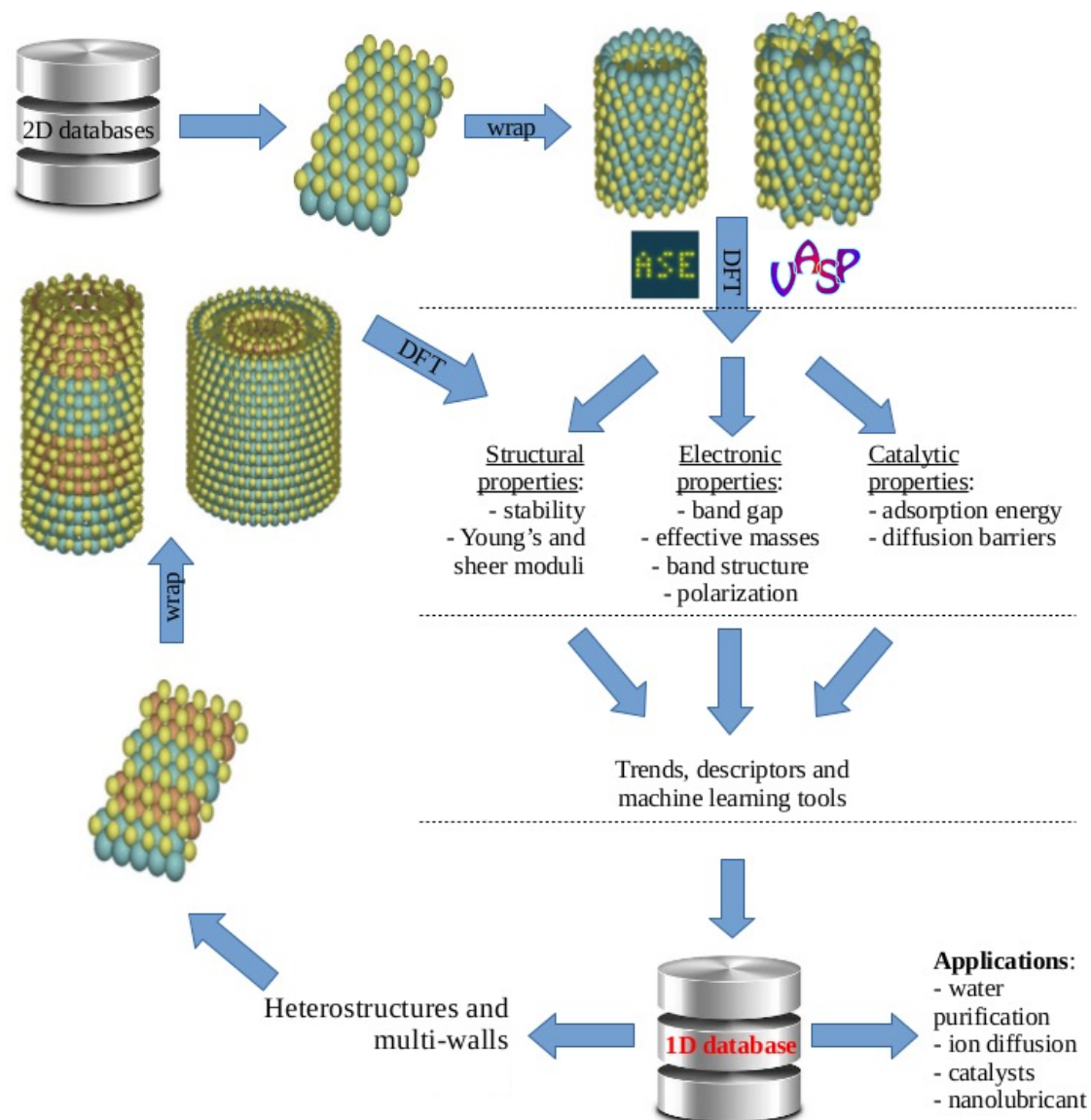


Mikkelsen, Bølle, Thygesen, Vegge, and Castelli, *Phys. Rev. Mater* **5**, 014002 (2021)

Janus Nanotube VS Parent Sheets



Discovery of Nanometric Nanotubes – The Workflow



Why a workflow?

- manages calculations
- manage dependences
- use always the same parameters
- error handling
- collect and share the data

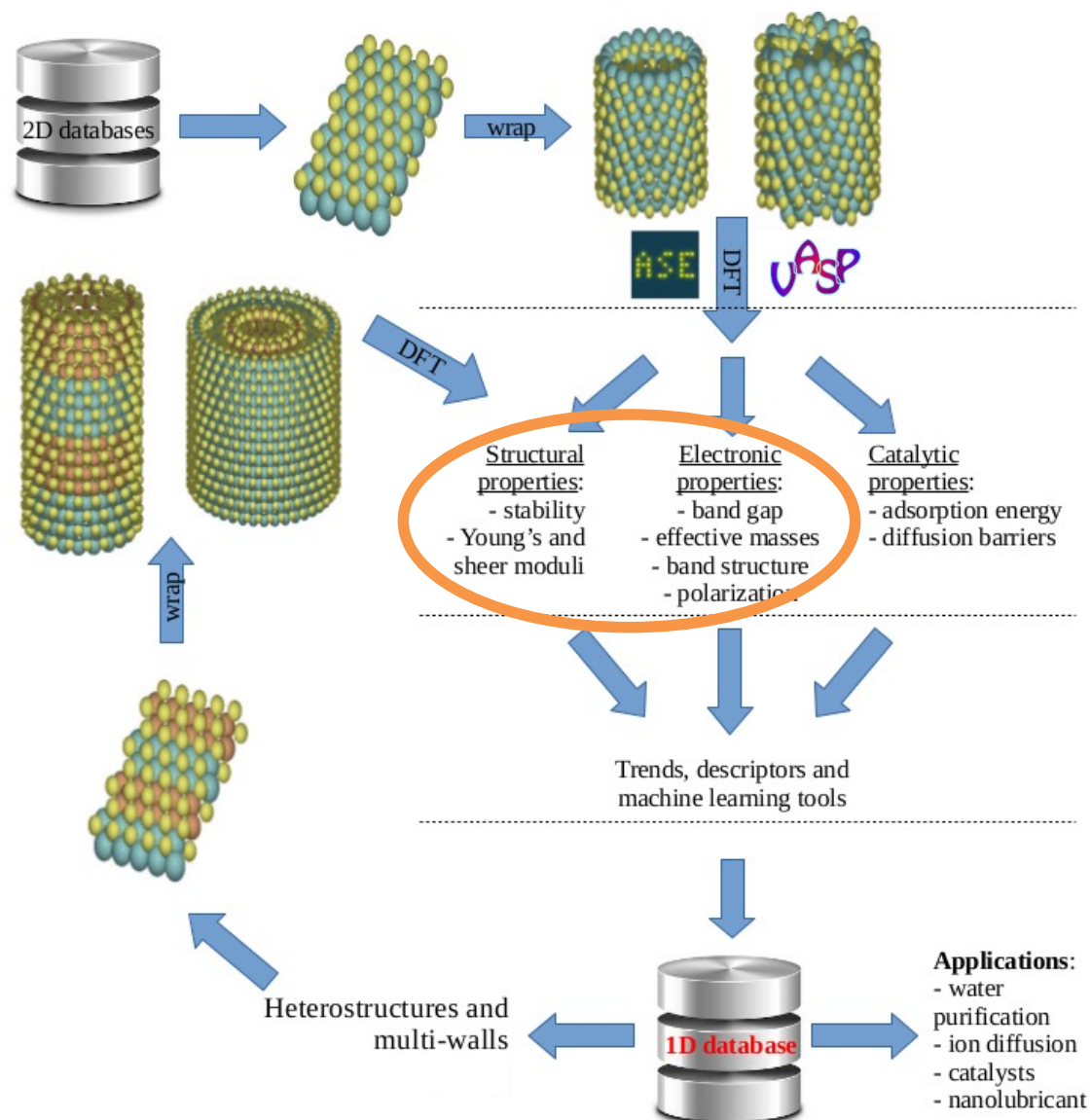
Myqueue:

<http://myqueue.readthedocs.io>

Atomic Simulation Environment (ASE):

<http://wiki.fysik.dtu.dk/ase>

Discovery of Nanometric Nanotubes – The Workflow



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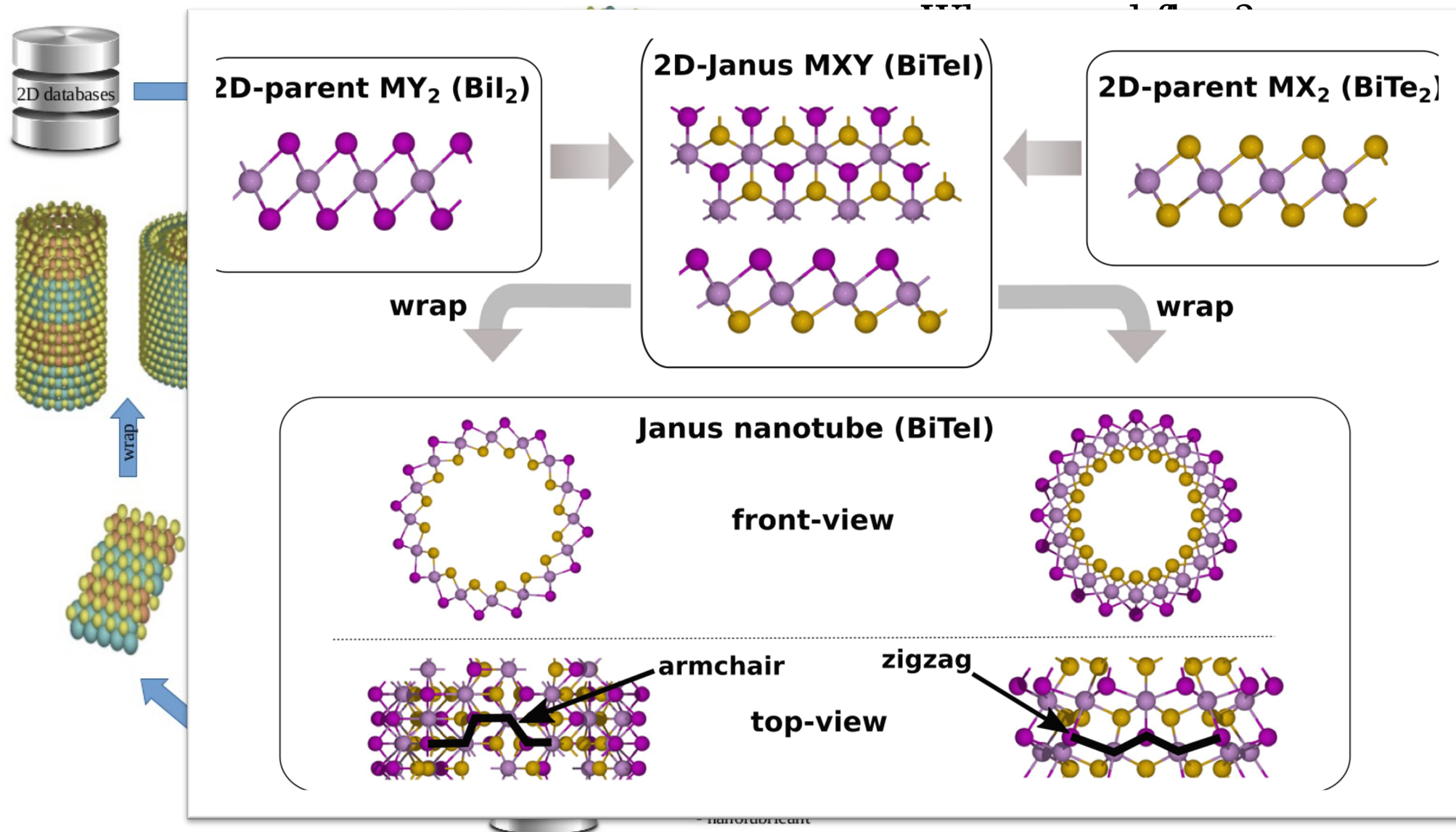
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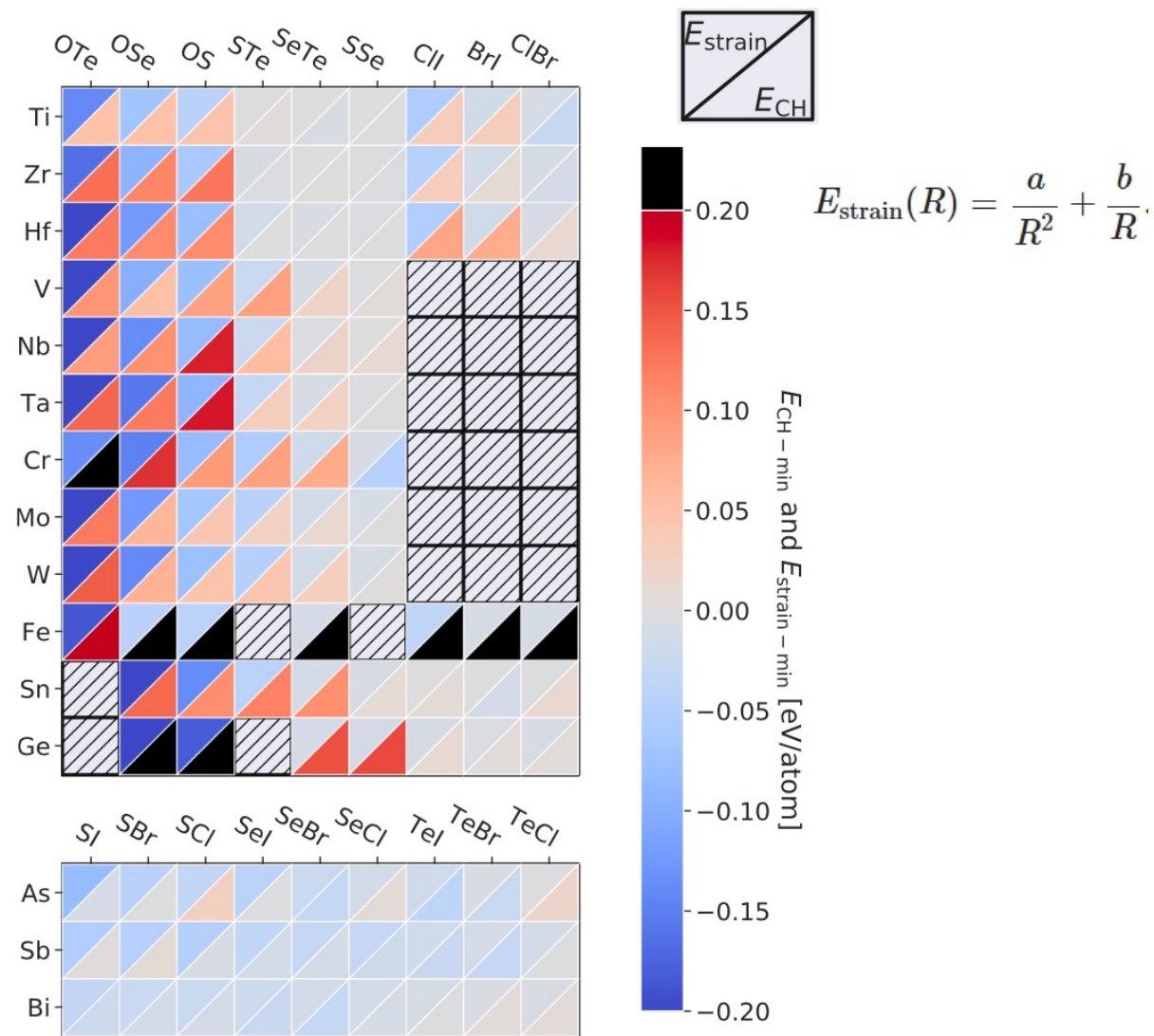
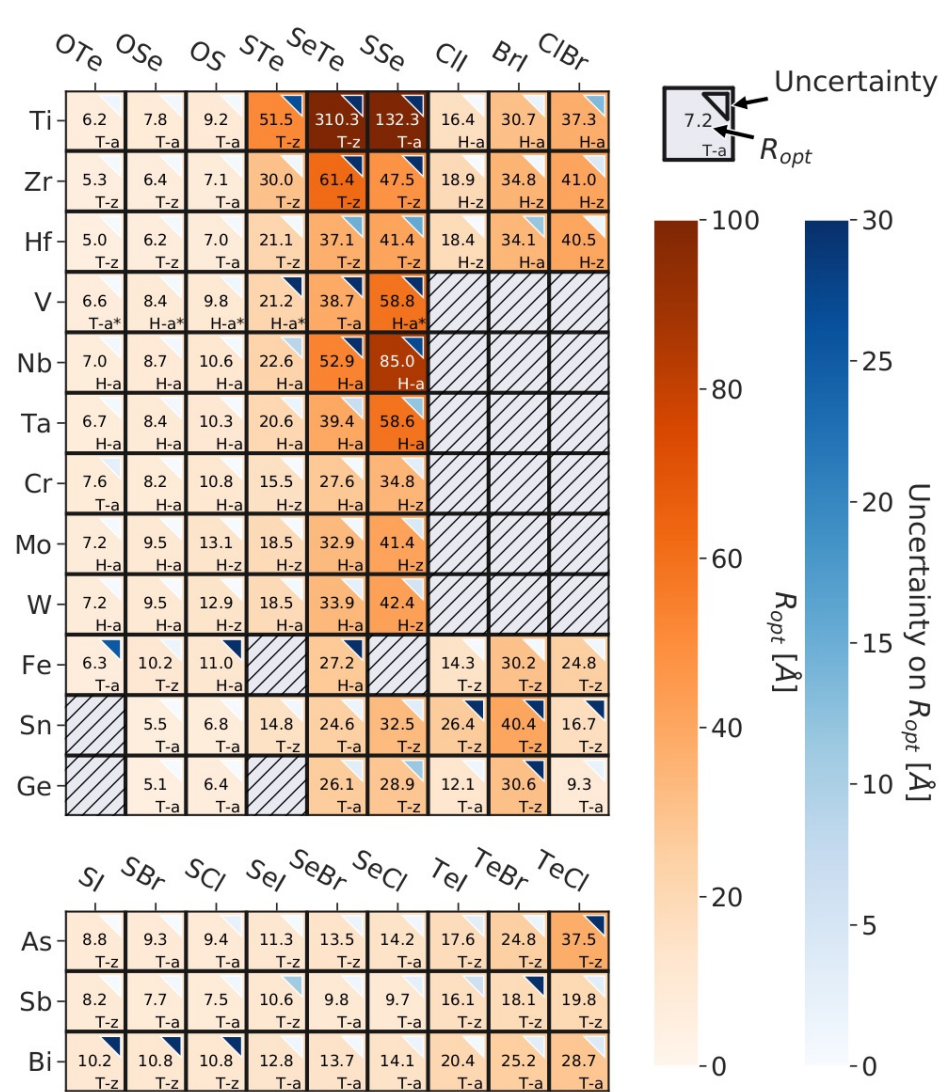
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Discovery of Nanometric Nanotubes – The Workflow



Dichalcogenide Nanotubes – Strain and Formation Energies



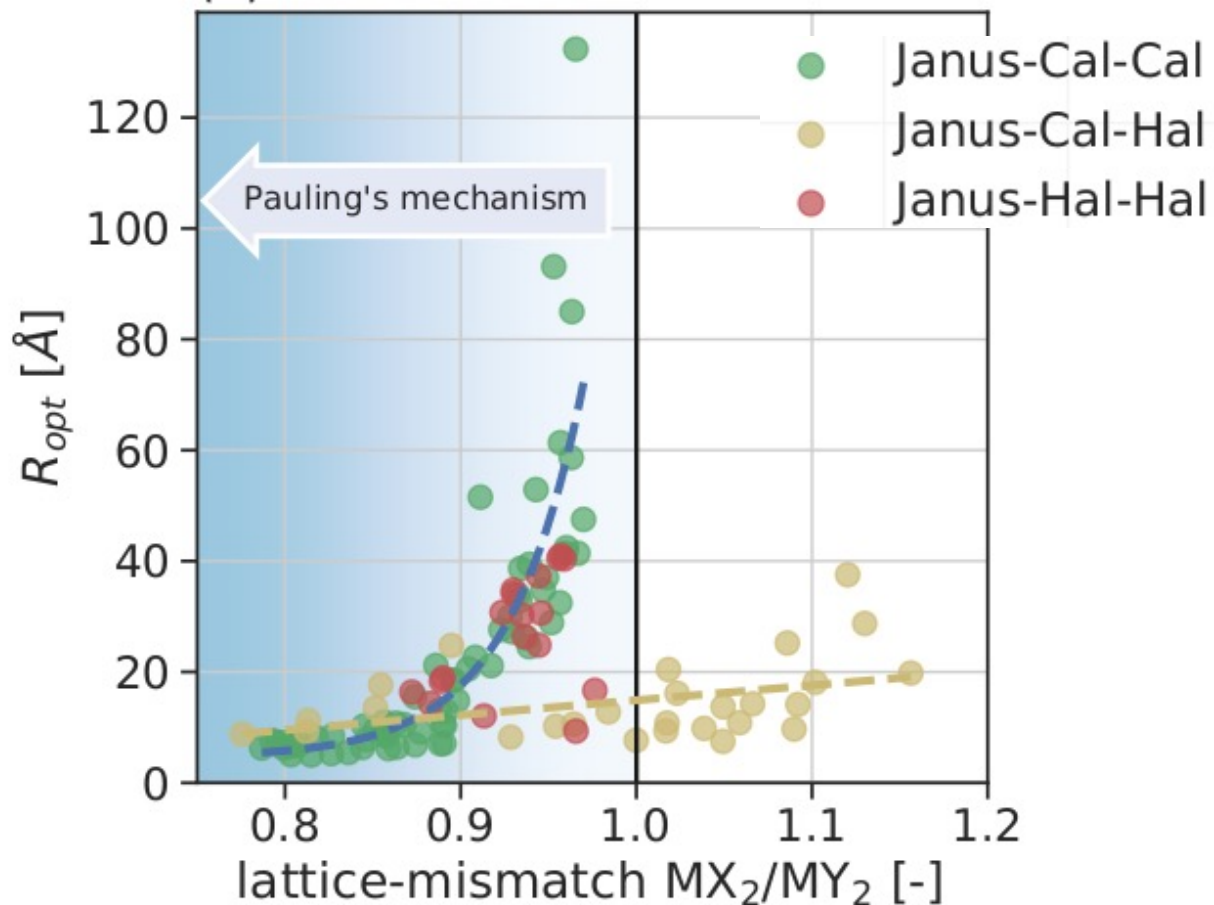
Dichalcogenide Nanotubes – Formation Mechanism

Trends:

OTe < OSe < OS < STe < SeTe < SSe

???

ClI < BrI < ClBr



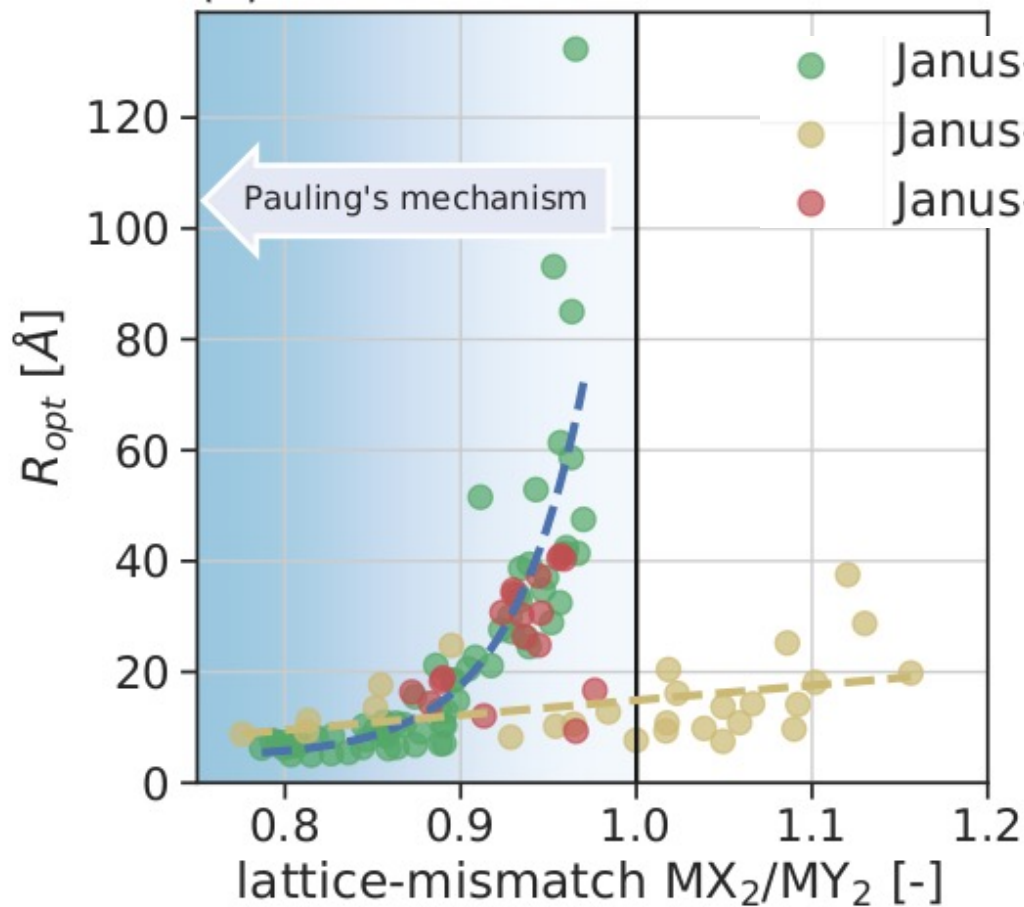
Dichalcogenide Nanotubes – Formation Mechanism

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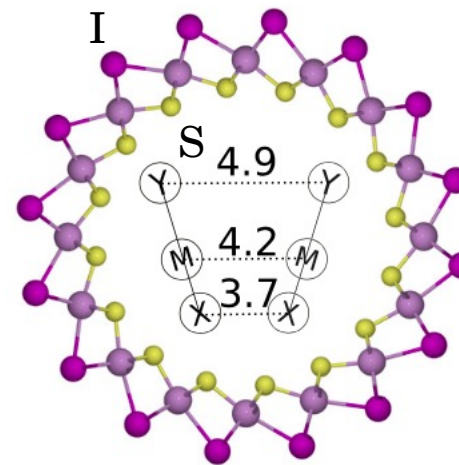
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???

ClI < BrI < ClBr

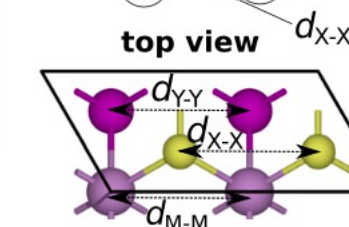
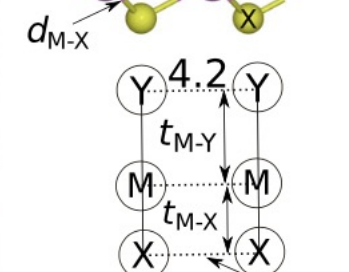
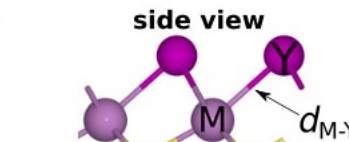


Stable

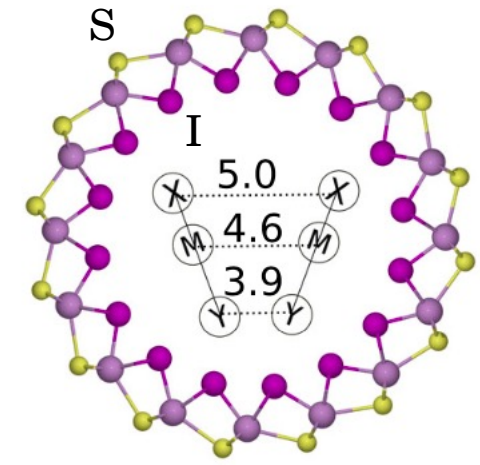


BiSI

2D Janus



Unstable



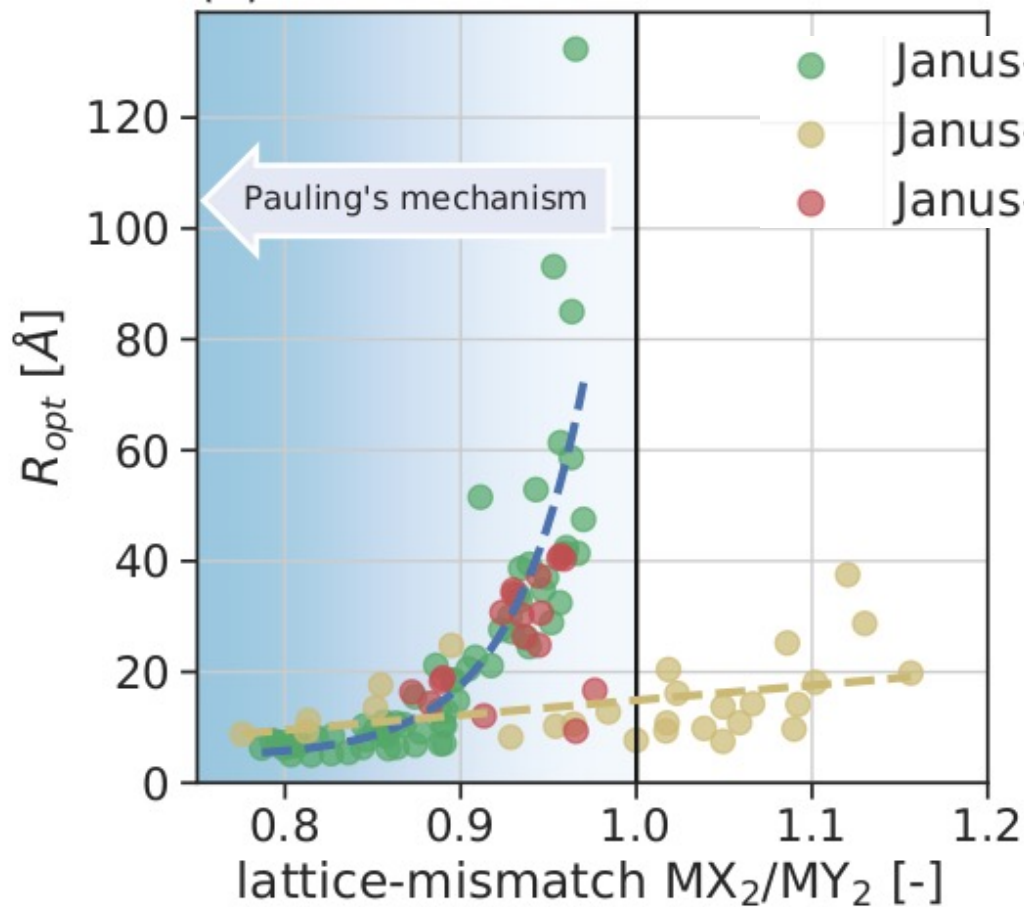
Dichalcogenide Nanotubes – Formation Mechanism

Trends:

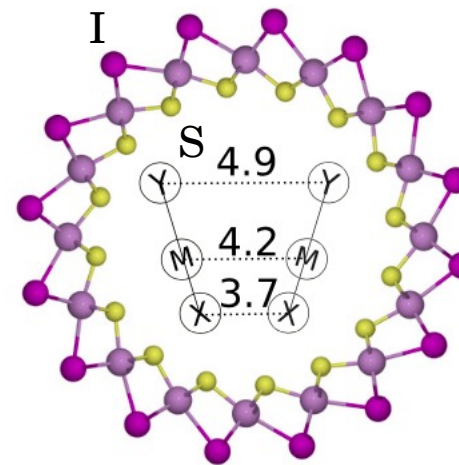
OTe < OSe < OS < STe < SeTe < SSe

Element with stronger bonds determines wrapping direction

ClI < BrI < ClBr

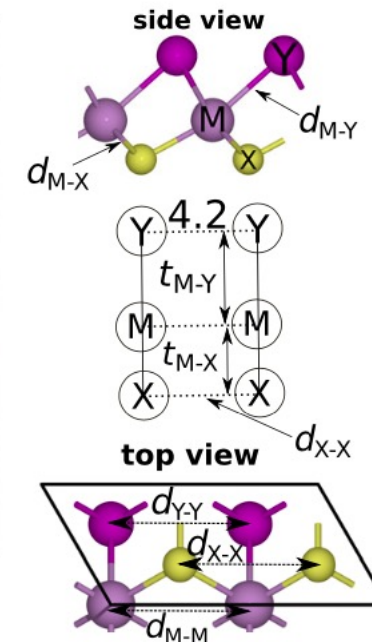


Stable

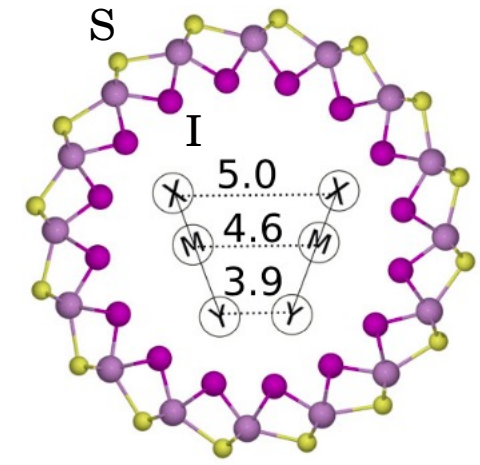


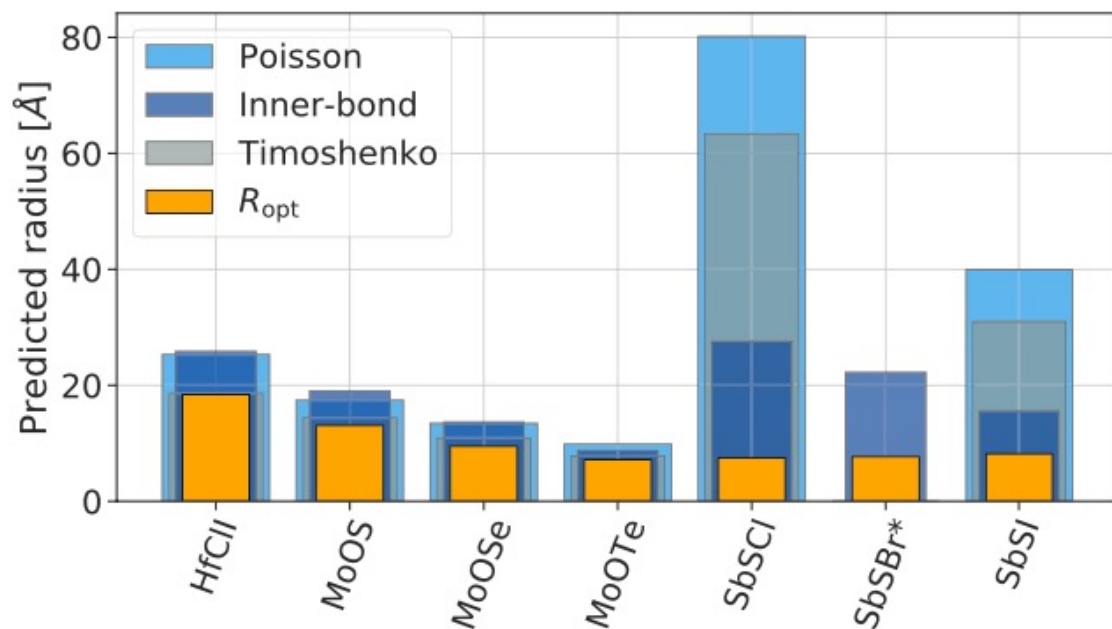
BiSI

2D Janus



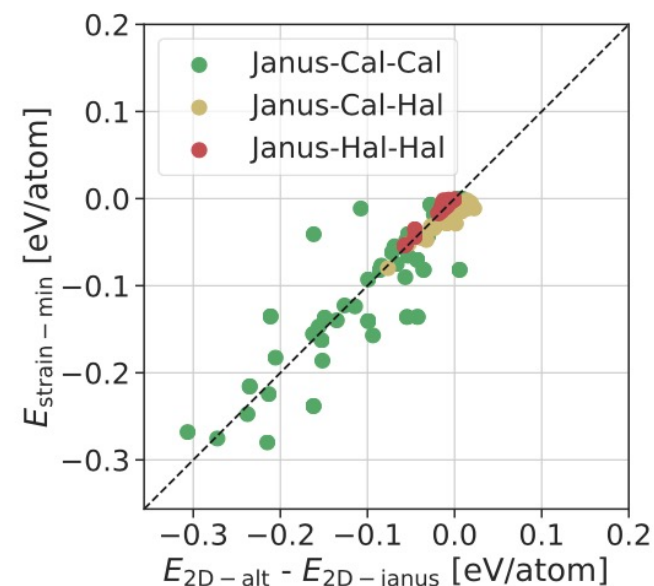
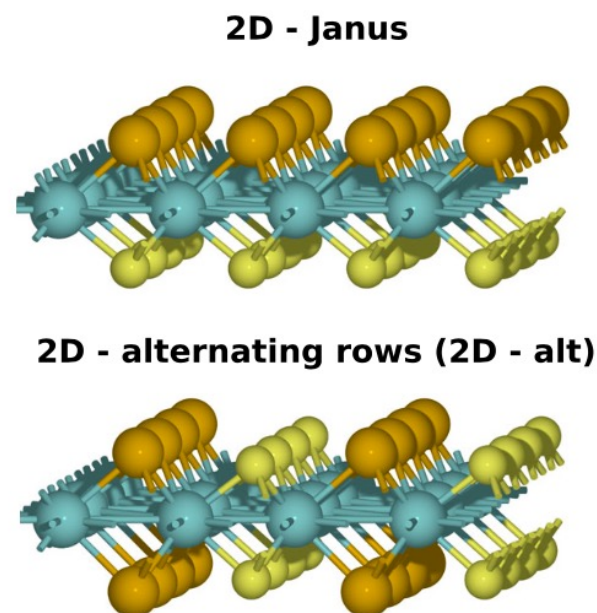
Unstable



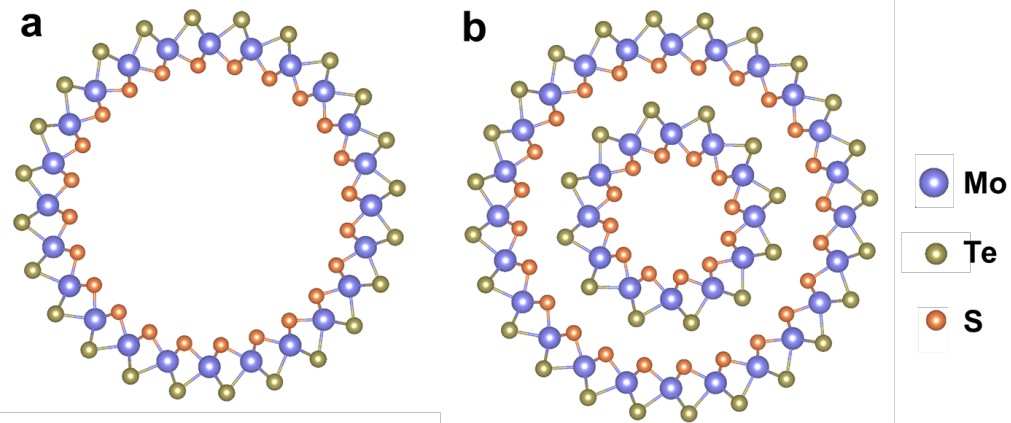
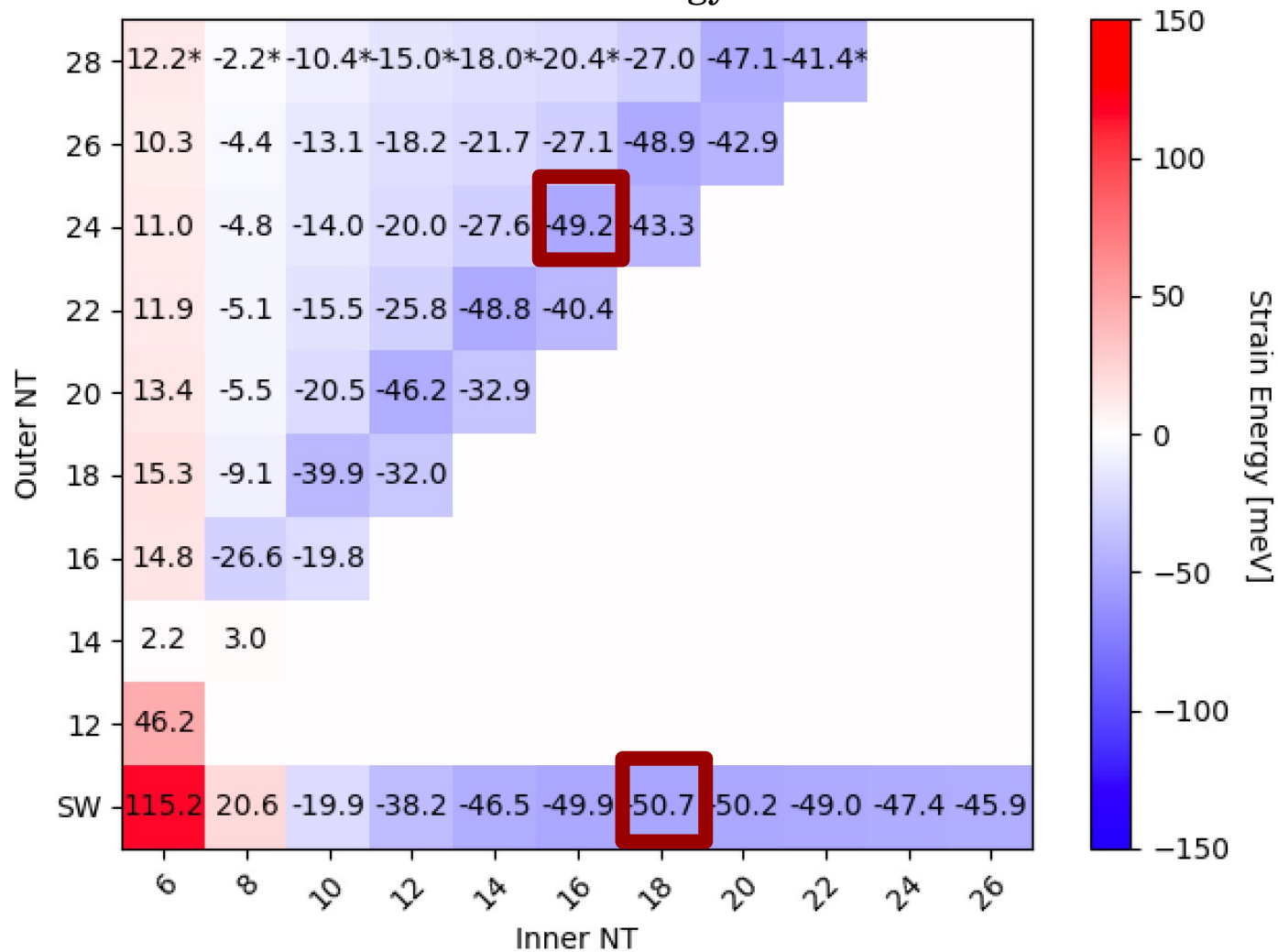


Timoshenko: lattice mismatch between MX_2 and MY_2 2D sheets

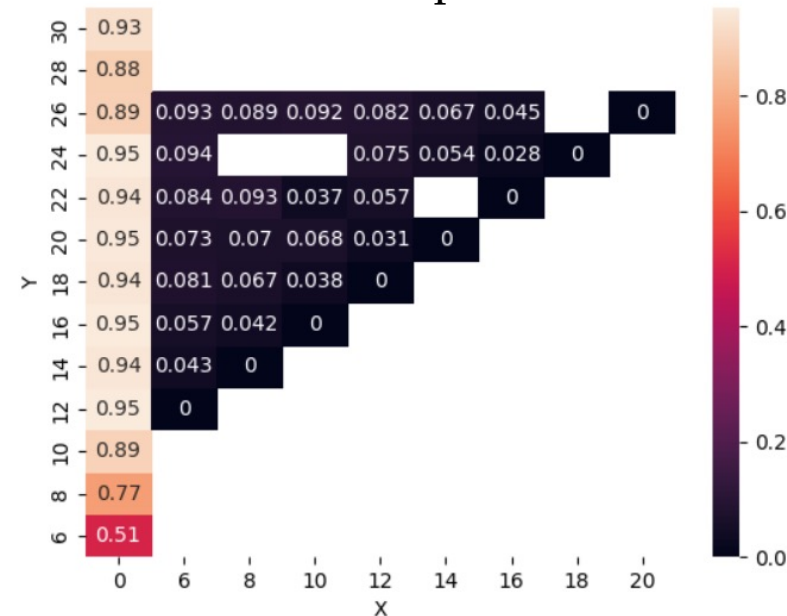
Inner-bond: lattice mismatch between MX_2 and MY_2 2D sheets



Strain Energy



Band Gap

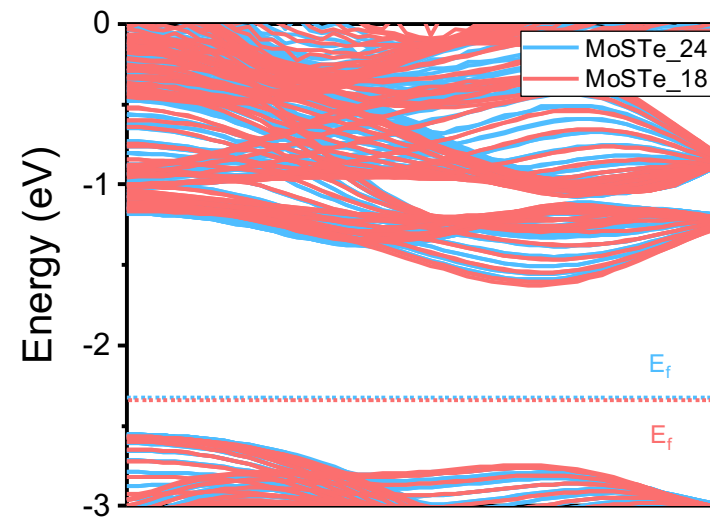
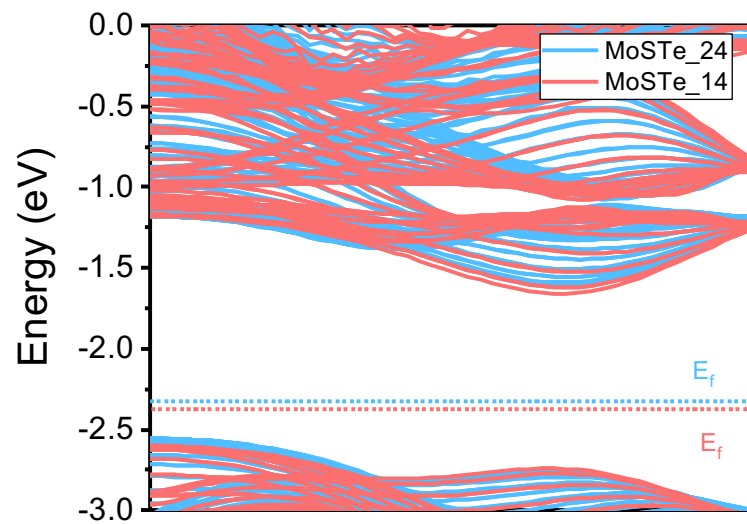
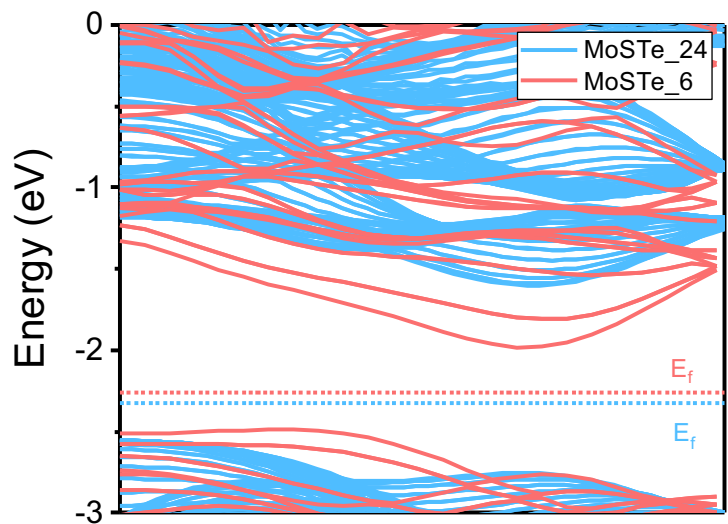




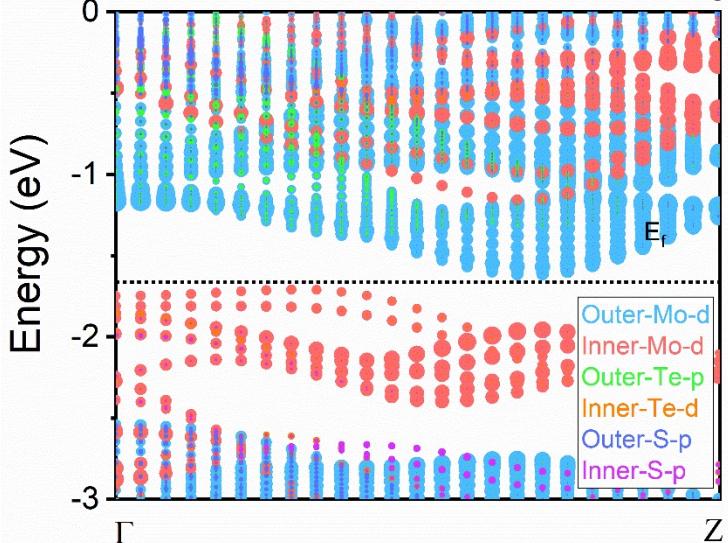
Multiwalls Nanotubes: MoSTe – Band Gap

Single wall nanotube

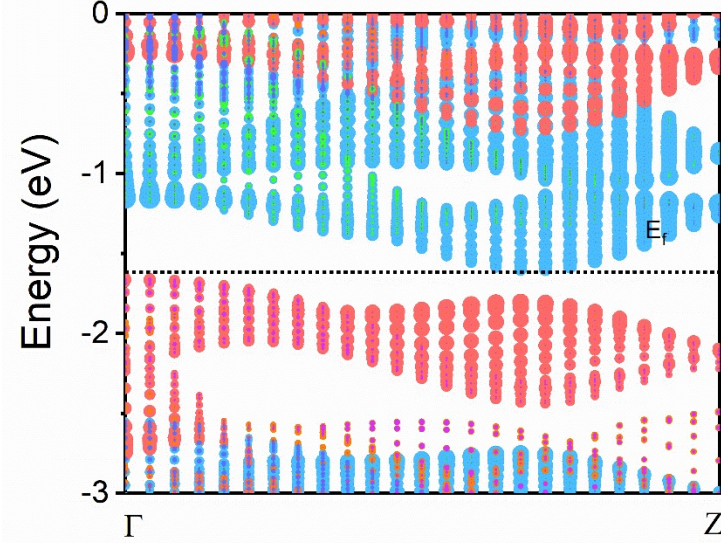
What is happening?



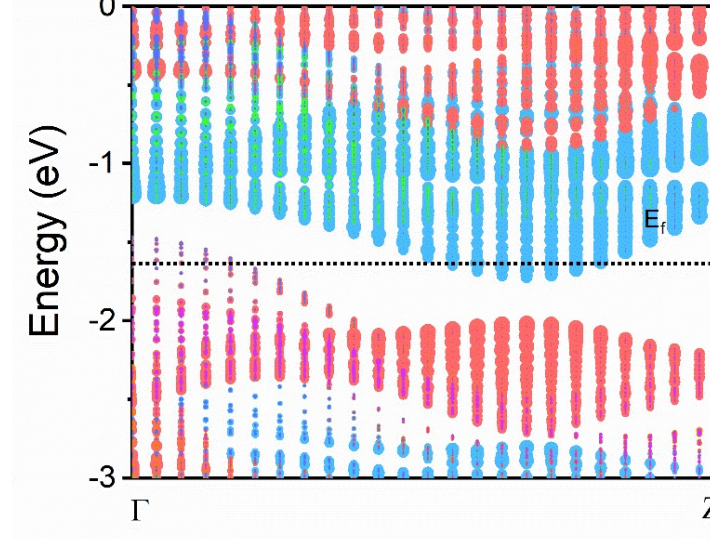
Biwall nanotube 24-6



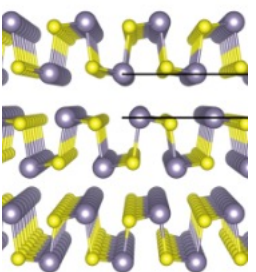
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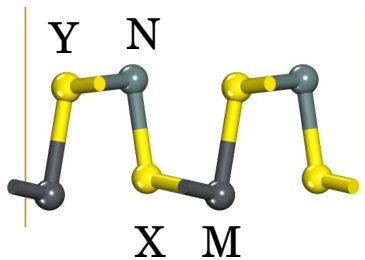
24-18



Beyond the $\text{MoS}_2/\text{CdI}_2$ Nanotubes – IV-VI Janus Tubes



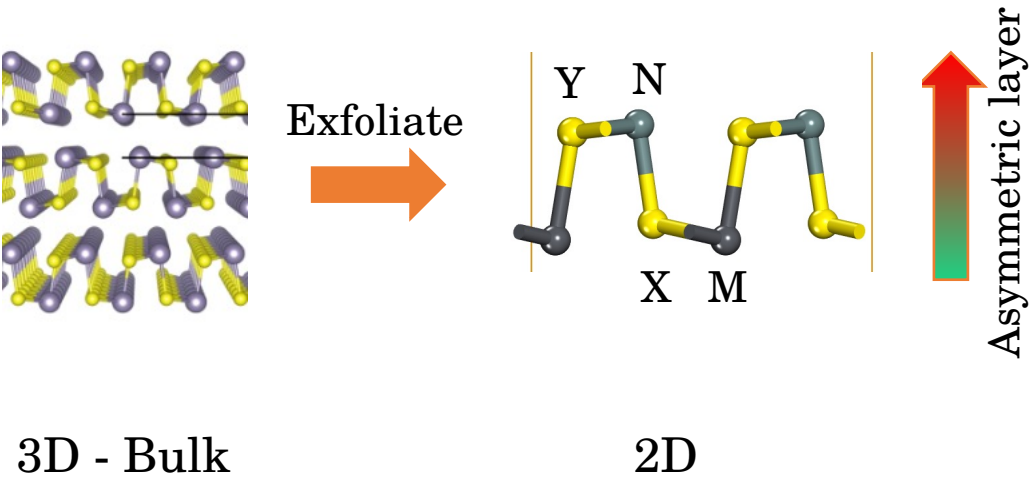
Exfoliate



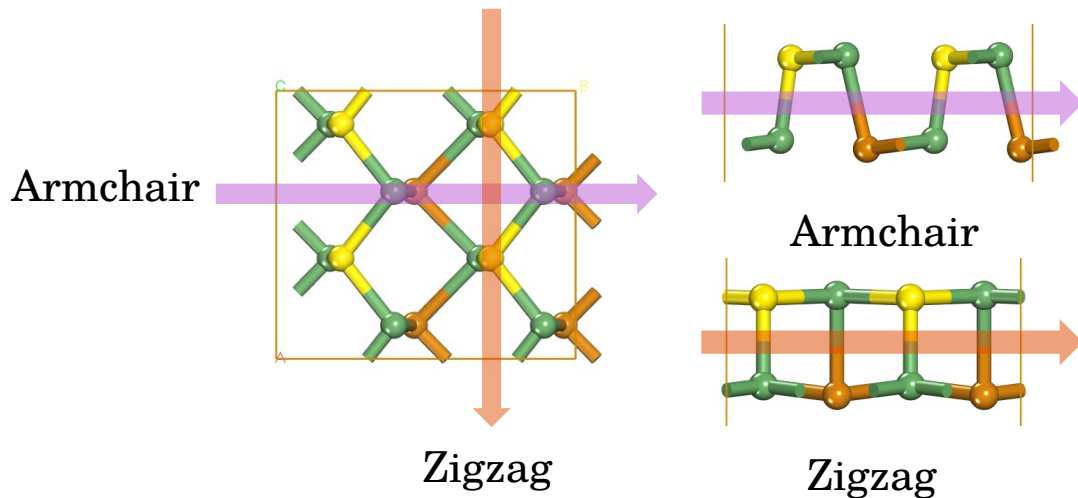
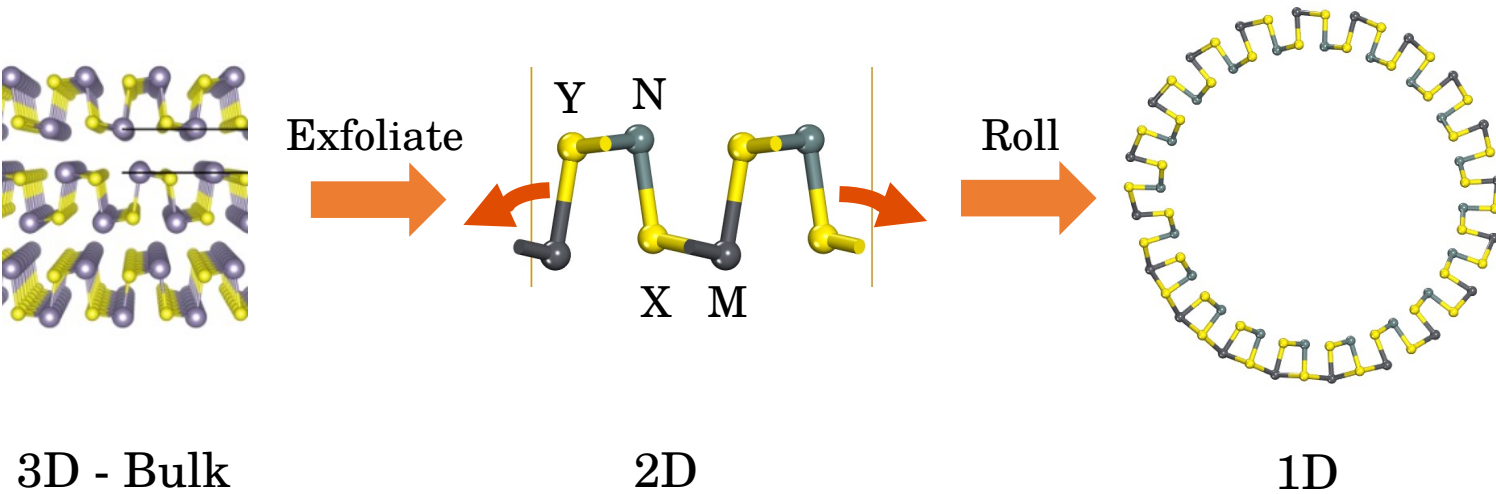
3D - Bulk

2D

Beyond the $\text{MoS}_2/\text{CdI}_2$ Nanotubes – IV-VI Janus Tubes

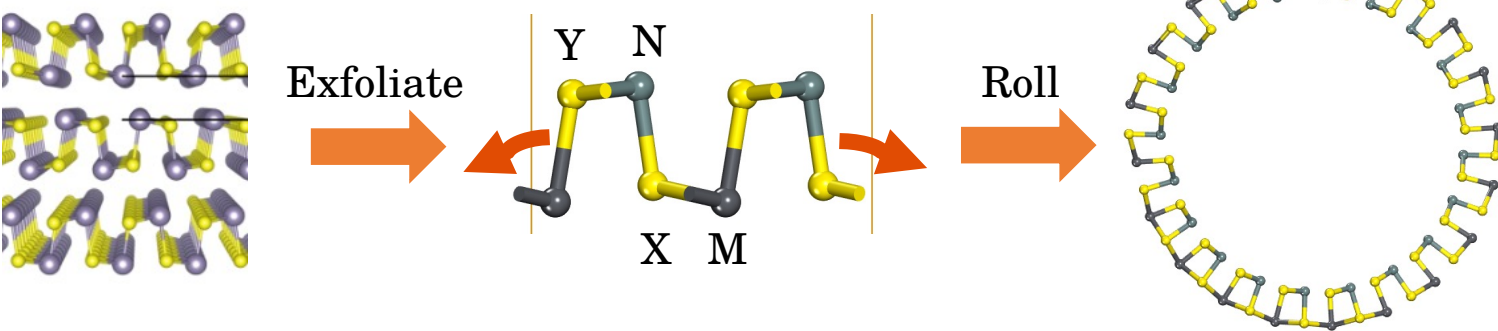


Beyond the $\text{MoS}_2/\text{CdI}_2$ Nanotubes – IV-VI Janus Tubes



Zheng, Bölle, Vegge, and Castelli, in preparation (2022)

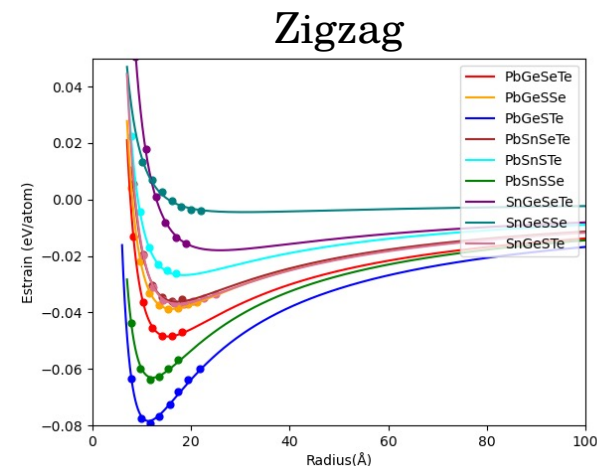
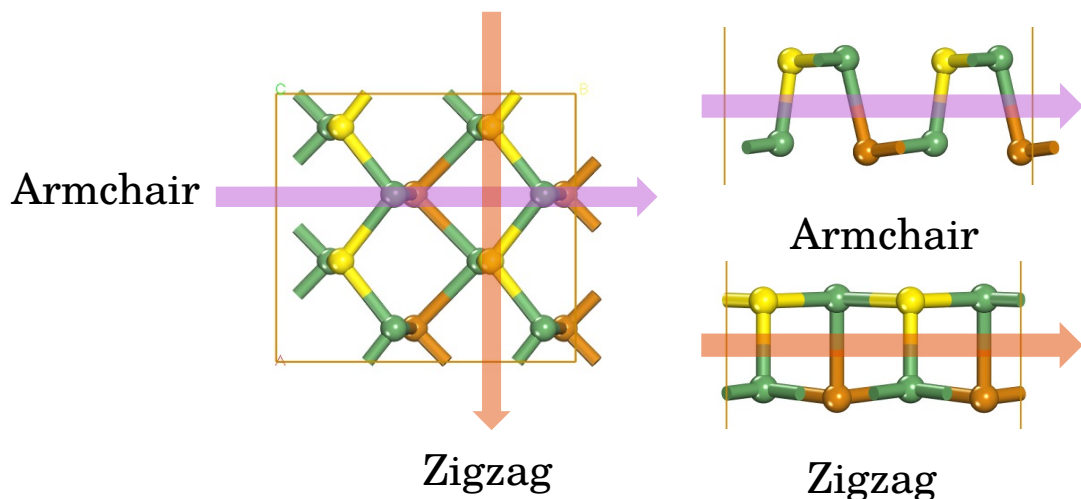
Beyond the $\text{MoS}_2/\text{CdI}_2$ Nanotubes – IV-VI Janus Tubes



3D - Bulk

2D

1D

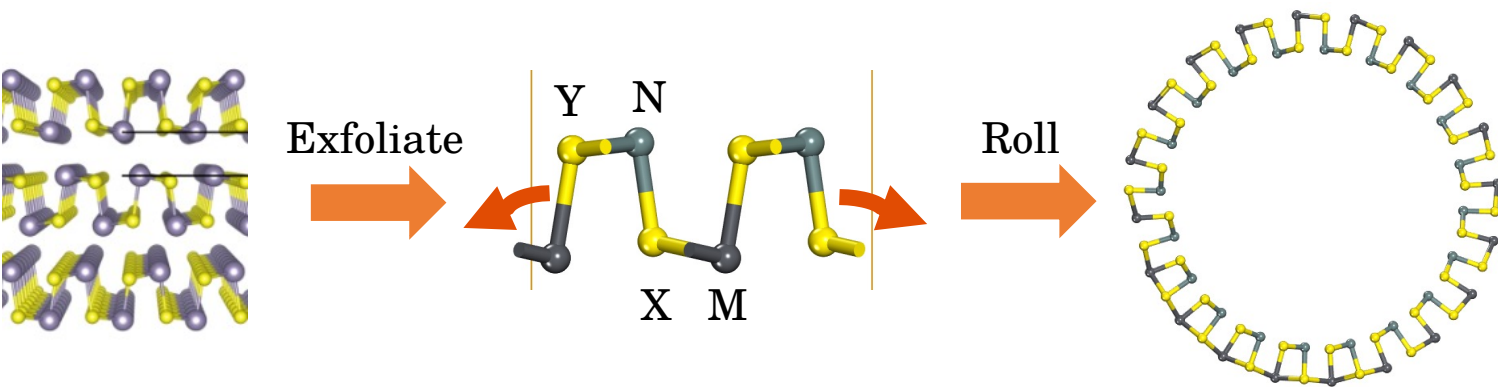


	SeTe	SSe	STe	S ₂	Se ₂	Te ₂
PbGe	15.37	16.17	11.34	23.77	25.15	25.31
PbSn	25.91	30.76	17.3	67.69	464.28	110.92
SnGe	17.16	18.42	12.19	30.75	32.4	32.38
Pb ₂	23.09	41.74	18.62			
Sn ₂	24.16	41.6	18.6			
Ge ₂	22.12	30.98	17.32			

R_{min} (Å)

Zheng, Bölle, Vegge, and Castelli, in preparation (2022)

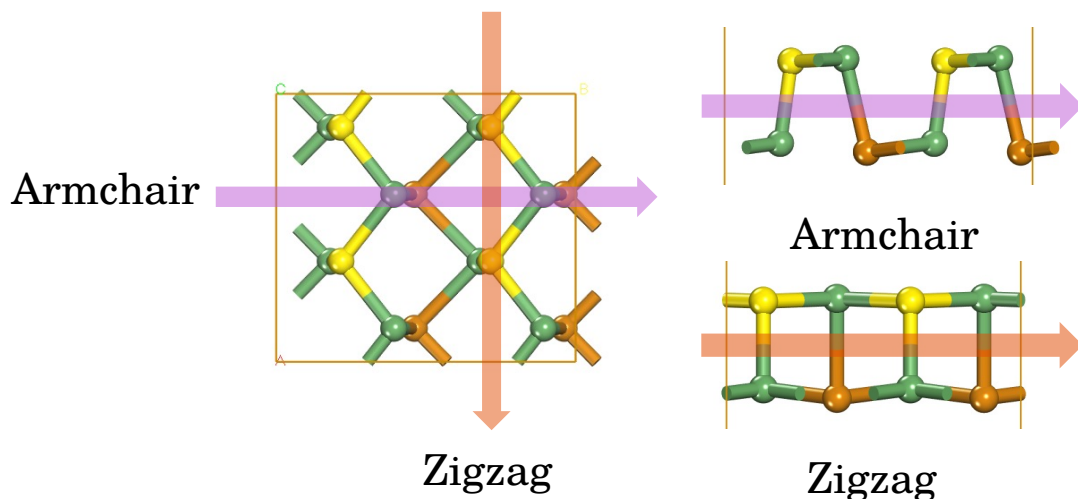
Beyond the MoS₂/CdI₂ Nanotubes – IV-VI Janus Tubes



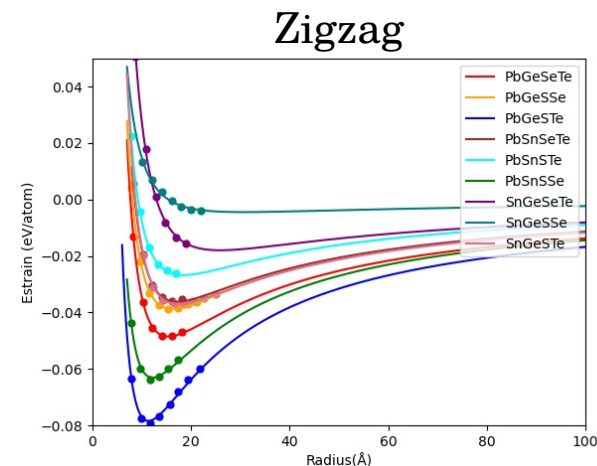
3D - Bulk

2D

1D



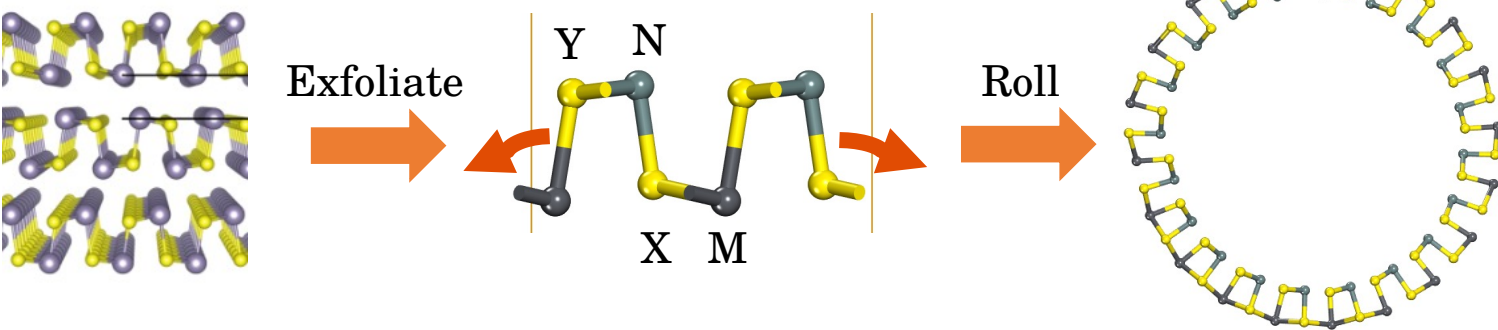
What is the formation mechanism?



	SeTe	SSe	STe	S ₂	Se ₂	Te ₂	R _{min} (Å)
PbGe	15.37	16.17	11.34	23.77	25.15	25.31	
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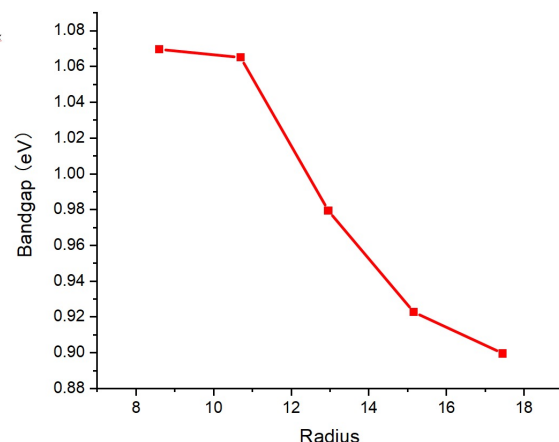
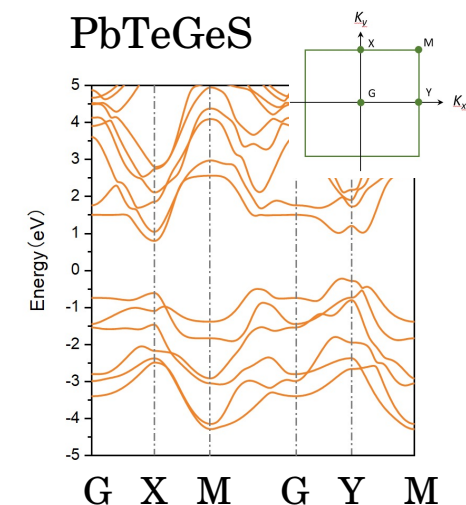
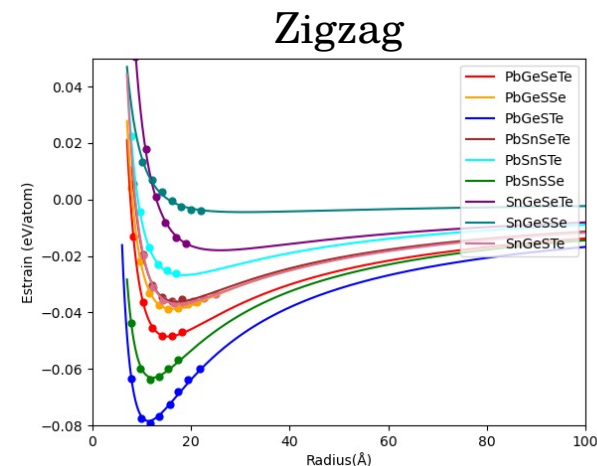
Beyond the MoS₂/CdI₂ Nanotubes – IV-VI Janus Tubes



3D - Bulk

2D

1D



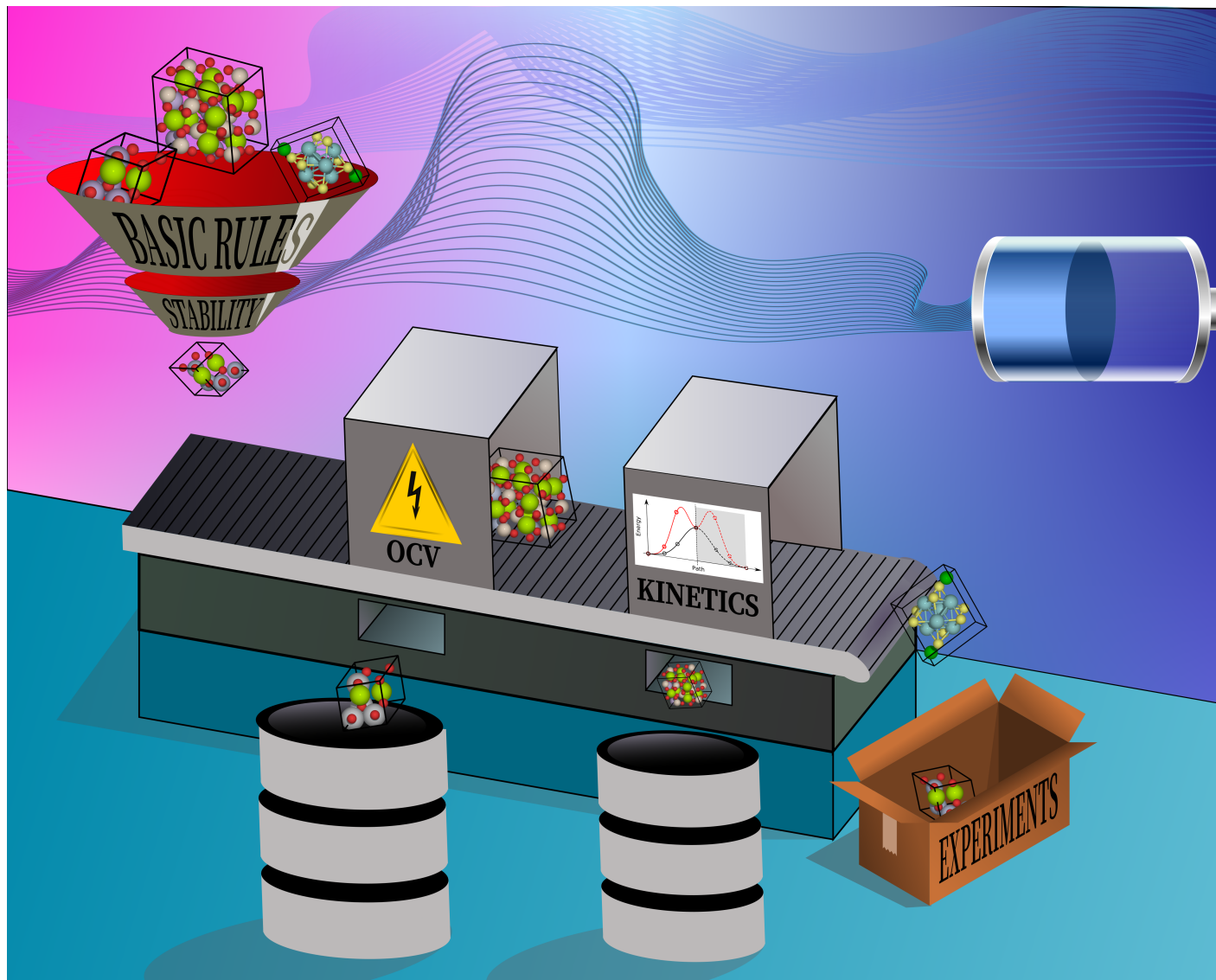
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Zheng, Bölle, Vegge, and Castelli, in preparation (2022)

Autonomous Discovery of Battery Electrodes



Bölle, Mathiesen, Nielsen, Vegge, Garcia Lastra, and Castelli, *BatteriesSupercaps* **3**, 488 (2021)

Why Mg-ion Batteries?

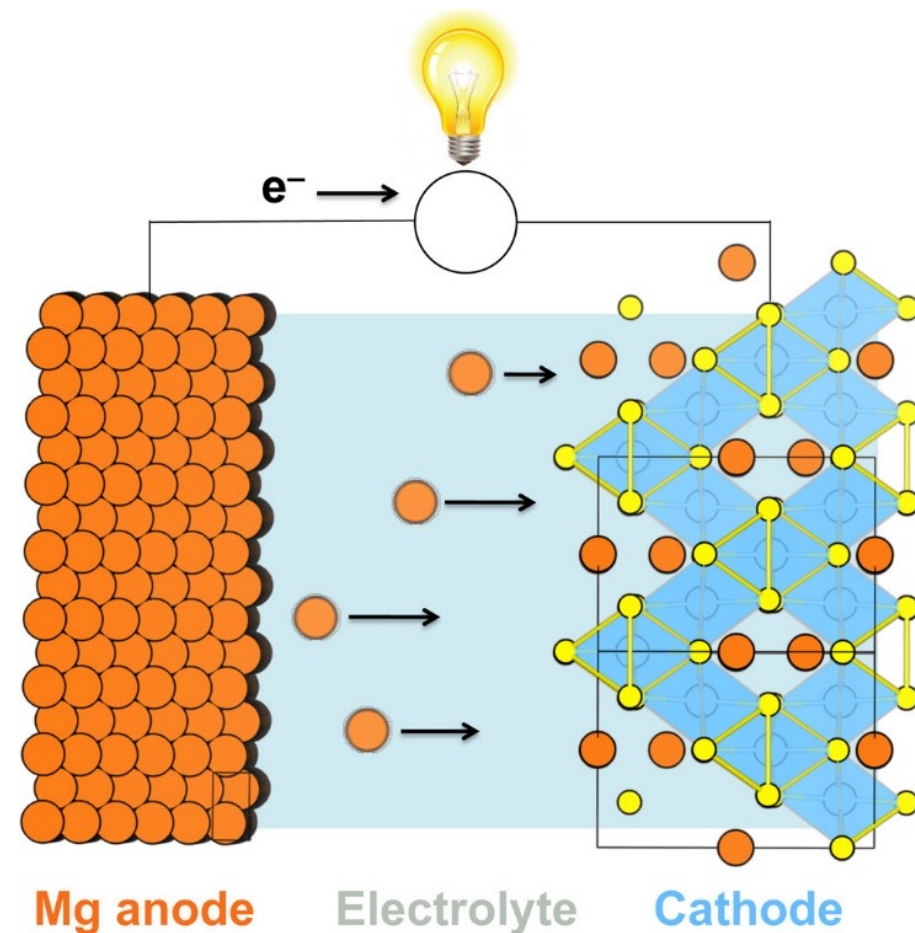
- Multivalent battery
- Increased capacity compared to LIB
- Abundant resources

Main challenges?

- Unstable electrolytes
- Poor cathode materials:
 - low open circuit voltage (OCV)
 - low charge/discharge rate

State-of-the-art cathodes:

$\text{Mg}_x\text{Mo}_6\text{O}_8$ ($0 < x < 2$, Chevrel) – OCV \sim 1.1 V

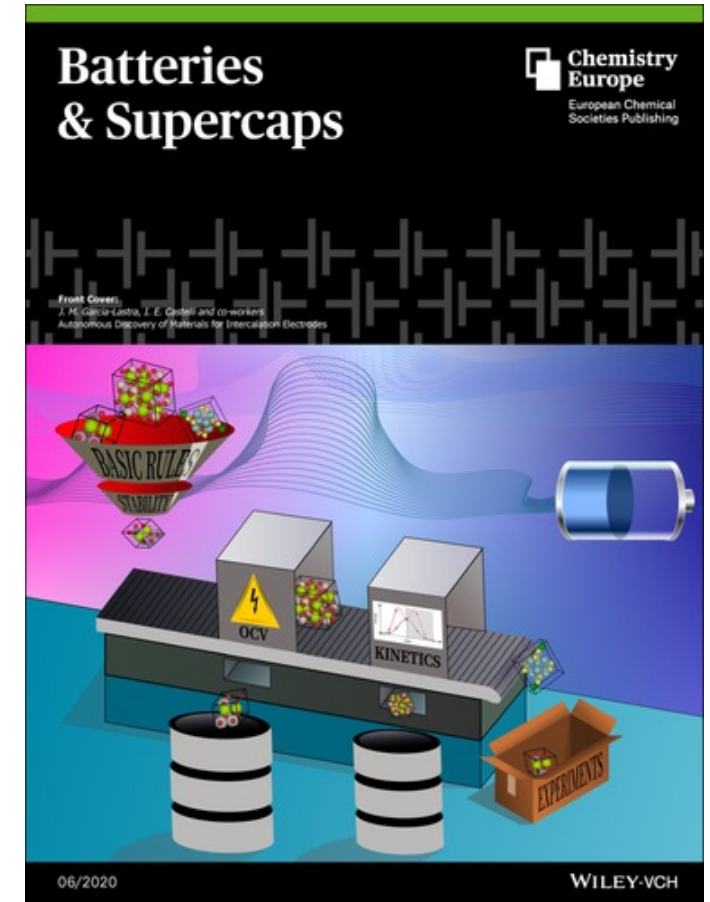
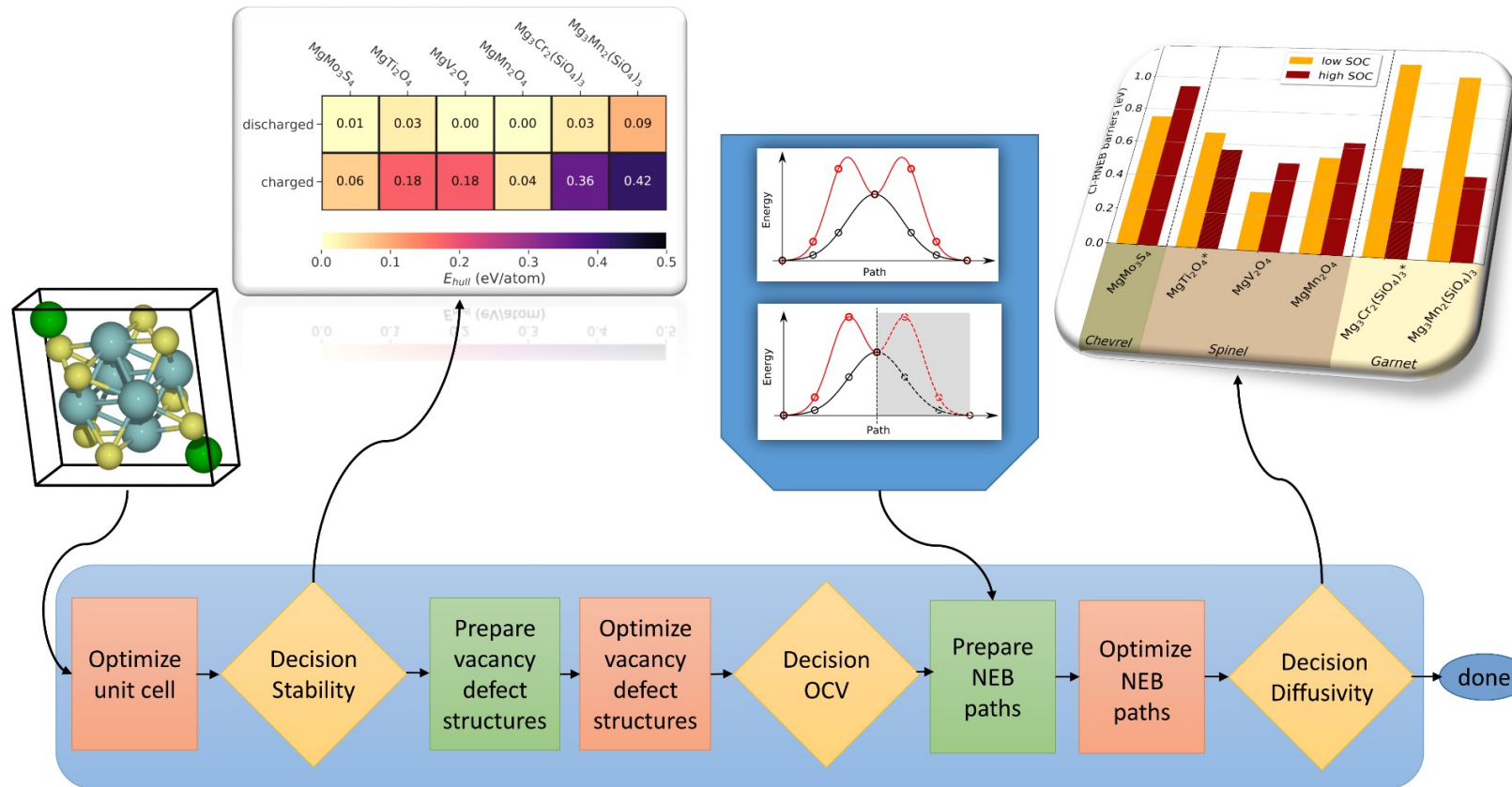


Canepa et al., *Chem. Rev.* **117**, 4287 (2017)

Aurbach et al., *Nature* **407**, 724 (2000)

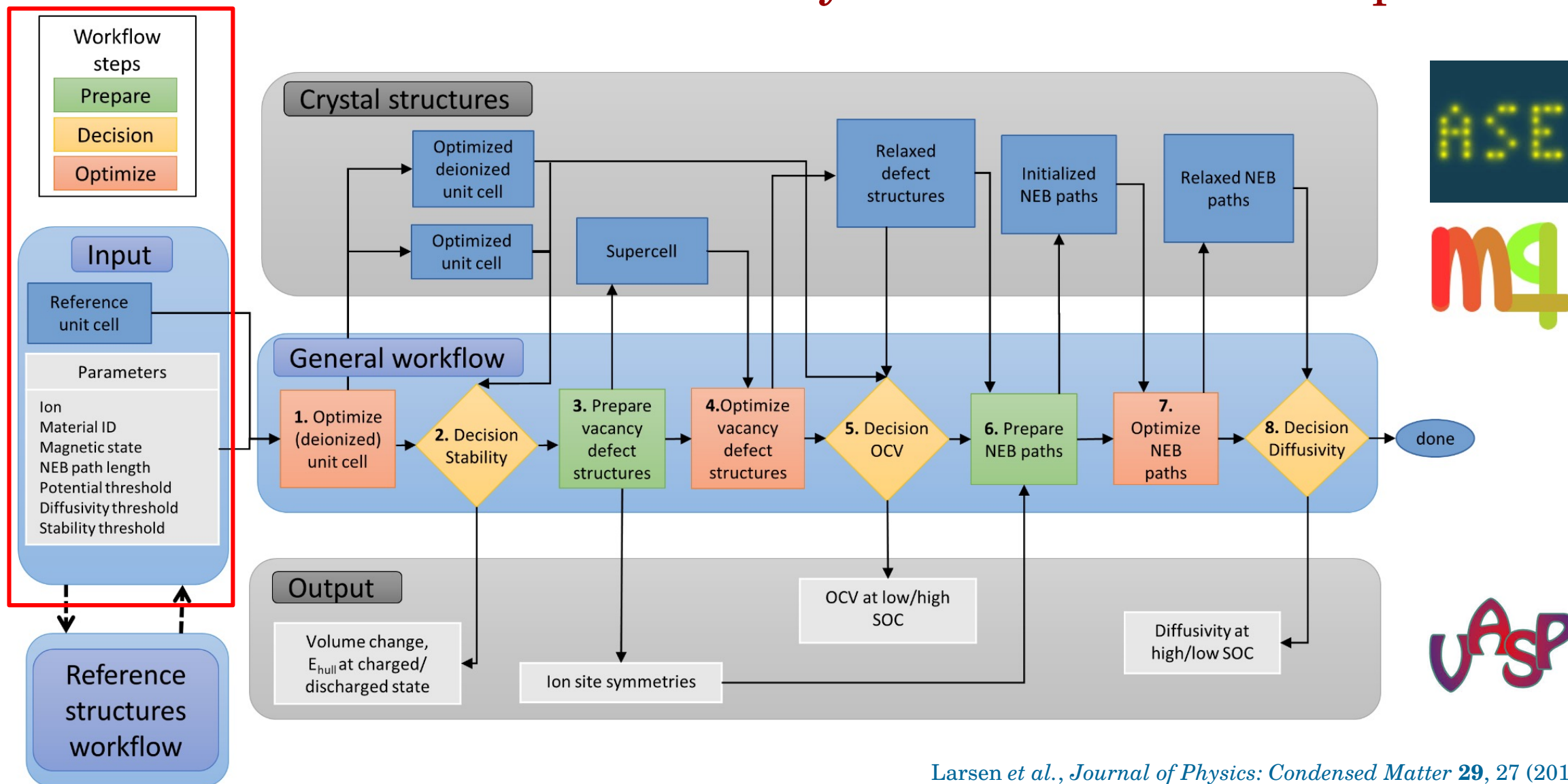
Automate the calculations of key thermodynamics and kinetic properties

Accelerated screening of known (ICSD/Materials project) or unknown materials (structure prototype)





Autonomous Workflow for Thermodynamic and Kinetic Properties

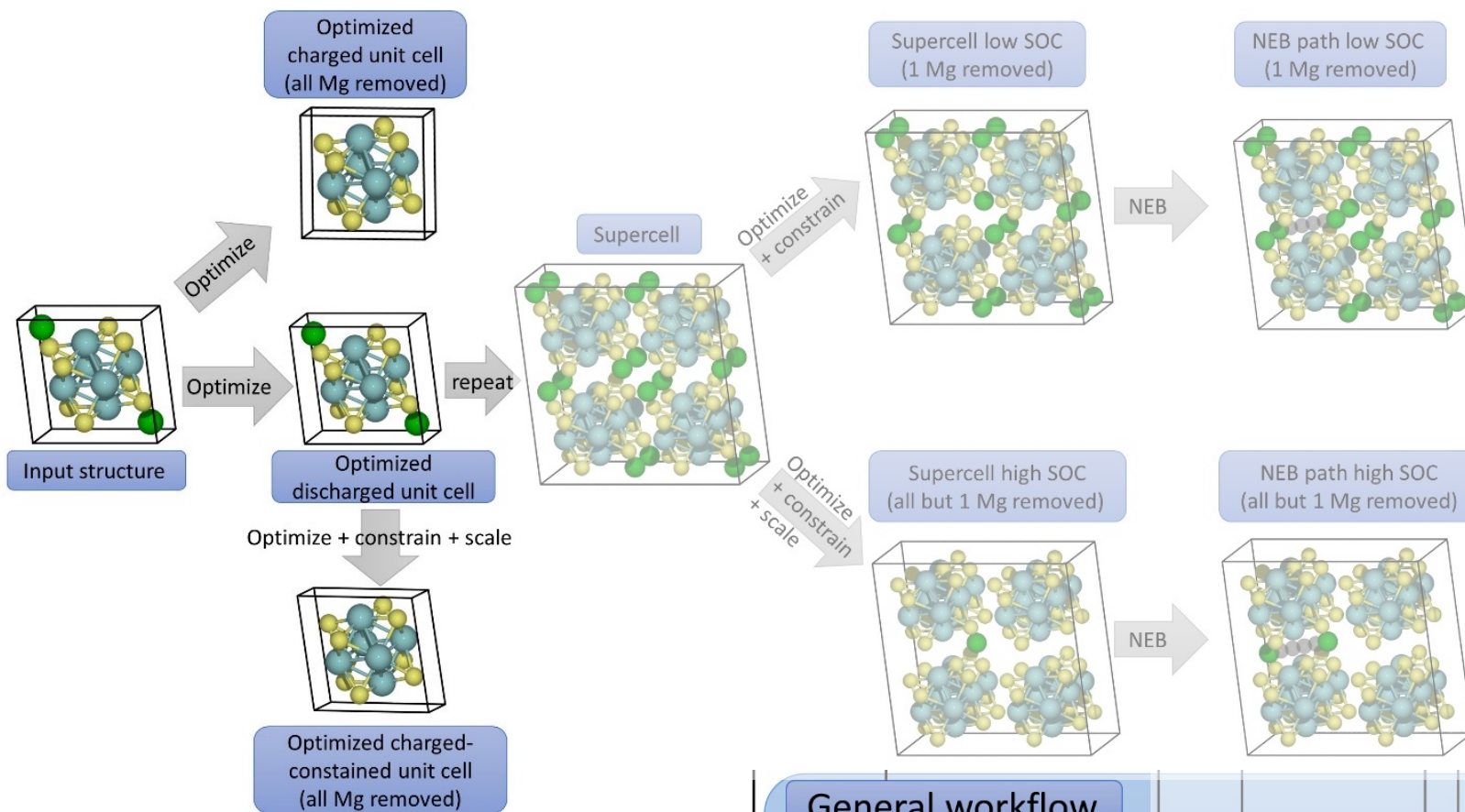


Larsen et al., *Journal of Physics: Condensed Matter* **29**, 27 (2017)

Bölle, Mathiesen, Nielsen, Vegge, Garcia Lastra, and Castelli, *BatteriesSupercaps* **3**, 488 (2021)

Mortensen et al., *Journal of Open Source Software* **5**, 1844 (2020)

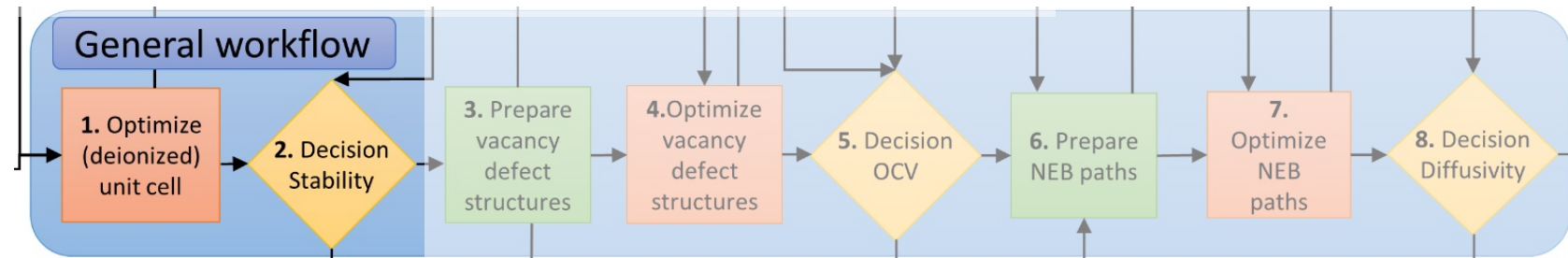
DTU Preparation of the Structures



9 unique structures to optimize

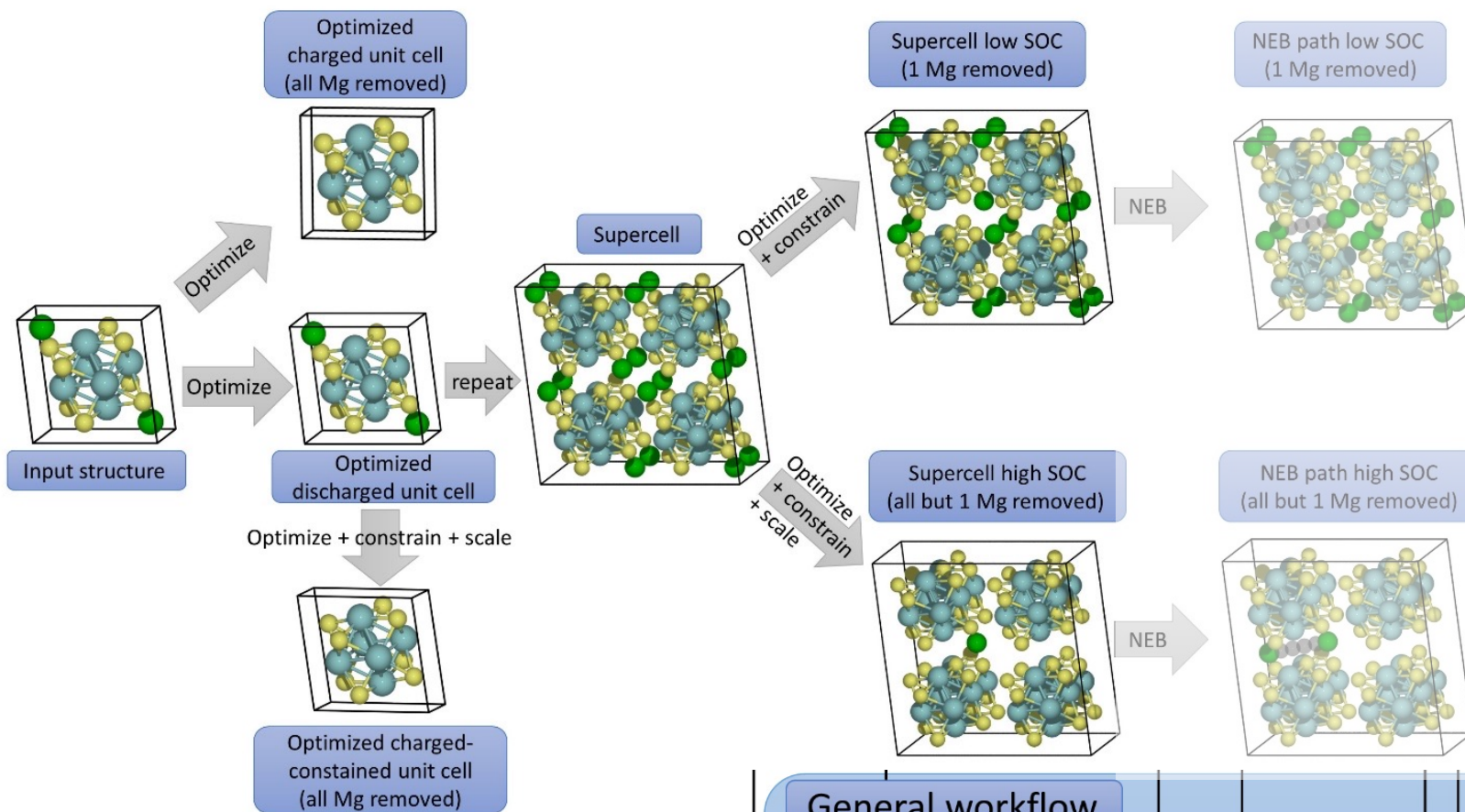
Properties:

1) Stability (< 0.2 eV/atom)



Bölle, Mathiesen, Nielsen, Vegge, Garcia Lastra, and Castelli, *BatteriesSupercaps* **3**, 488 (2021)

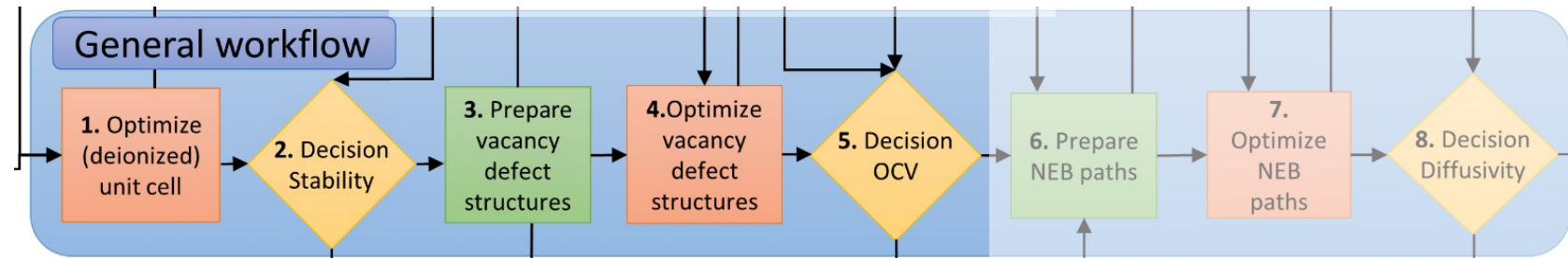
DTU Preparation of the Structures



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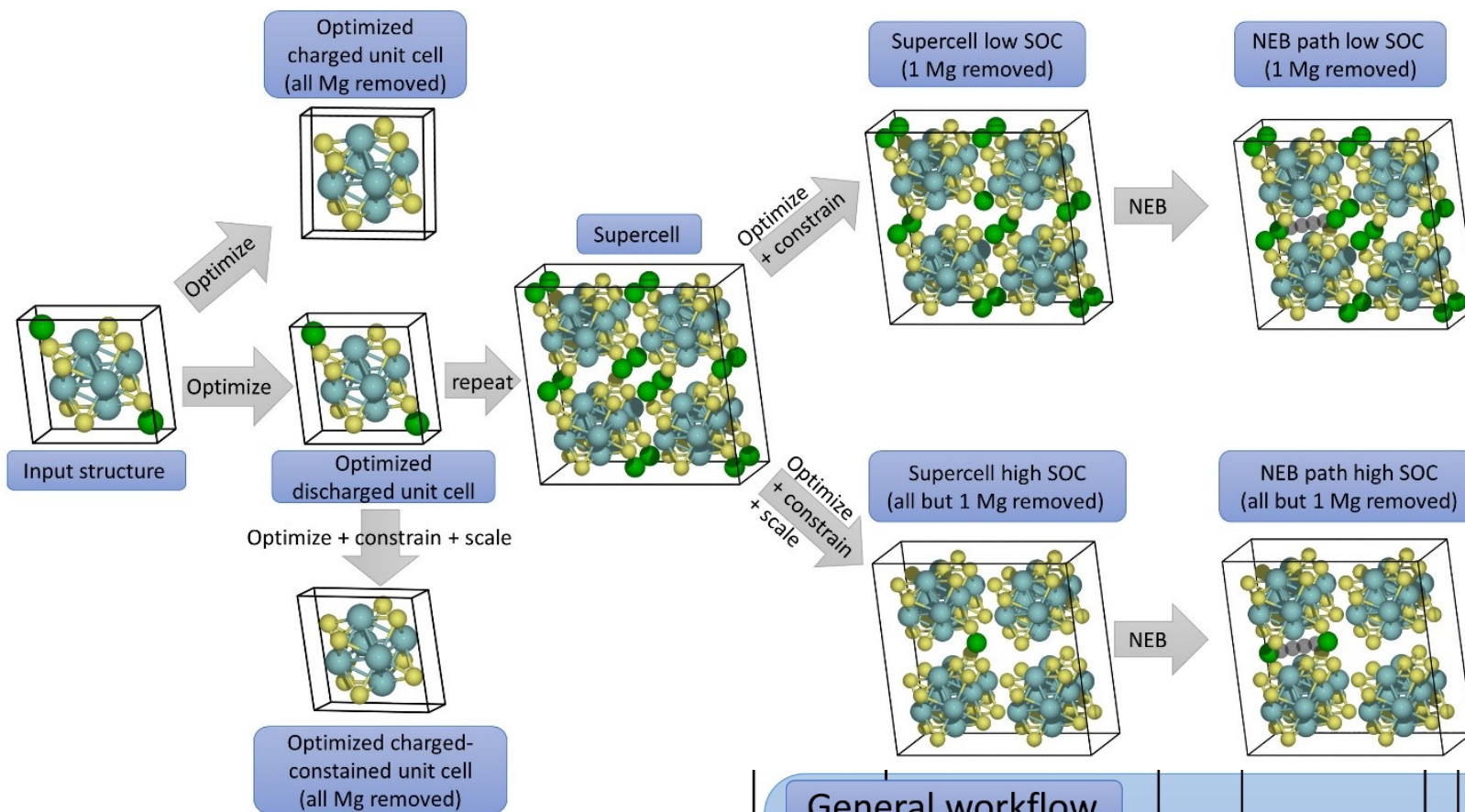
Properties:

- 1) Stability (< 0.2 eV/atom)
- 2) OCV (better than state-of-the-art)



Bölle, Mathiesen, Nielsen, Vegge, Garcia Lastra, and Castelli, *BatteriesSupercaps* **3**, 488 (2021)

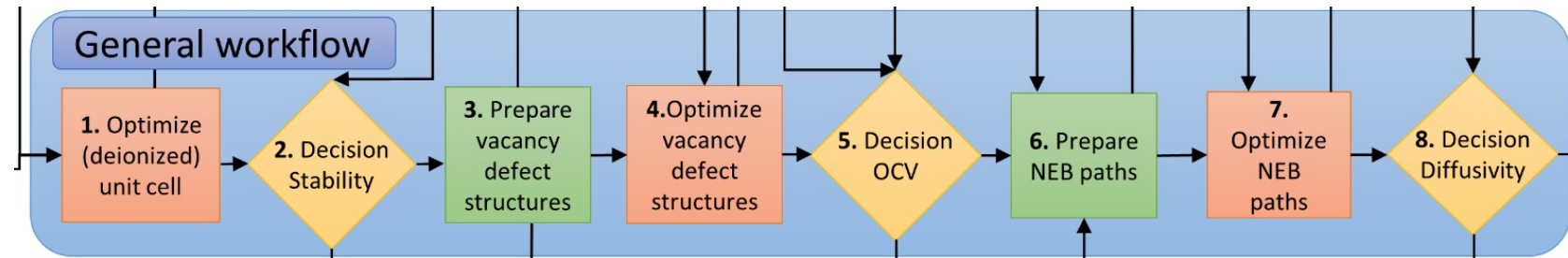
DTU Preparation of the Structures



9 unique structures to optimize

Properties:

- 1) Stability (< 0.2 eV/atom)
- 2) OCV (better than state-of-the-art)
- 3) Diffusion (Nudged Elastic Bands Method, NEB)



Bölle, Mathiesen, Nielsen, Vegge, Garcia Lastra, and Castelli, *BatteriesSupercaps* **3**, 488 (2021)

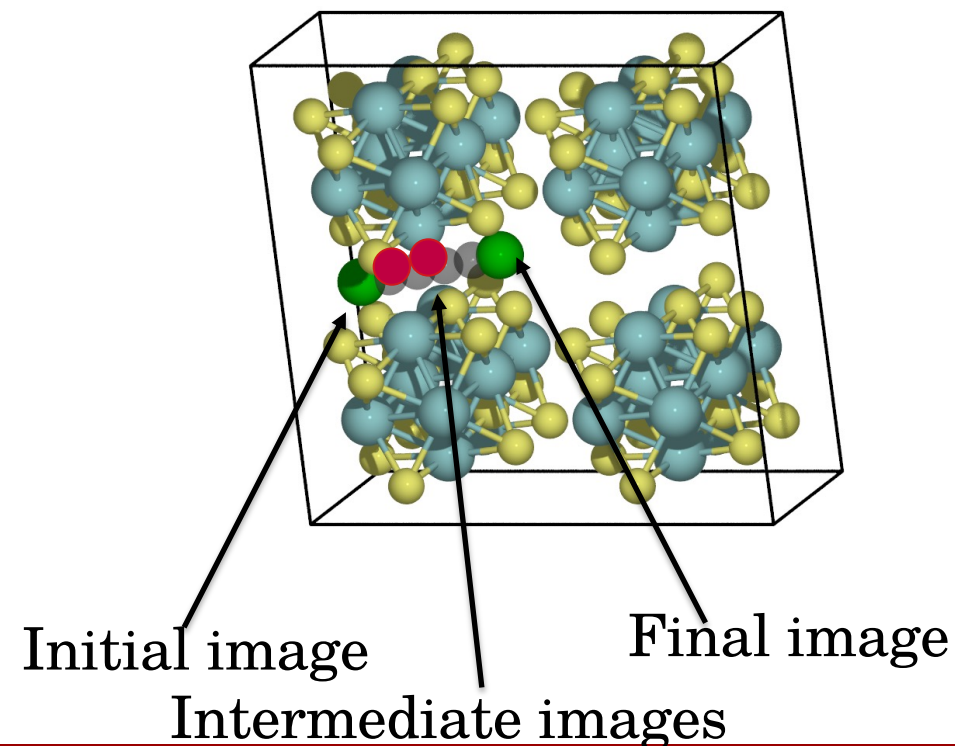
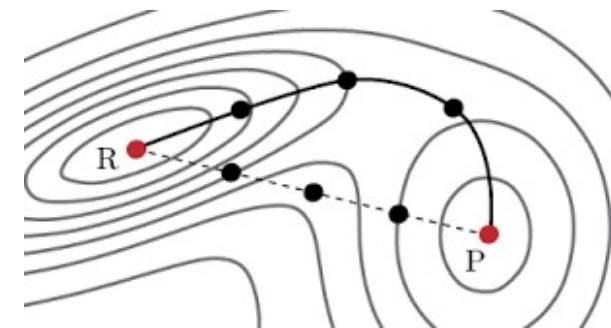
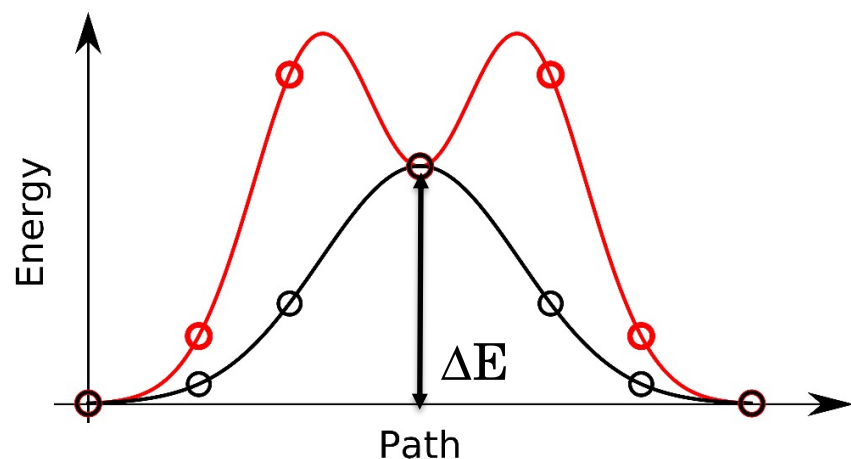
DTU NEB Method

Find transition states and minimum energy paths between reactants (initial) and products (final)

Expensive as many images needs to be calculated

Accelerated NEBs: use symmetries to reduce the number of images

Conventional NEB



Henkelman and Jonsson, *J. Chem. Phys.* **113**, 9978 (2000)

Mathiesen *et al.*, *Journal of chemical theory and computation* **15**, 3215 (2019)

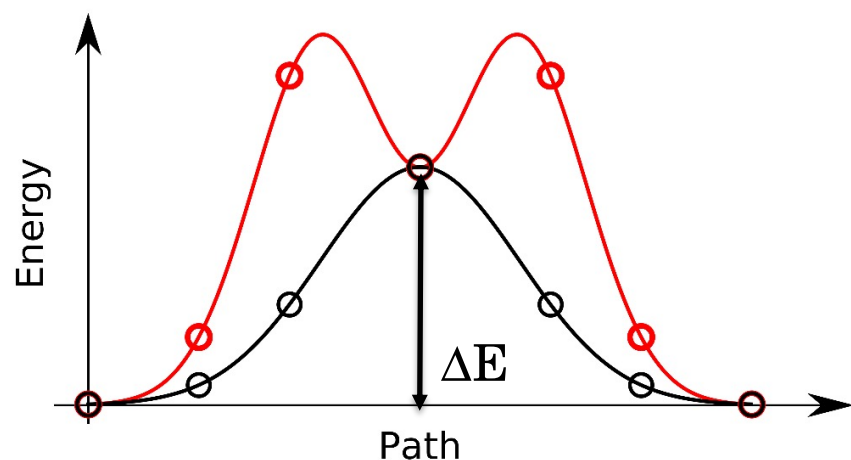
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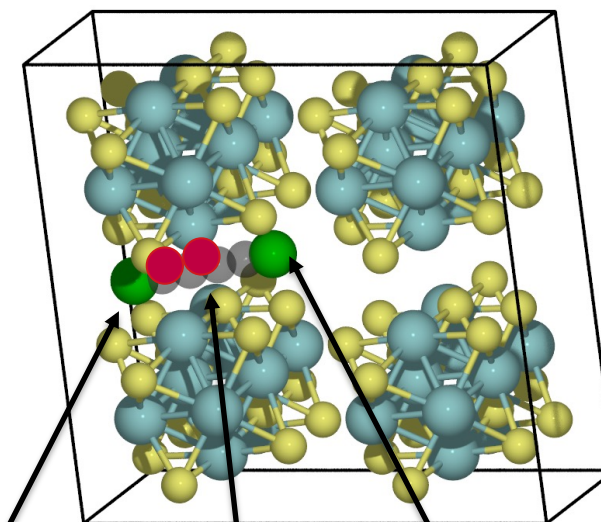
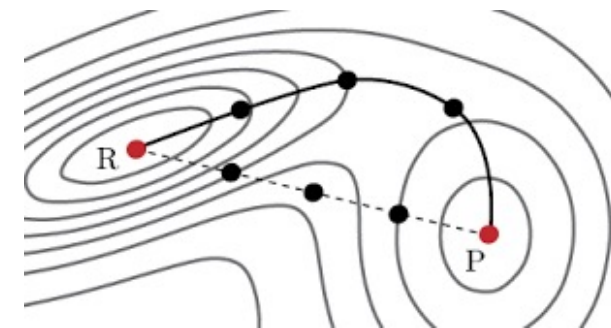
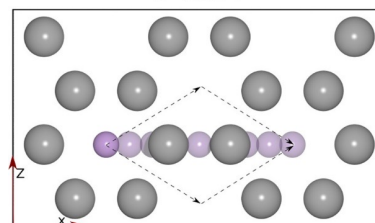
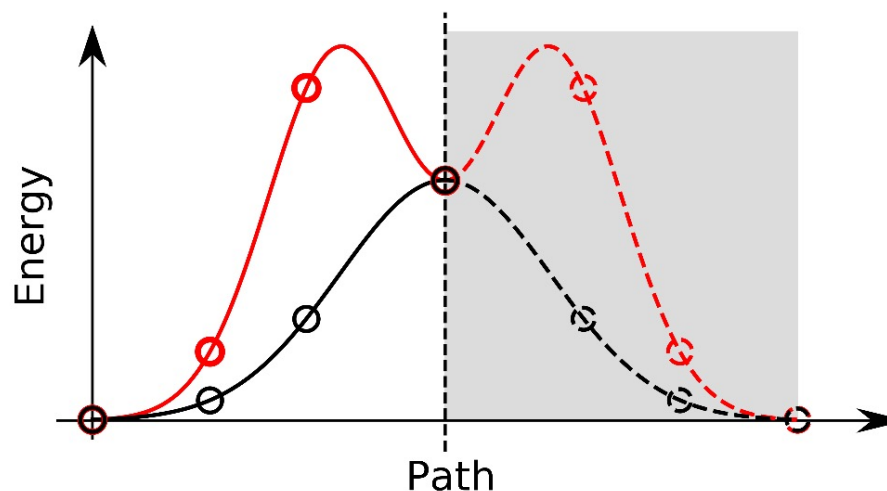
Accelerated NEBs: use symmetries to reduce the number of images

Conventional NEB



Speed-up: 2

Reflective NEB



Initial image

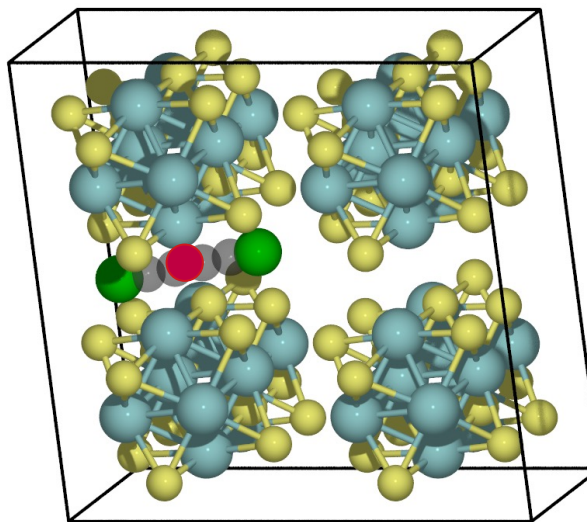
Final image

Intermediate images

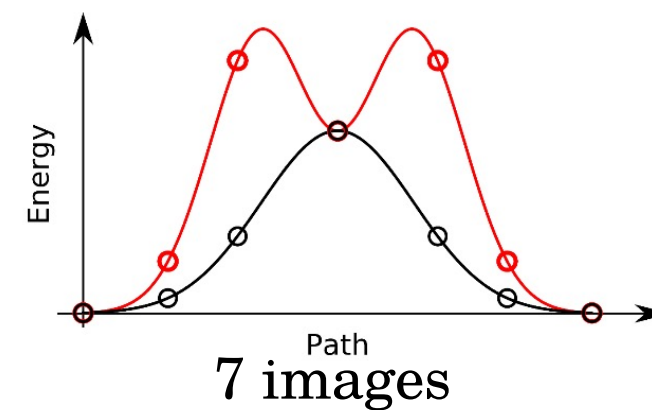
Henkelman and Jonsson, *J. Chem. Phys.* **113**, 9978 (2000)

Mathiesen *et al.*, *Journal of chemical theory and computation* **15**, 3215 (2019)

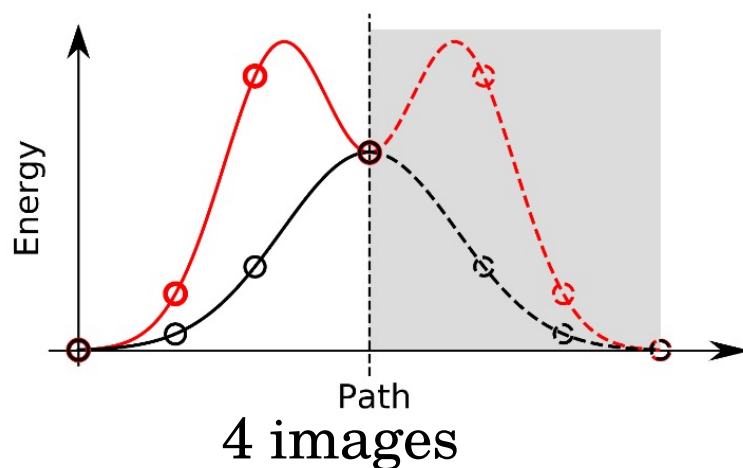
Climbing image (find transition state)



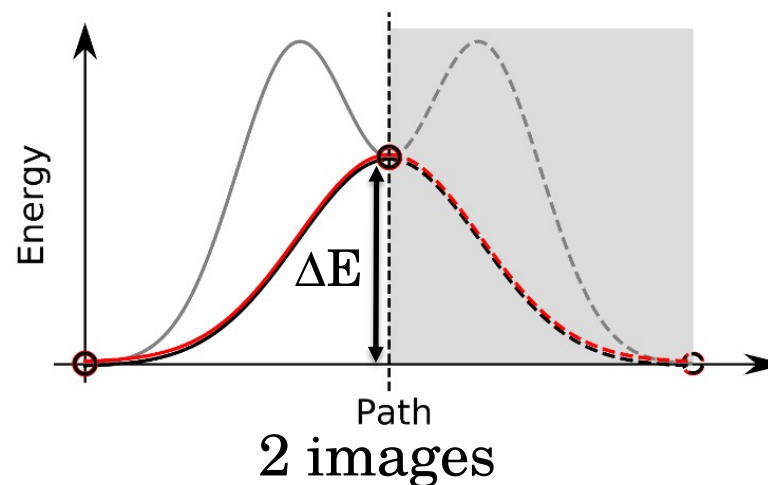
Conventional NEB



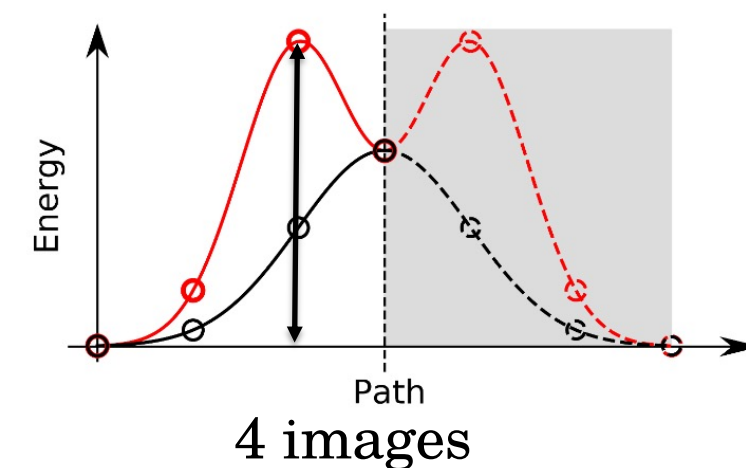
Reflective NEB (RNEB)



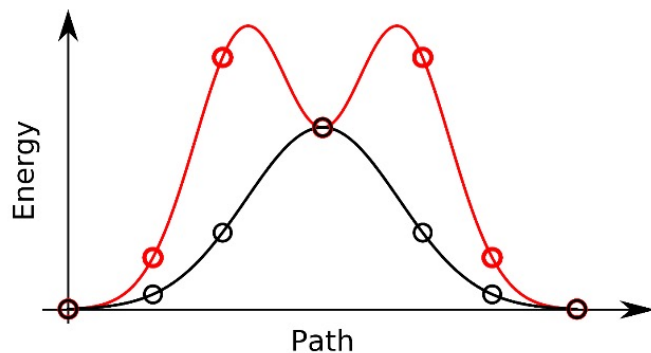
Reflective middle-image (RMI) NEB



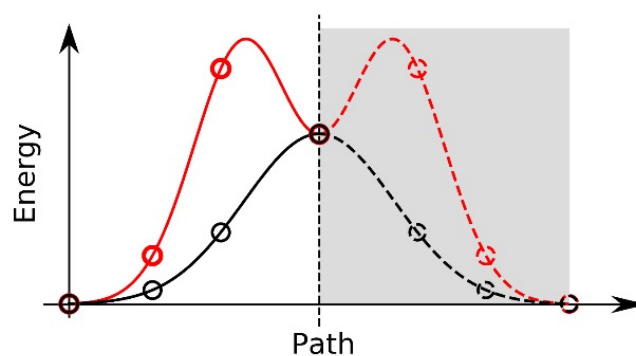
Climbing image RNEB



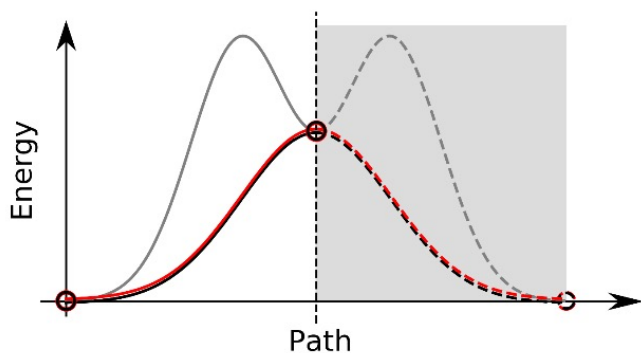
Conventional NEB



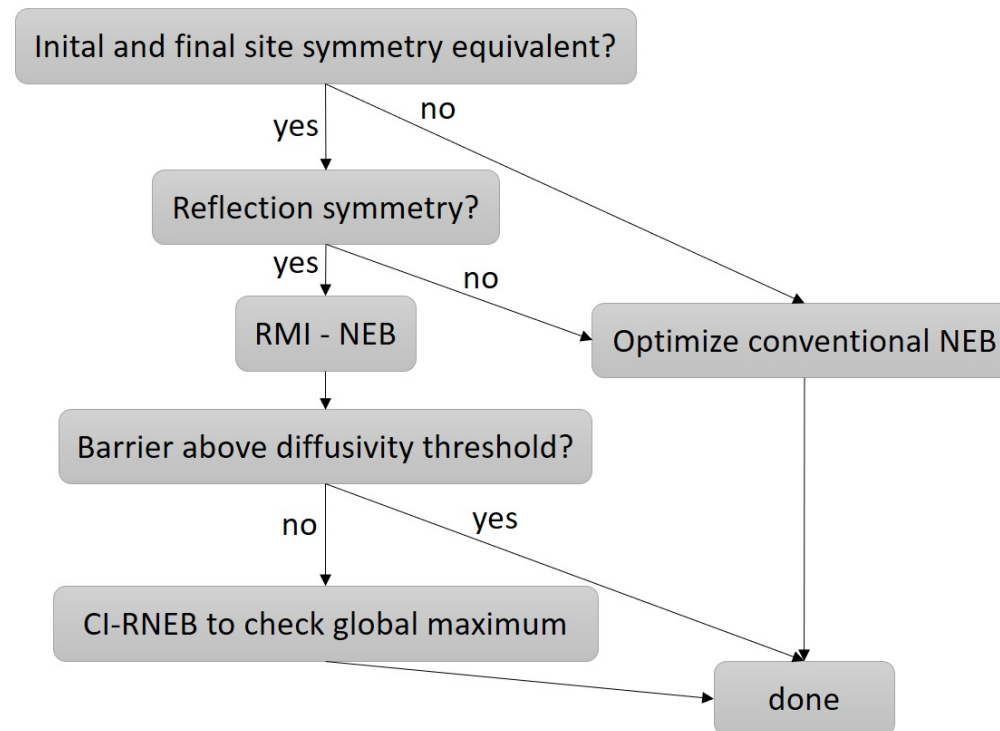
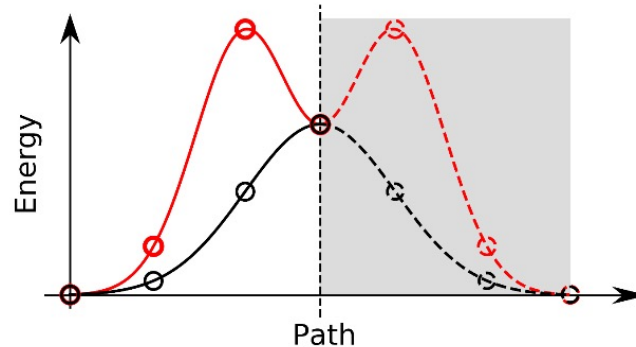
RNEB



RMI - NEB



CI - RNEB



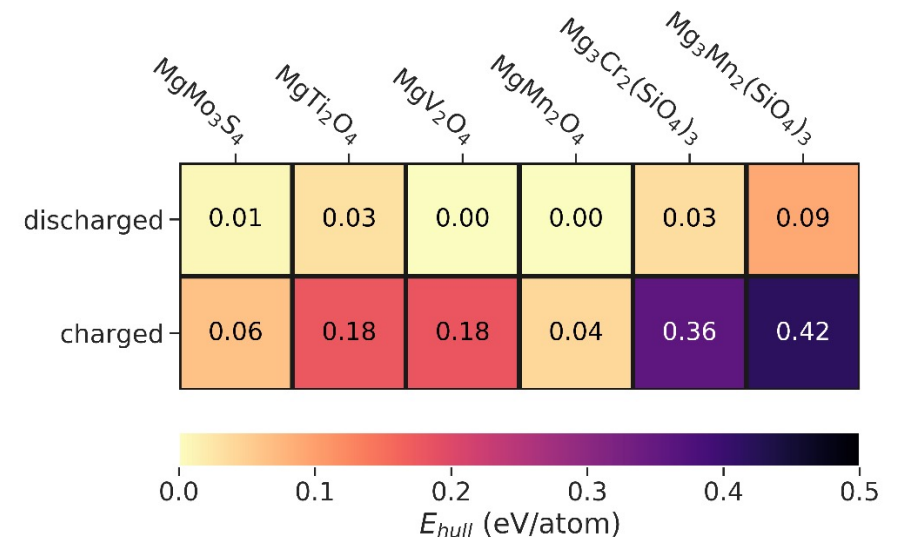
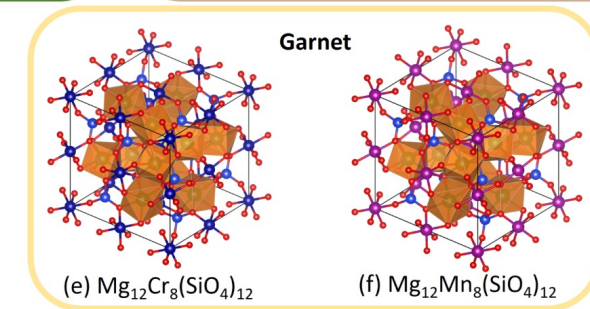
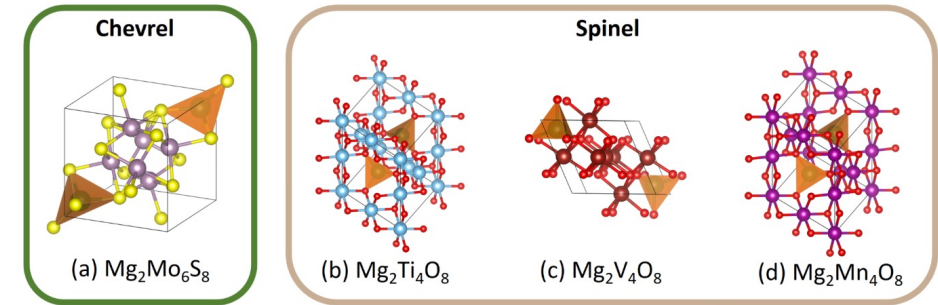
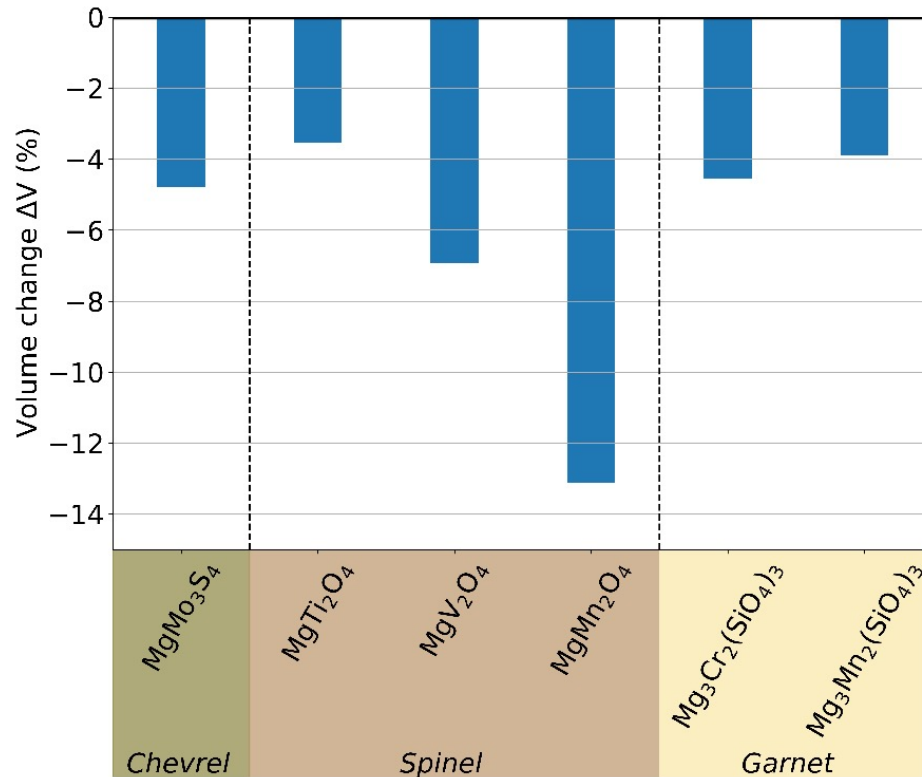
CAN WE ACCELERATE THIS EVEN FURTHER????

More details: Bølle – Wed 17th, 11.50

Candidate Materials from the ICSD Database

~100 Materials from the ICSD database

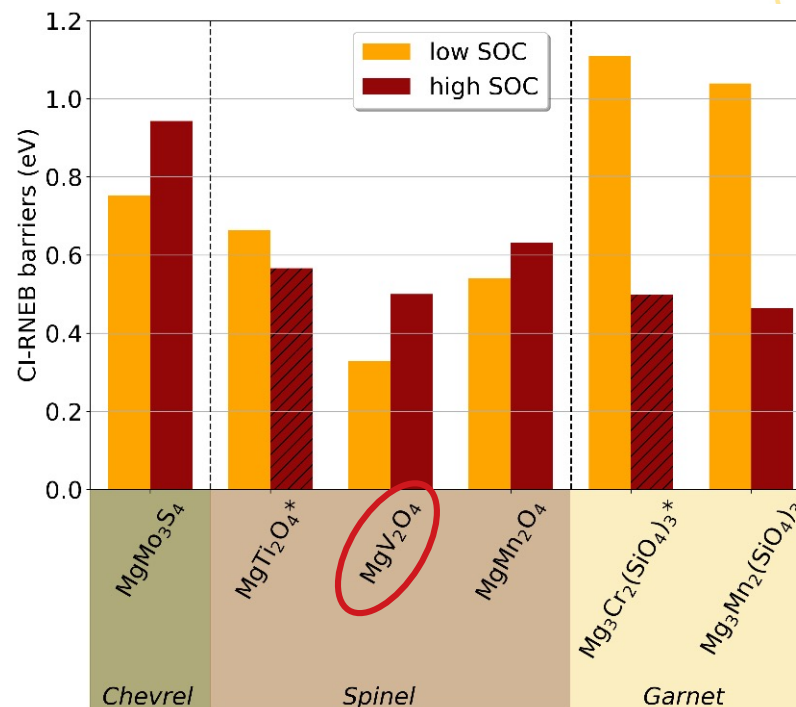
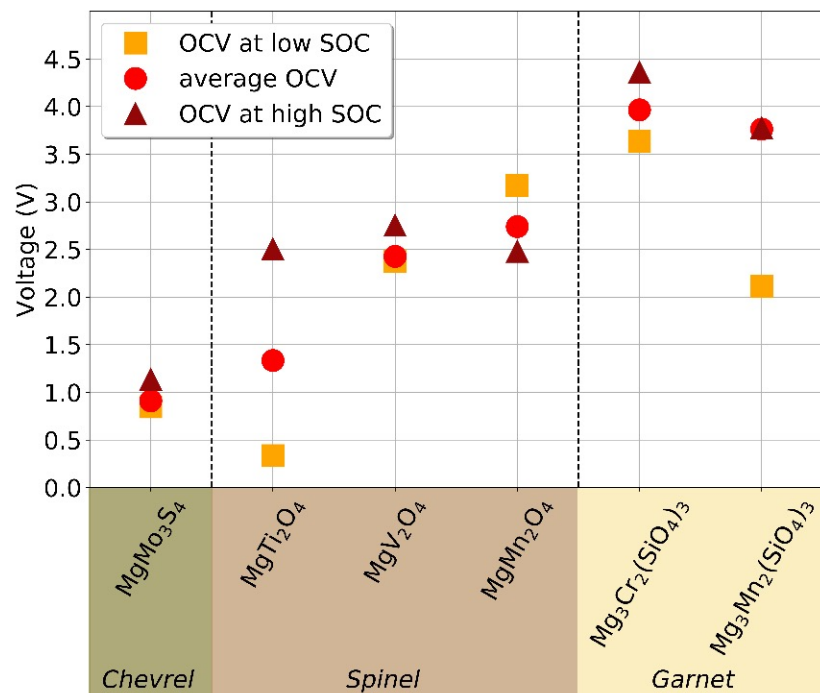
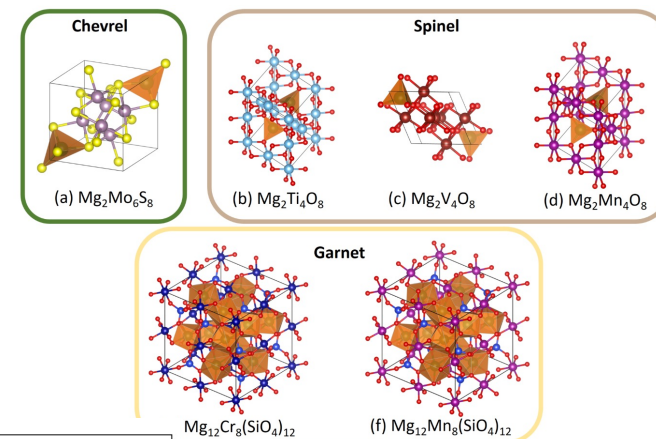
1) stability and volume change



Candidate Materials from the ICSD Database

~100 Materials from the ICSD database

- 1) stability and volume change
- 2) OCVs
- 3) Diffusion Barriers

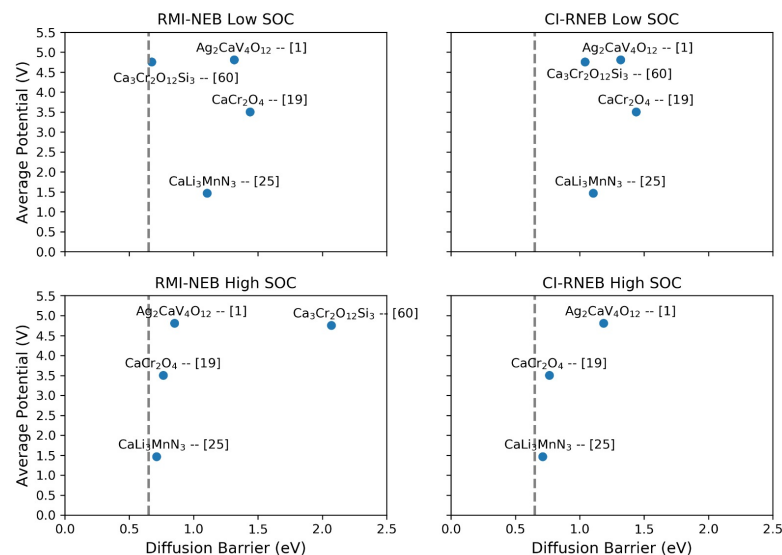
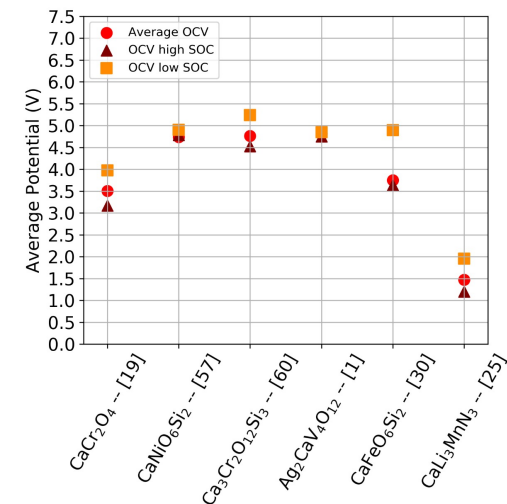
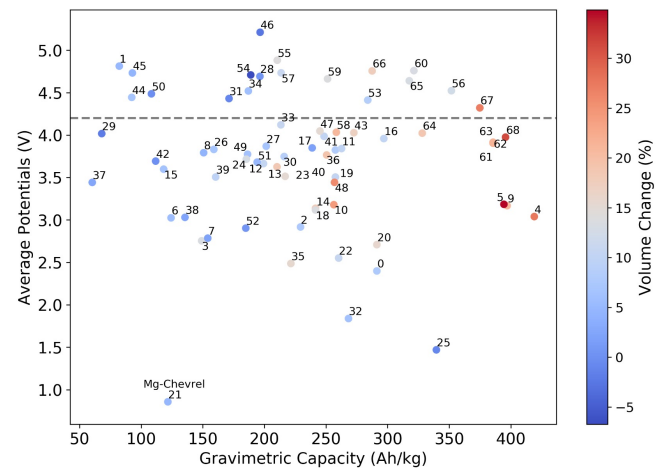
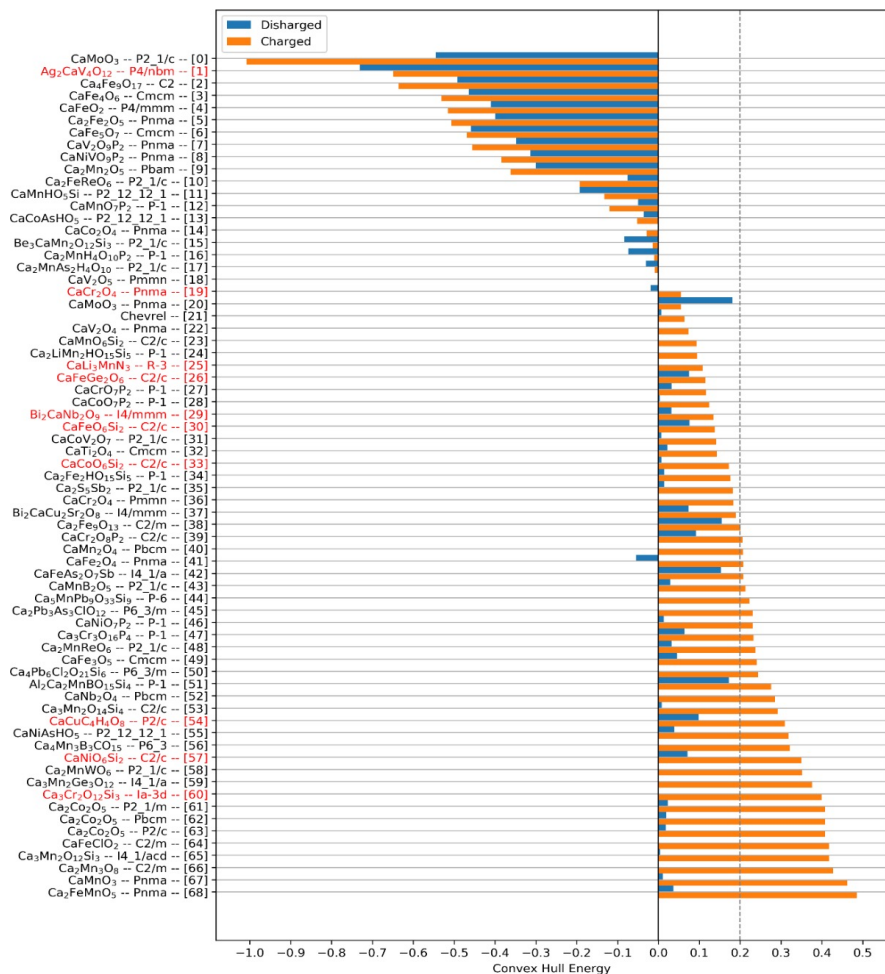




Re-use the Workflow: Ca-ion Batteries

Abundance, multivalent ion, redox potential

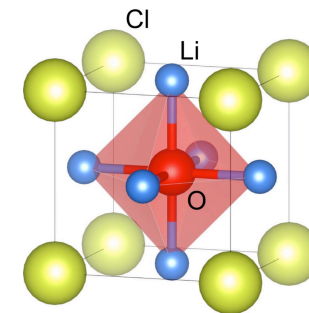
~250 Materials from the ICSD database





Re-use the Workflow: Antiperovskites for Solid-State Batteries

Higher capacity and safer devices – Limitations: low conductivity



Antiperovskites (Li_3BrO , Li_3ClO ,...)

1) Electronic conduction

- Large band gaps

2) Ionic conductivity

- High ionic conductivity

3) Stability:

- Stability against lithium-metal anode
- Li_3BrO and Li_3ClO stable vs Li/Li^+ redox potential
- Often unstable in water

4) Mechanical properties

- Resistance to dendrite penetration

5) Sustainability

- It can accommodate diverse elements (Li, Na mostly synthesized)

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Database-based selection:

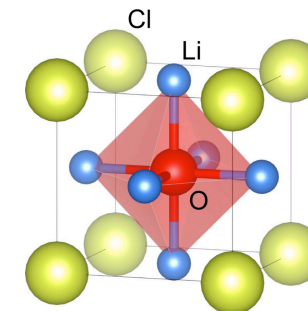
- Geometric filter
- Band gap value
- Sustainability constituent elements

Thermodynamic stability:

- Energy above the convex hull

Ionic Conductivity:

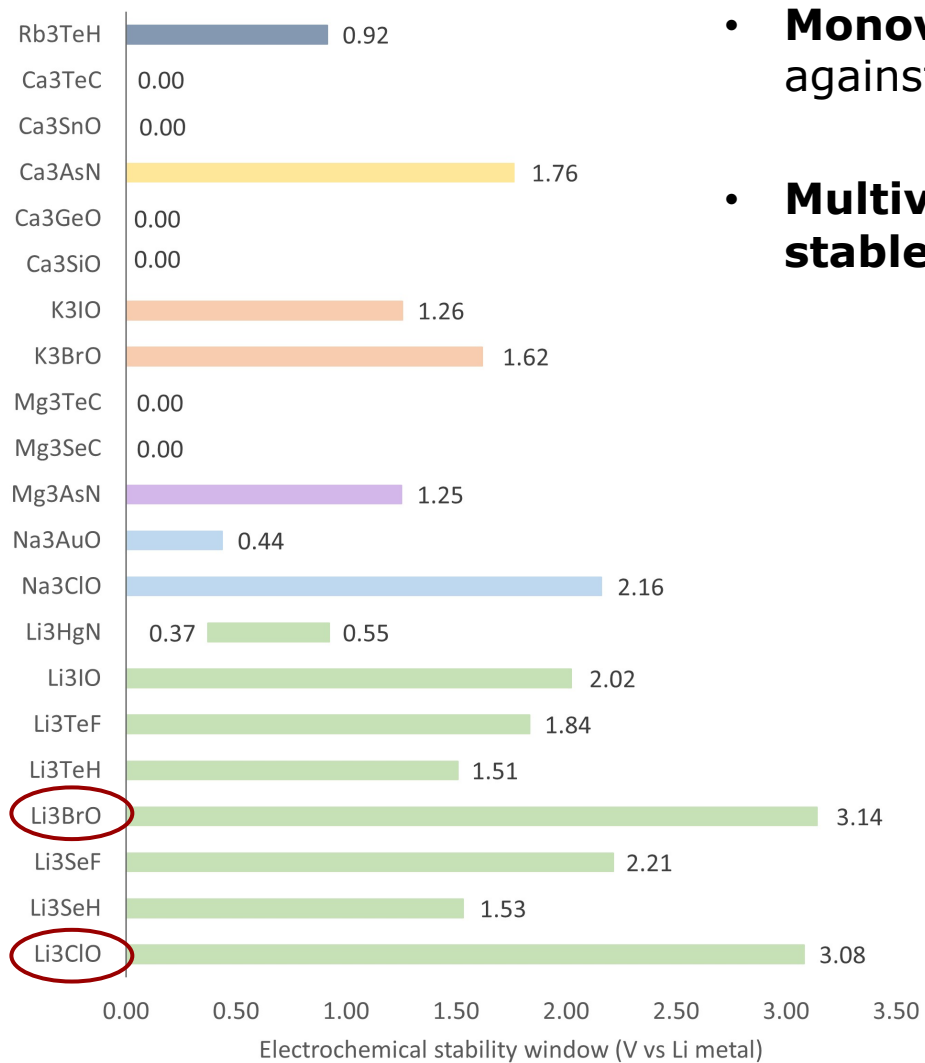
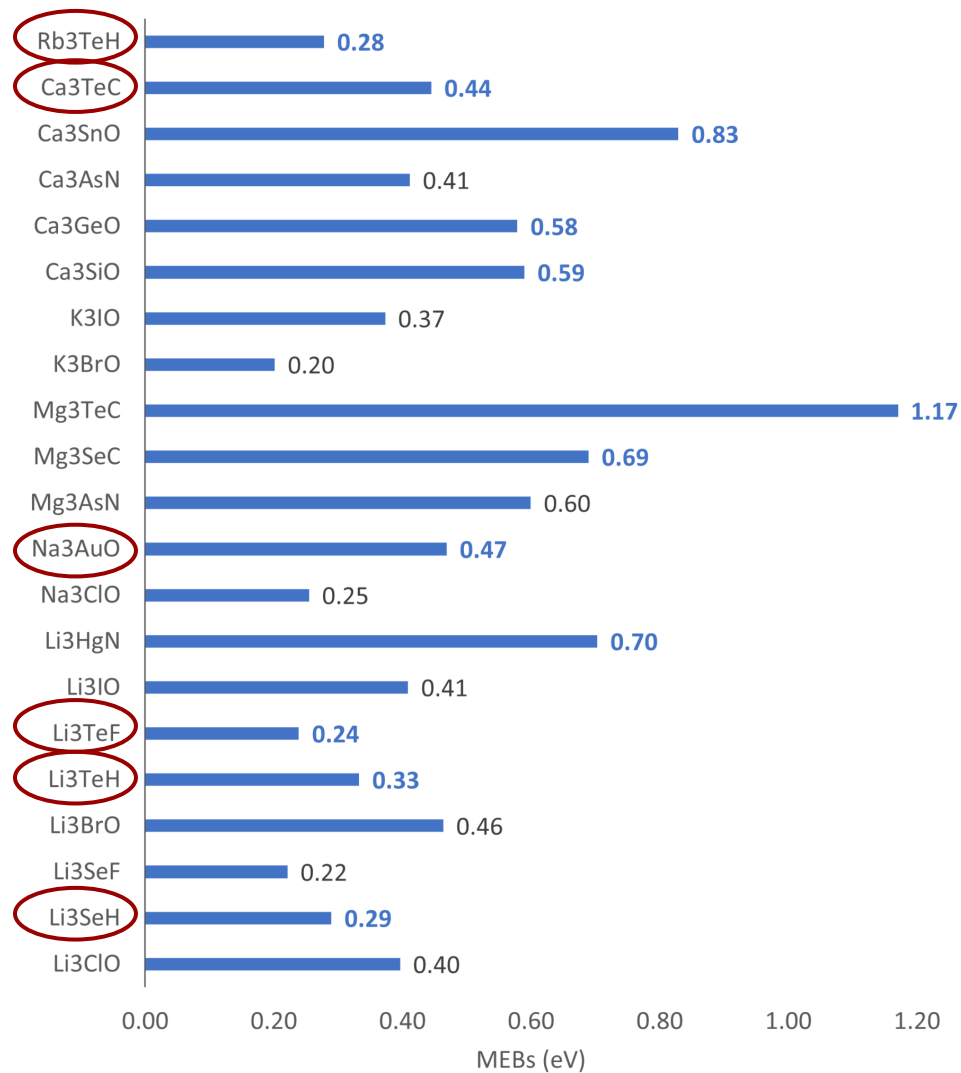
- Migration energy barriers (NEB methods within DFT)



Stability under operating conditions:

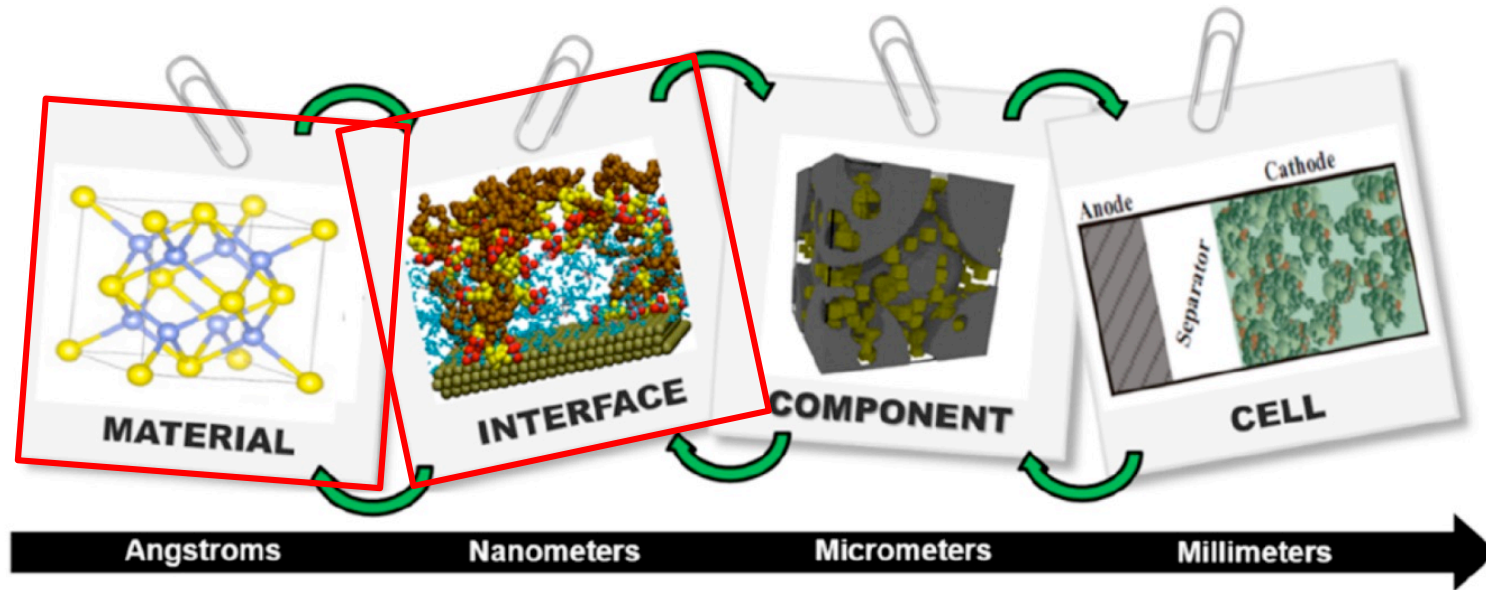
- Real band gap
- Stability with moisture
- Electrochemical stability window vs metal anode

Migration Barriers and Electrochemical Stability



- **Monovalent cation stable** against cation-metal
- **Multivalent cation less stable** against metal

Can We Implement Workflows for Interfaces?



Franco *et al.*, *Chem. Rev.* **119**, 4569 (2019)

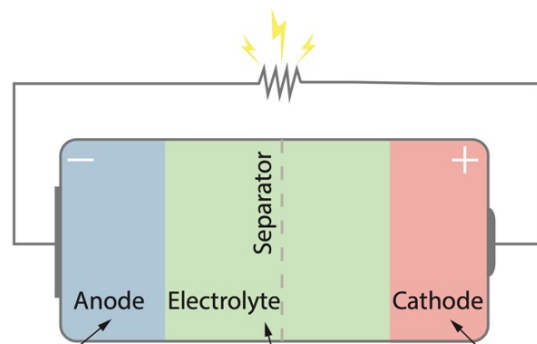
INTERFACES

GOD MADE THE BULK; ~~SURFACES~~ WERE
INVENTED BY THE DEVIL.

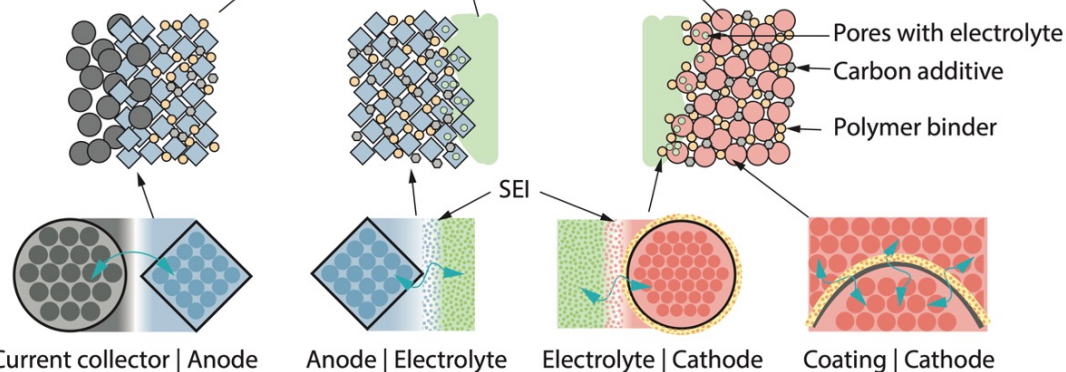
- WOLFGANG PAULI -

Can We Implement Workflows for Interfaces?

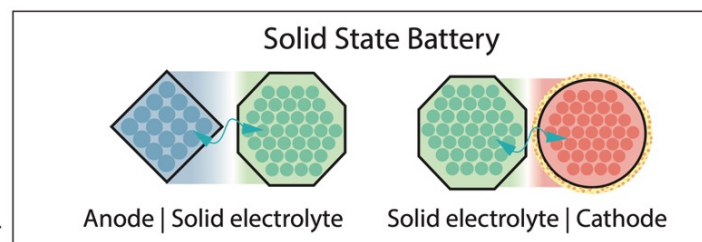
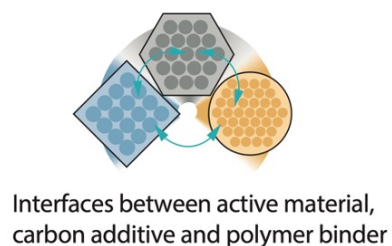
Device



Meso



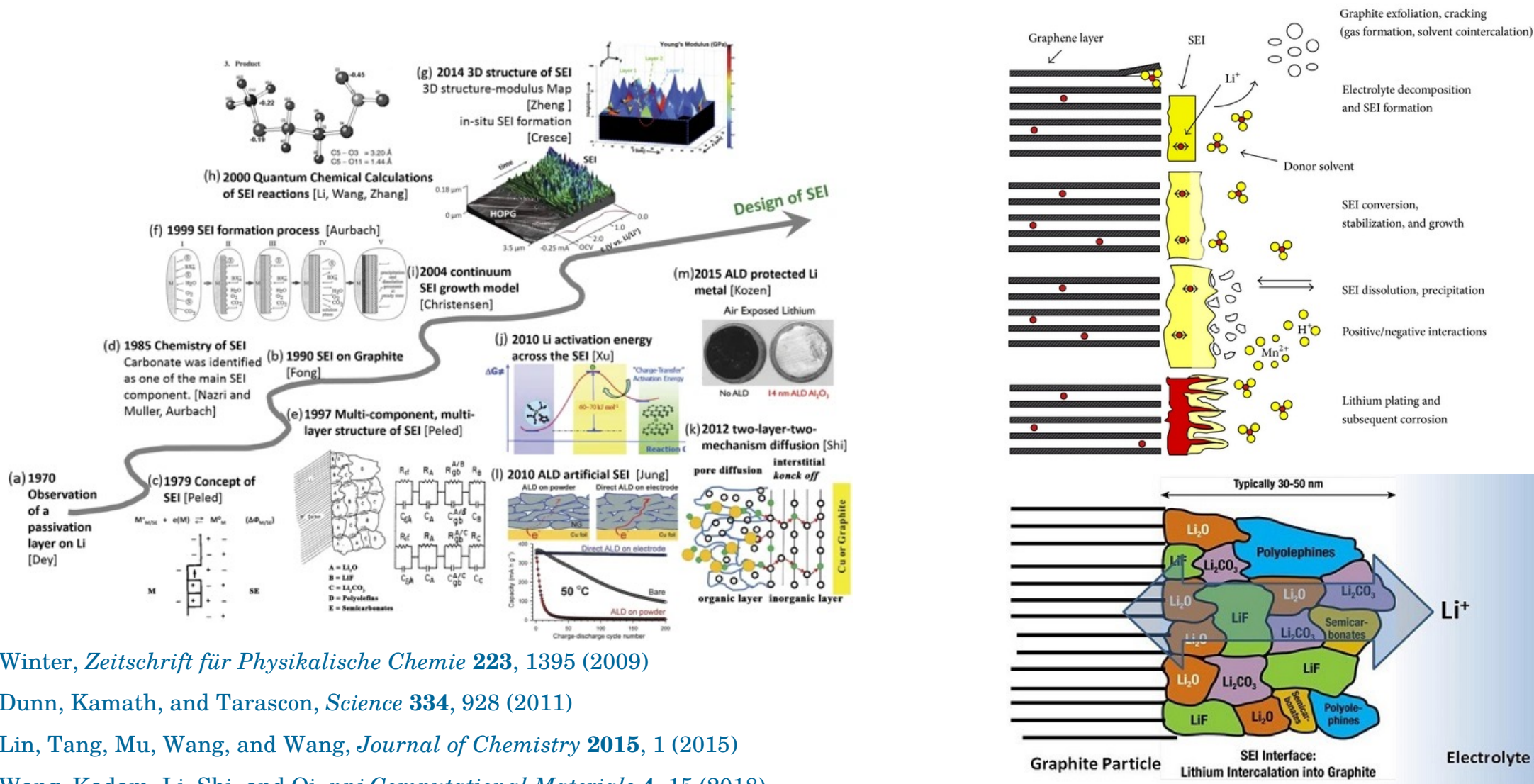
Atomic



Materials Project – https://www.materialsproject.org/ – Provenance: USA					
	Entries	App	Battery	Workflow	API
Total Entries	711,561	Phase Diagram	–		
Intercalation Electrodes	4,730	Reaction Calculator	YES	Fireworks	YES
Conversion Electrodes	16,128	Battery explorer	YES	Automate Pymatgen Custodian	
Liquid Electrolytes	49,705	Explore Molecule Redox Flow Batteries	YES		
Bulk Interfacial Reactivity	N.A.	Interface Reaction (Bulk)	YES		
The Open Quantum Materials Database – OQMD – http://oqmd.org – Provenance: USA					
Total entries	815,654	Phase Diagram	NO	OQMD	YES
Automatic FLOW for Materials Discovery – http://www.aflowlib.org – Provenance: USA					
Total entries	3,482,348	Phase Diagram (AFLOW-Chull)	NO	AFLOW	YES
Novel Materials Discovery – NOMAD – https://nomad-lab.eu/ – Provenance: Germany/EU					
Total entries	>11,000,000	Artificial Intelligence Toolkit	NO	Qmpy	YES
Materials Cloud – https://www.materialscloud.org – Provenance: Switzerland/EU					
Total entries	7,502,686	–	YES	AiiDA	YES
BIG-MAP App Store – https://big-map.github.io/big-map-registry – Provenance: Denmark/EU					
Total entries	N.A.	BattINFO DFT-Surface	YES	AiiDA SimStack	YES
ARTISTIC Computational Portal – https://www.erc-artistic.eu/computational-portal – Provenance: France/EU					
Total entries	>111 (open since July 2021)	Online Manufacturing Simulator Data Explorer INNOV	YES	ARTISTIC	NO

Deng, Kumar, Bølle, Caro, Franco, Castelli, Canepa, Seh, under review (2021)

The Case of the Solid Electrolyte Interphase



Winter, *Zeitschrift für Physikalische Chemie* **223**, 1395 (2009)

Dunn, Kamath, and Tarascon, *Science* **334**, 928 (2011)

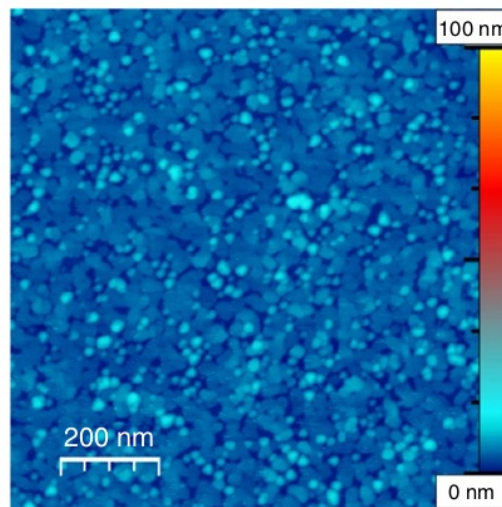
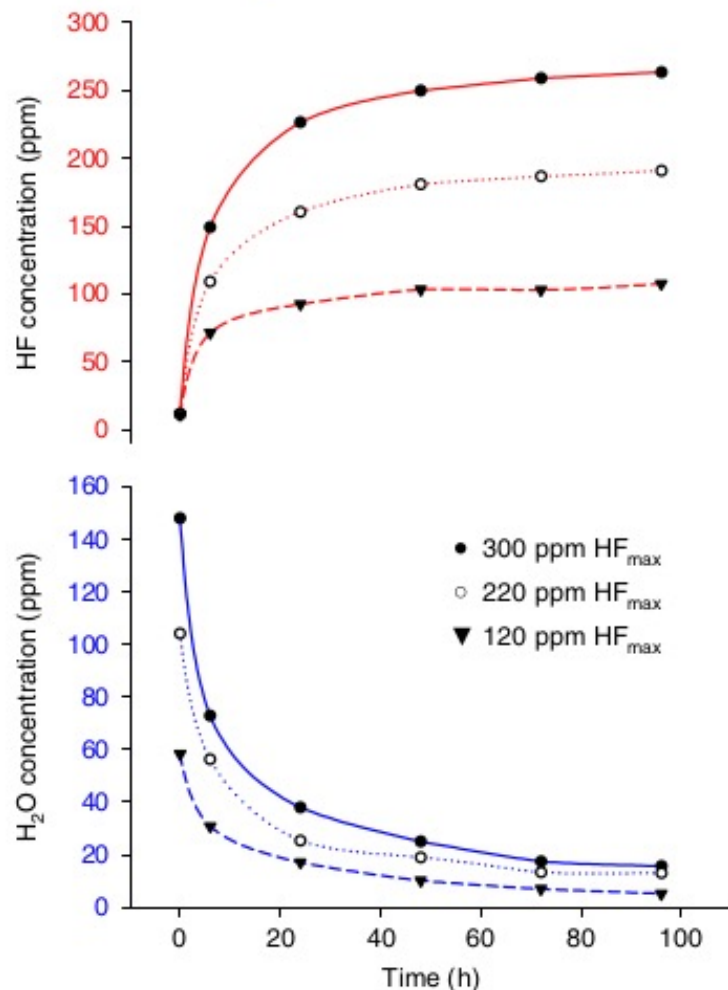
Lin, Tang, Mu, Wang, and Wang, *Journal of Chemistry* **2015**, 1 (2015)

Wang, Kadam, Li, Shi, and Qi, *npj Computational Materials* **4**, 15 (2018)

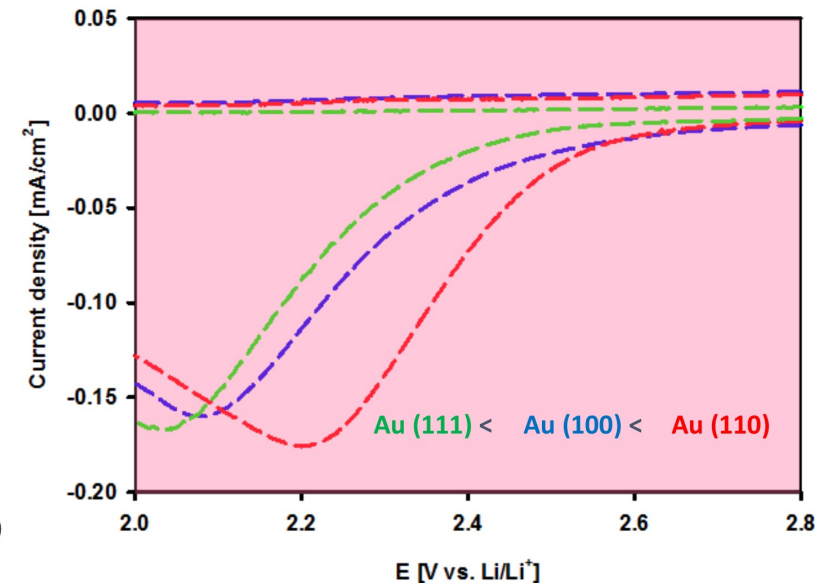
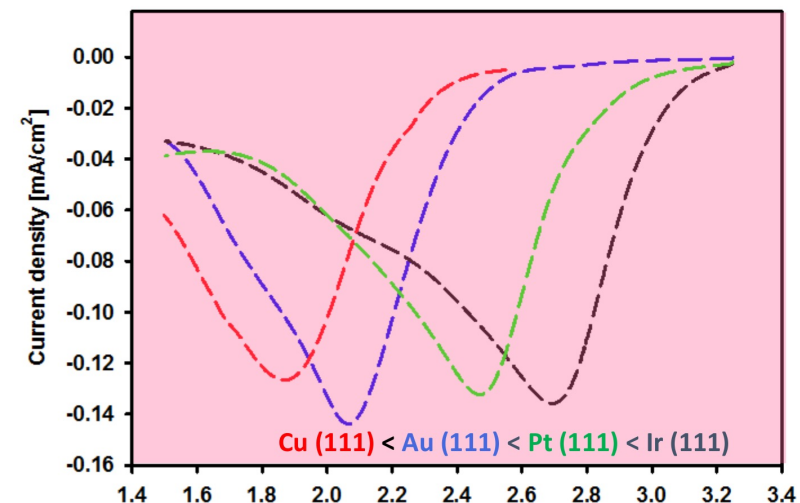
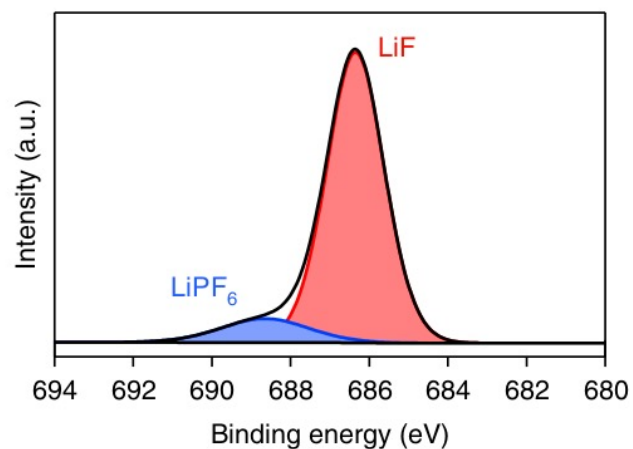
The SEI Formation on Metal Single Crystals

AFM: LiF@Au(111)

CV



XPS

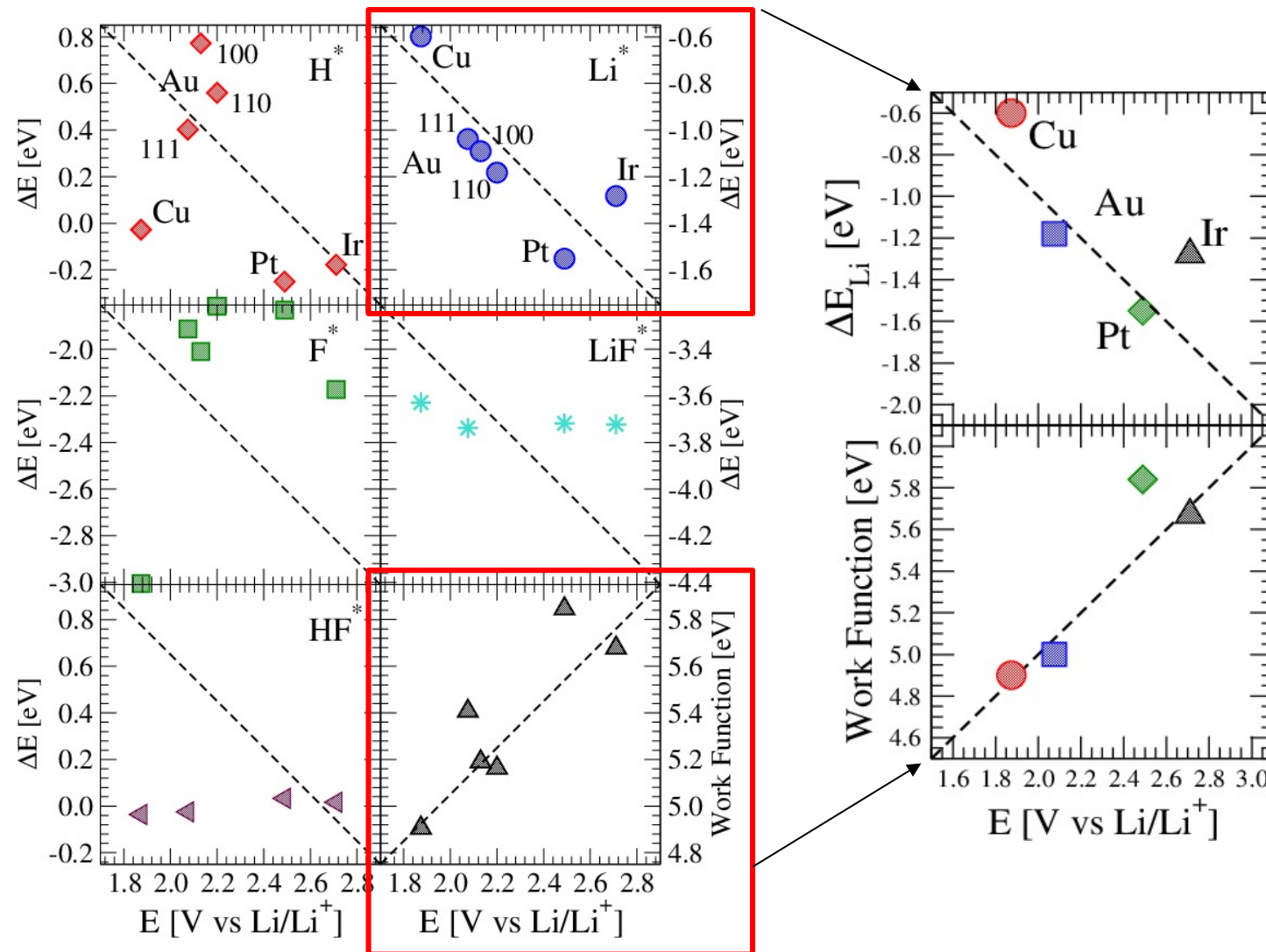
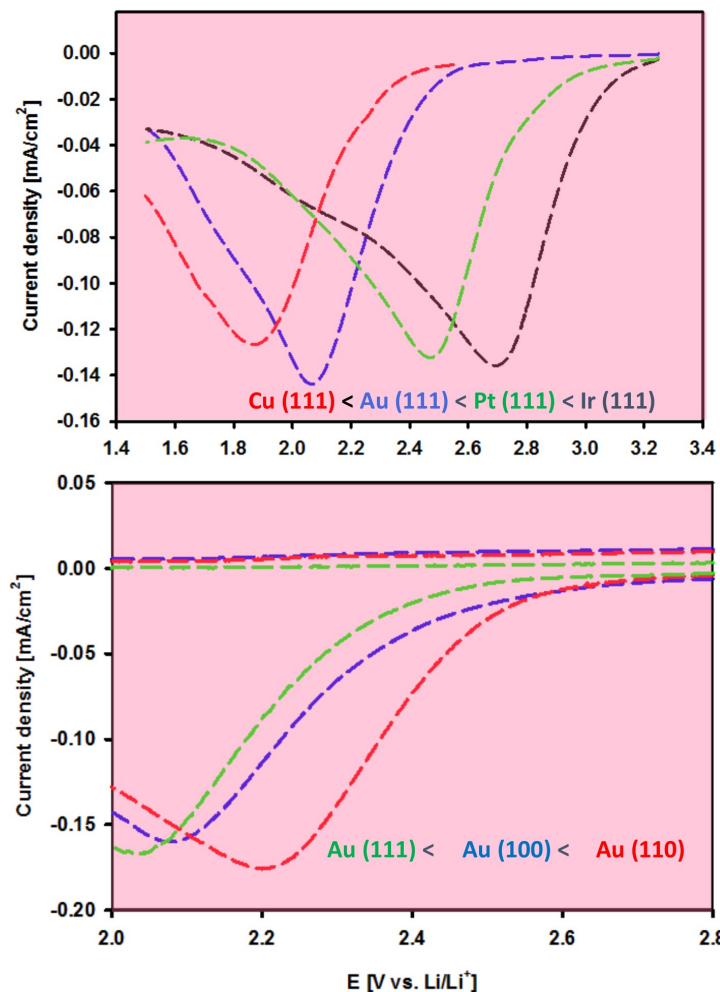


Electrolyte EC:EMC 3:7w (LP57)

Only impurity: HF

Strmcnik, Castelli, ..., Rossmeisl, and Markovic, *Nature Catalysis* 1, 255 (2018)

Possible Descriptors



What is the role played by Li and by the interface?

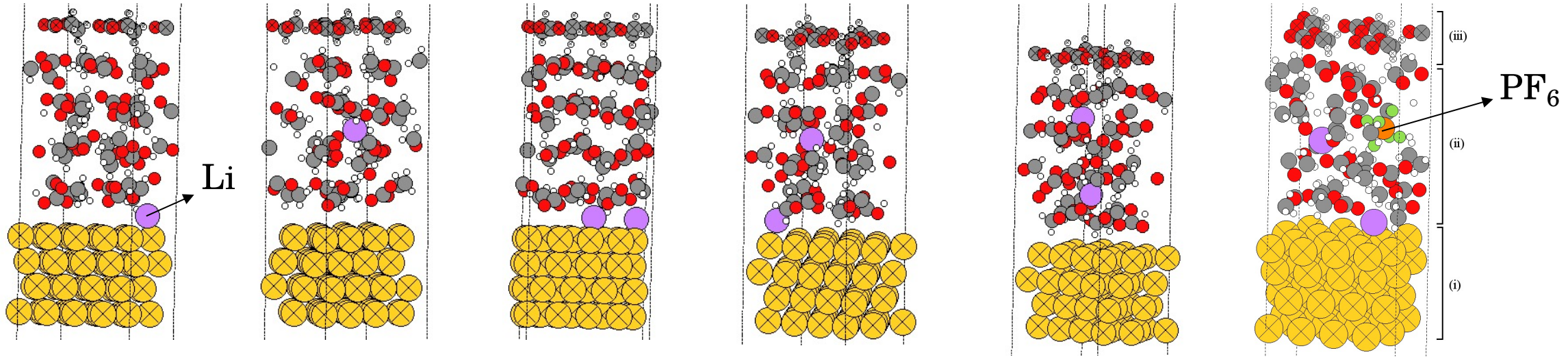
Strmcnik, Castelli, ..., Rossmeisl, and Markovic, *Nature Catalysis* 1, 255 (2018)



Explicit Modelling of the Solid/Electrolyte Interface

Combination of *molecular dynamics* and *DFT* (Ab-initio MD, AIMD)

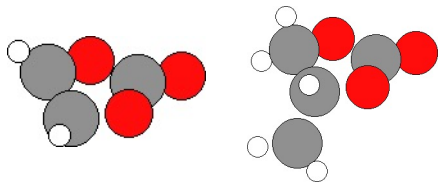
Study of the phase diagram of LP57 (EC:EMC 3:7w) with Li-atoms



Li coverages: 0 – 3 /16

Counter anion: PF_6

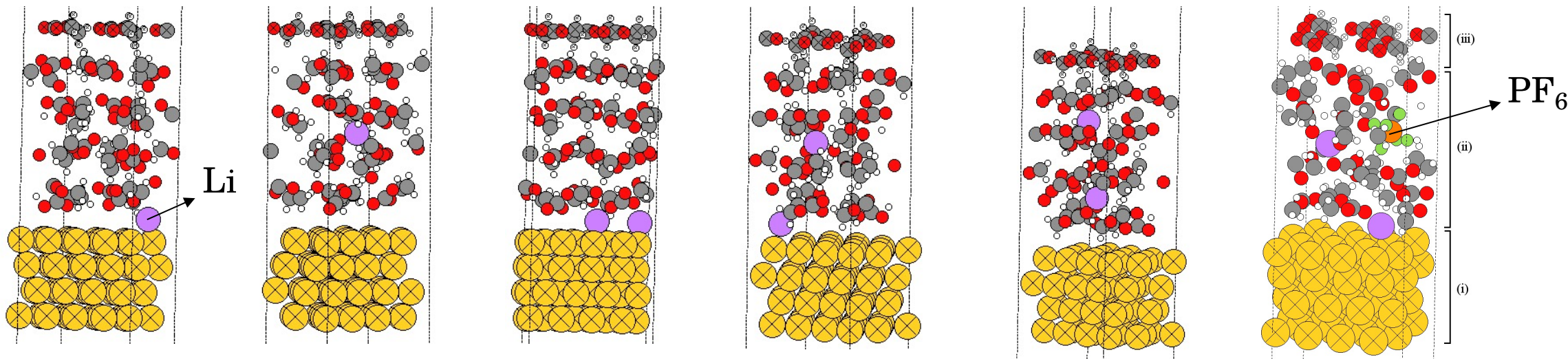
Additives: VC, PC, ...



Explicit Modelling of the Solid/Electrolyte Interface

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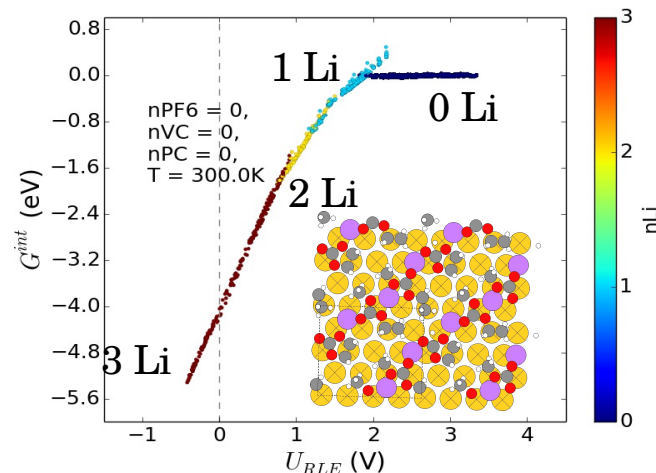
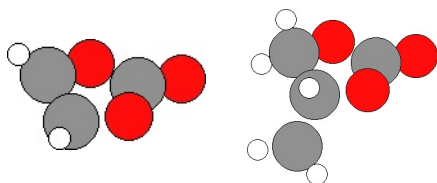
Study of the phase diagram of LP57 (EC:EMC 3:7w) with Li-atoms



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Counter anion: PF_6^-

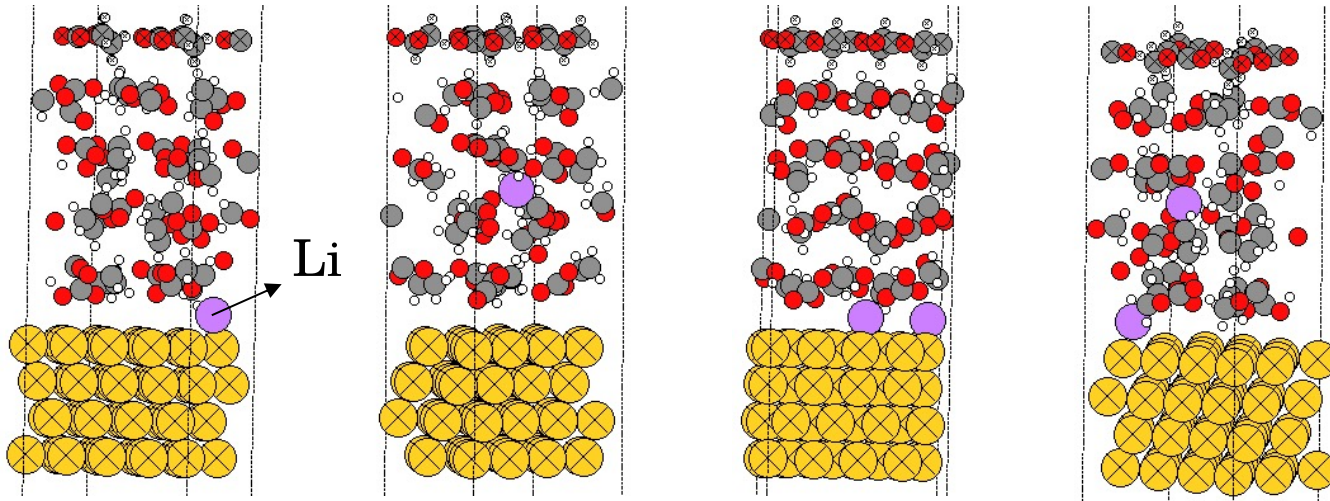
Additives: VC, PC, ...



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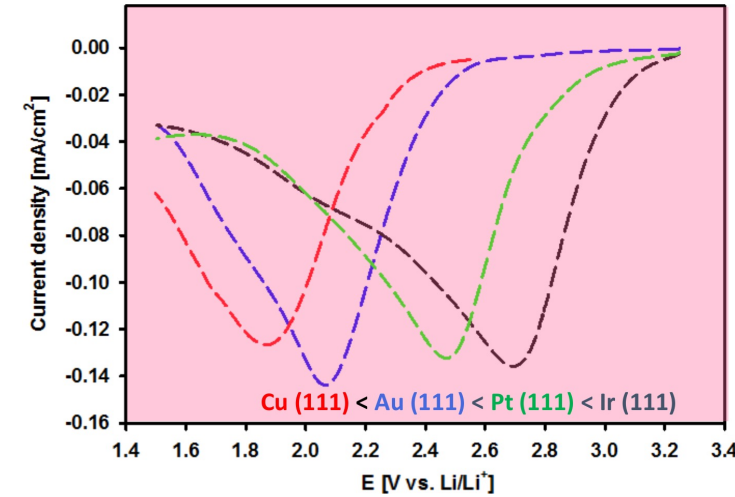
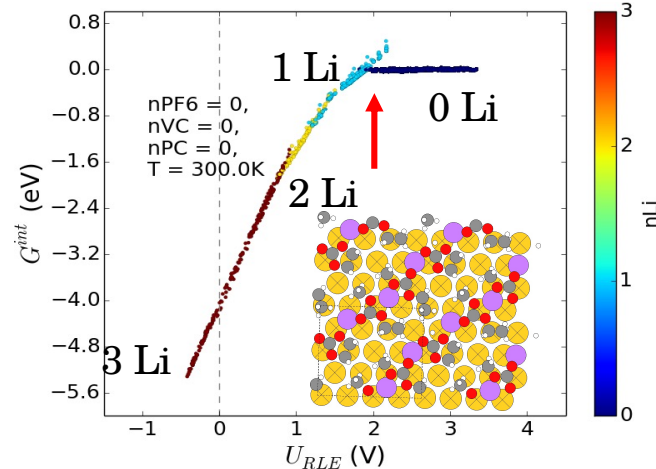
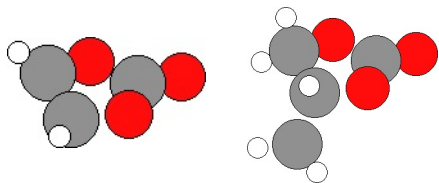
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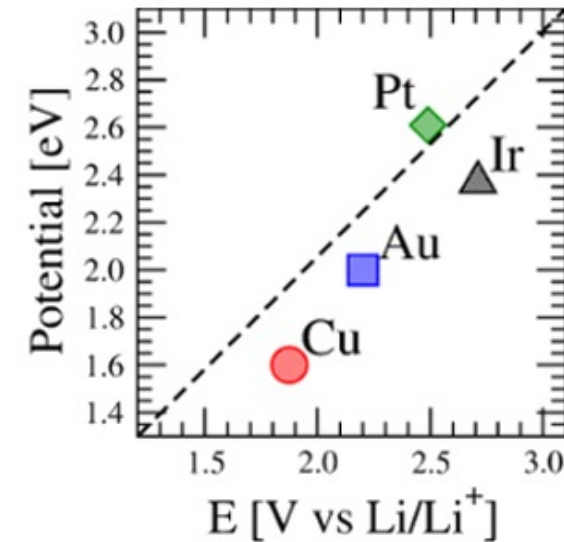
Li coverages: 0 – 3 /16

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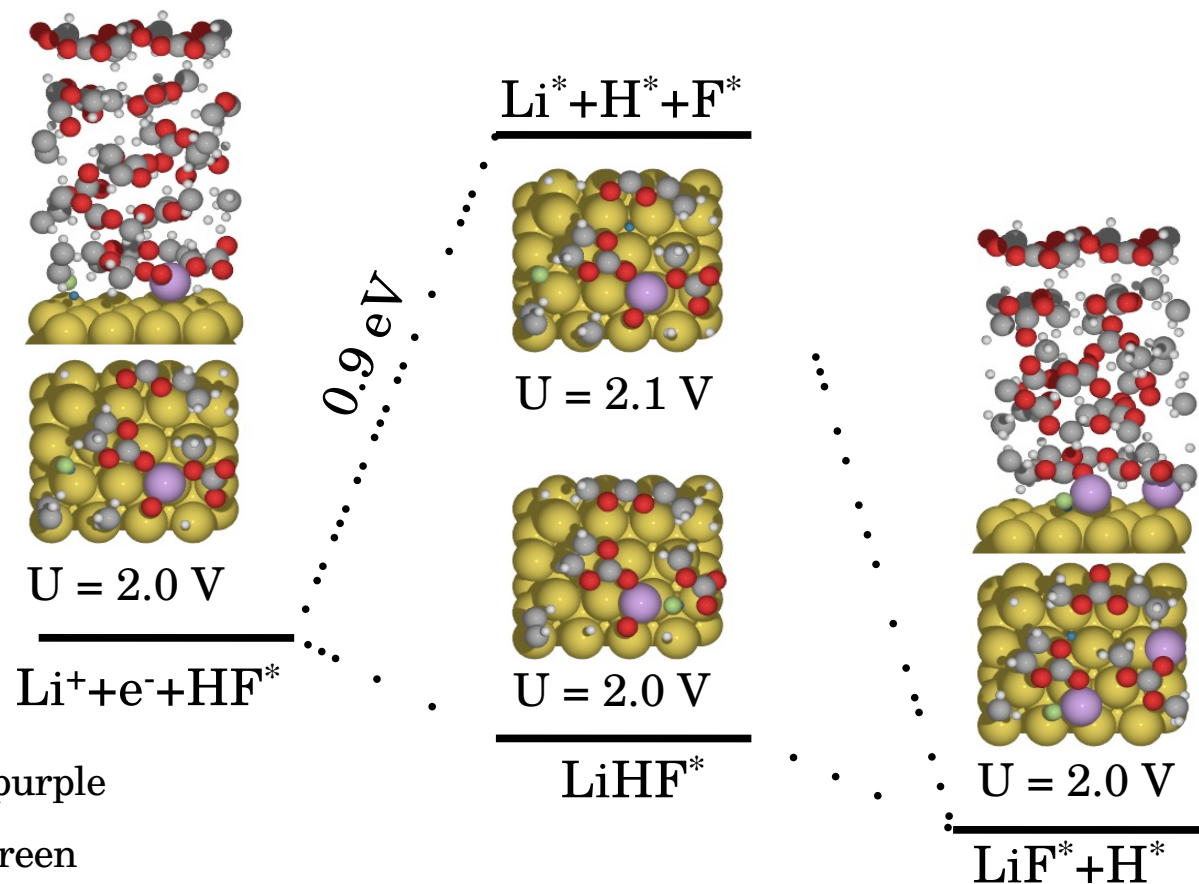


Li-adsorption Potential



Reaction Paths on Au(111) in LP57

Note: the potential should be kept constant during the reaction and equal to the Li- adsorption potential!

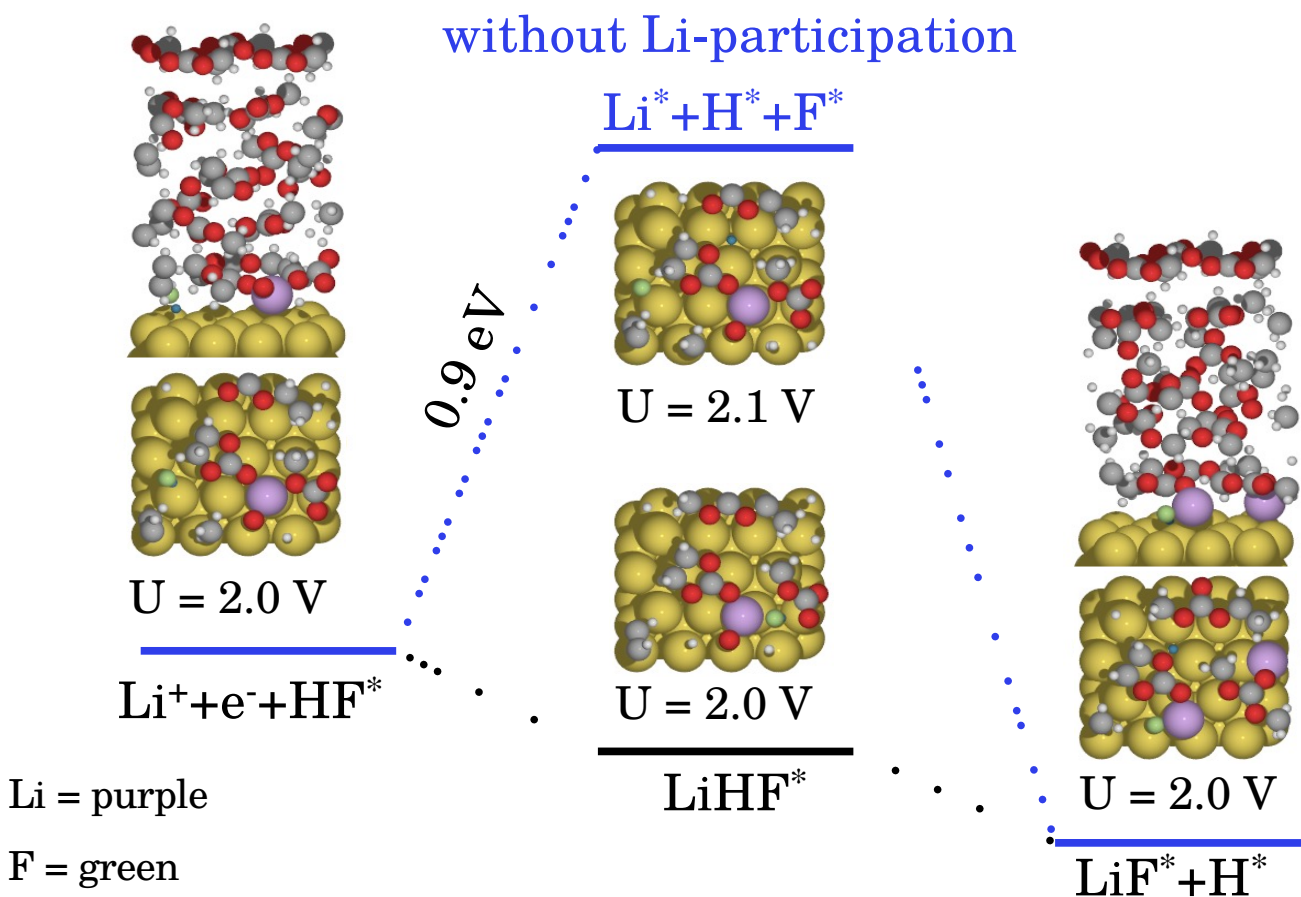


Strmcnik, Castelli, ..., Rossmeisl, and Markovic, *Nature Catalysis* **1**, 255 (2018)

Castelli, Zorko, ..., Strmcnik, and Rossmeisl, *Chem. Sci.* **11**, 3914 (2020)

Reaction Paths on Au(111) in LP57

Note: the potential should be kept constant during the reaction and equal to the Li- adsorption potential!



Li = purple

F = green

H = white and blue (from HF)

O = red

C = gray

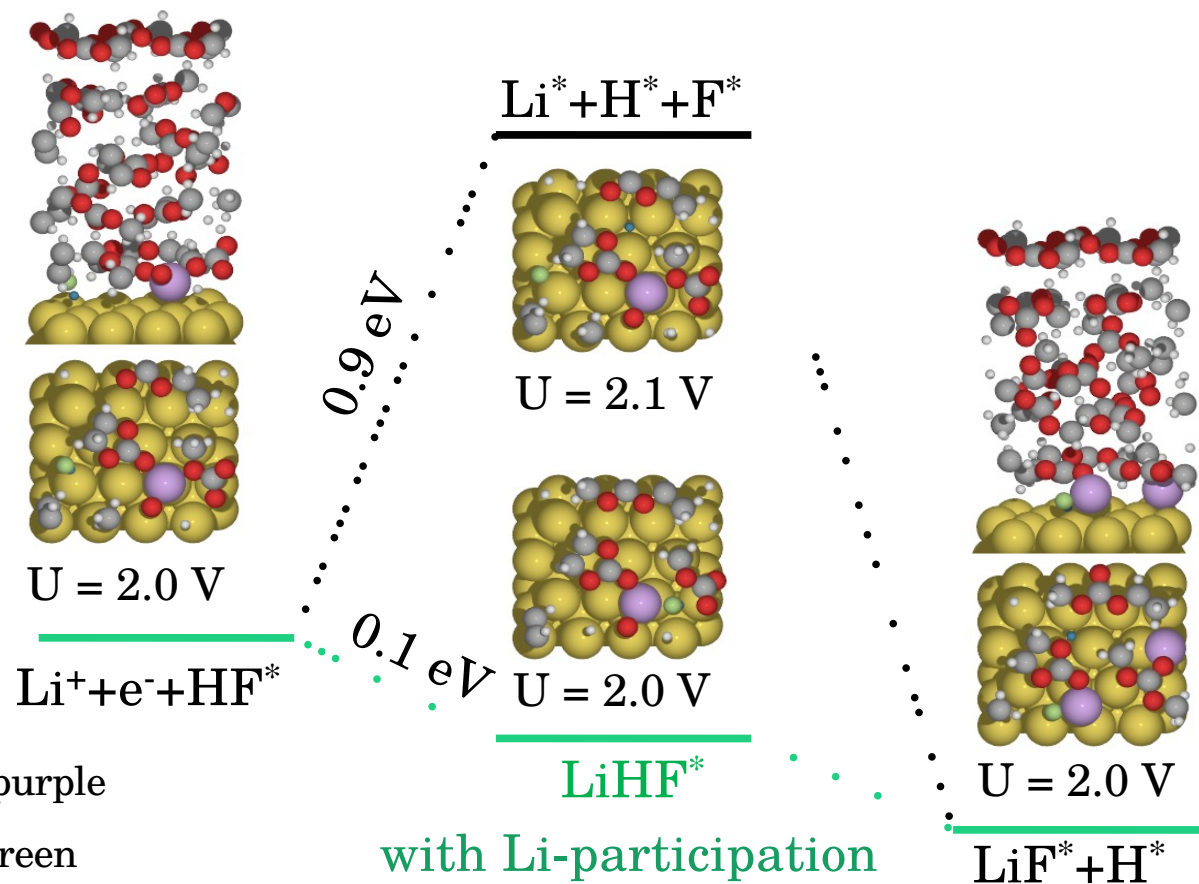
Au = yellow

Strmcnik, Castelli, ..., Rossmeisl, and Markovic, *Nature Catalysis* **1**, 255 (2018)

Castelli, Zorko, ..., Strmcnik, and Rossmeisl, *Chem. Sci.* **11**, 3914 (2020)

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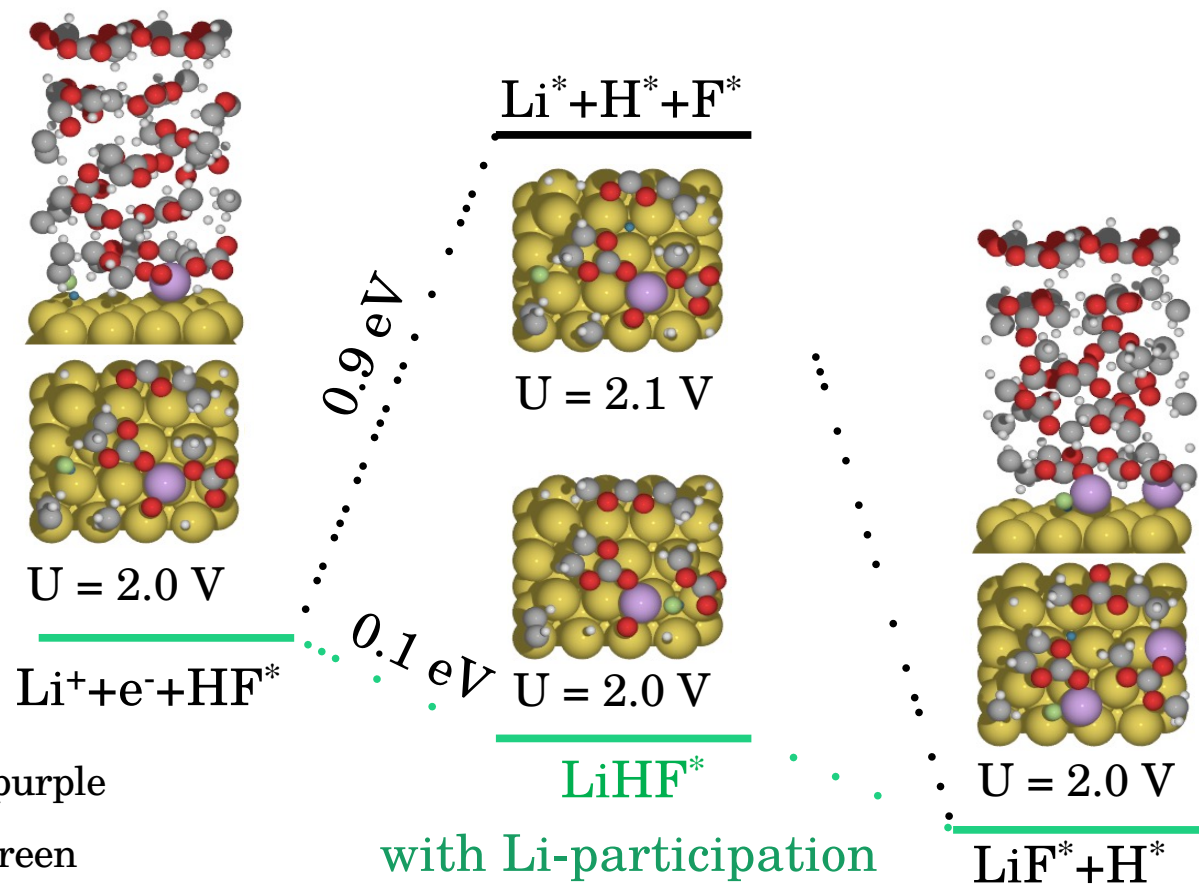
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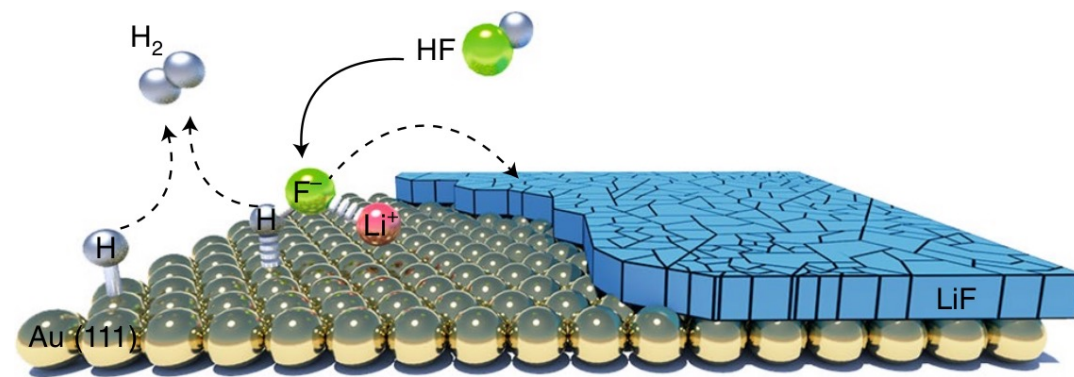
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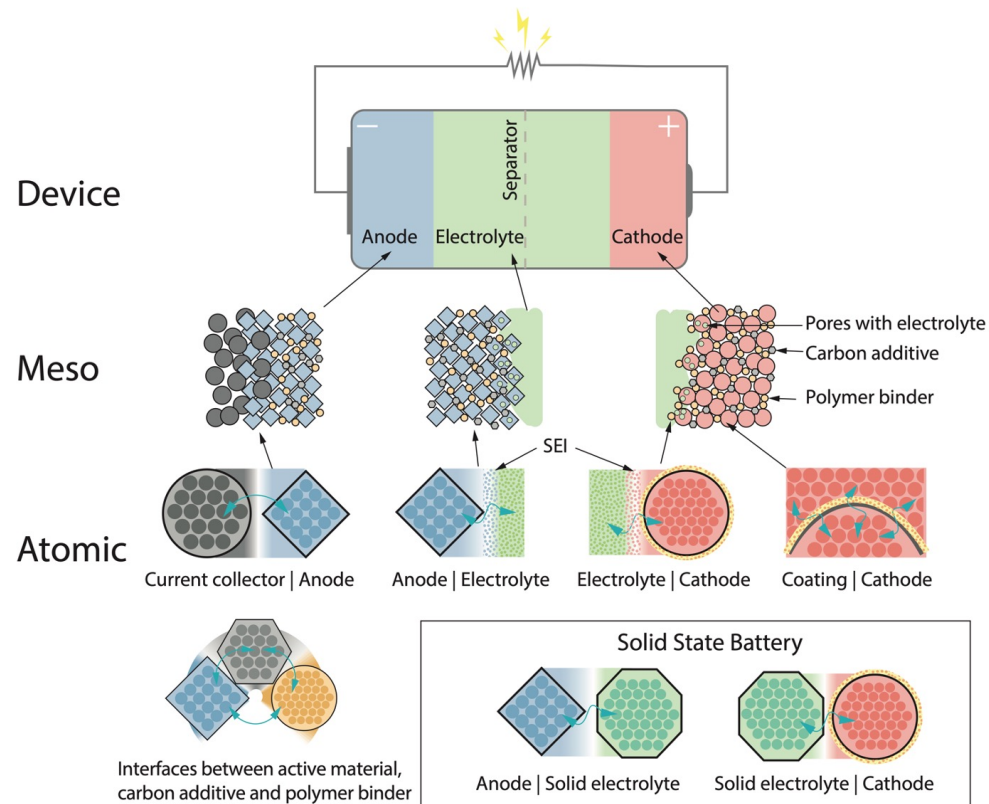
Suggested mechanism



Strmcnik, Castelli, ..., Rossmeisl, and Markovic, *Nature Catalysis* **1**, 255 (2018)

Castelli, Zorko, ..., Strmcnik, and Rossmeisl, *Chem. Sci.* **11**, 3914 (2020)

DTU Steps for a Possible Workflow



Inspired by the Atomic Scale Recipes (ASR), Go modular!

- 1) Workflow for adsorption energies
- 2) Workflow for MDs
- 3) Workflow for reaction paths/NEBs

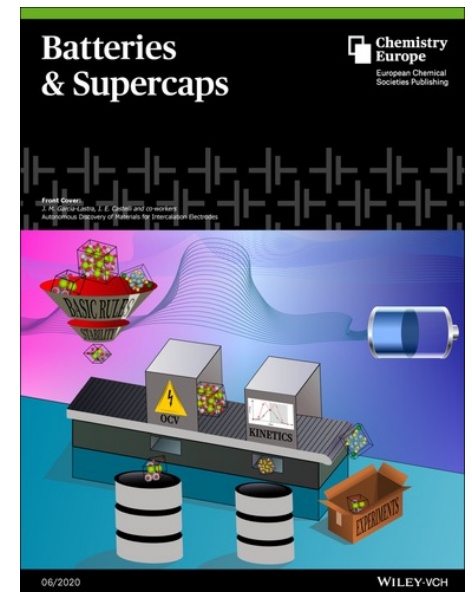
DTU Conclusions

- Accelerated design and investigation of 1D nanotubes (single-, multi-walls, various prototypes, from stability, to electronic and catalytic properties)
- Autonomous workflow for battery materials (intercalation and solid state electrodes)
- Workflows for interfaces... are we there yet? Let's go modular

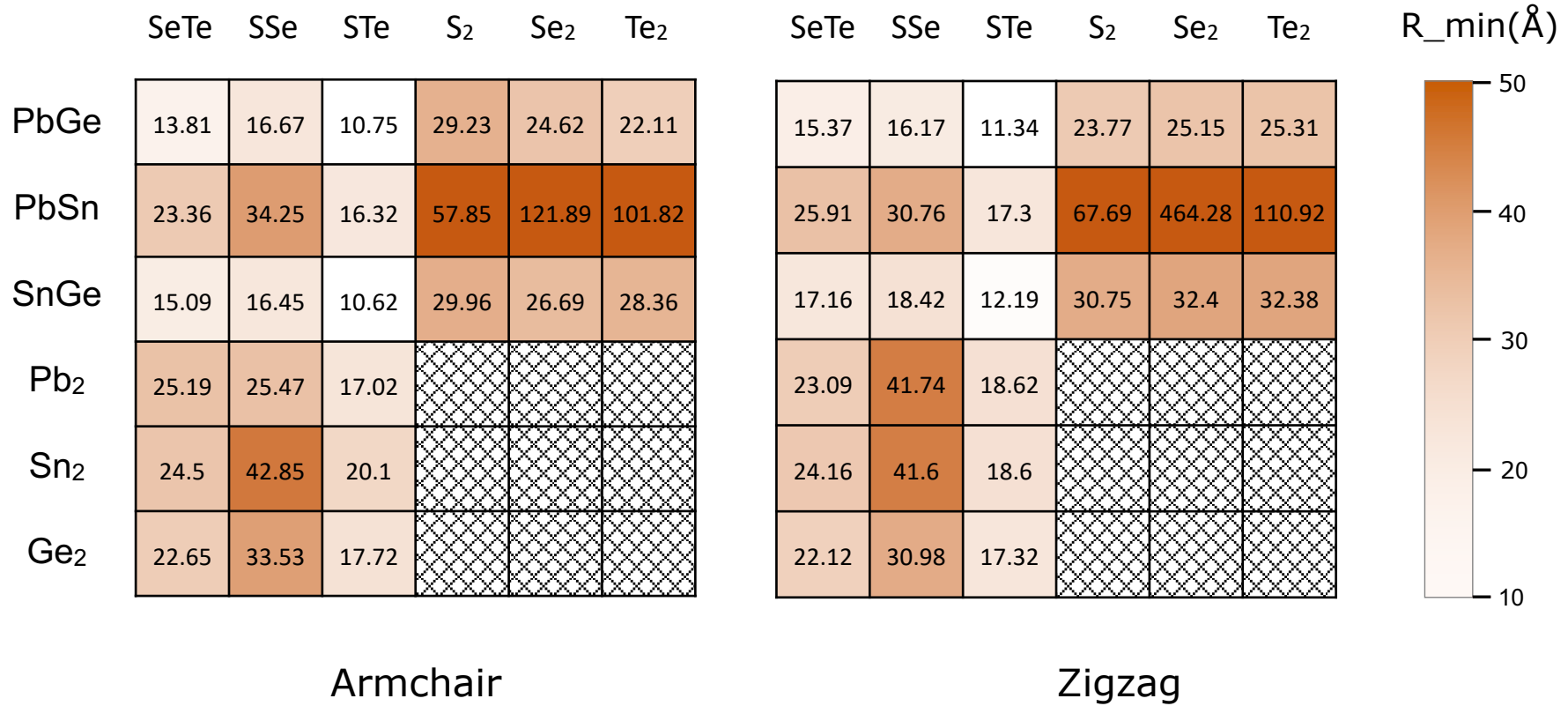
Acknowledgements

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- **Argonne National Laboratory (USA):** Dusan Strmcnik, Nenad Markovic
- **BMW Group (D):** Konstantinos Antonopoulos, Filippo Maglia

Contact: ivca@dtu.dk

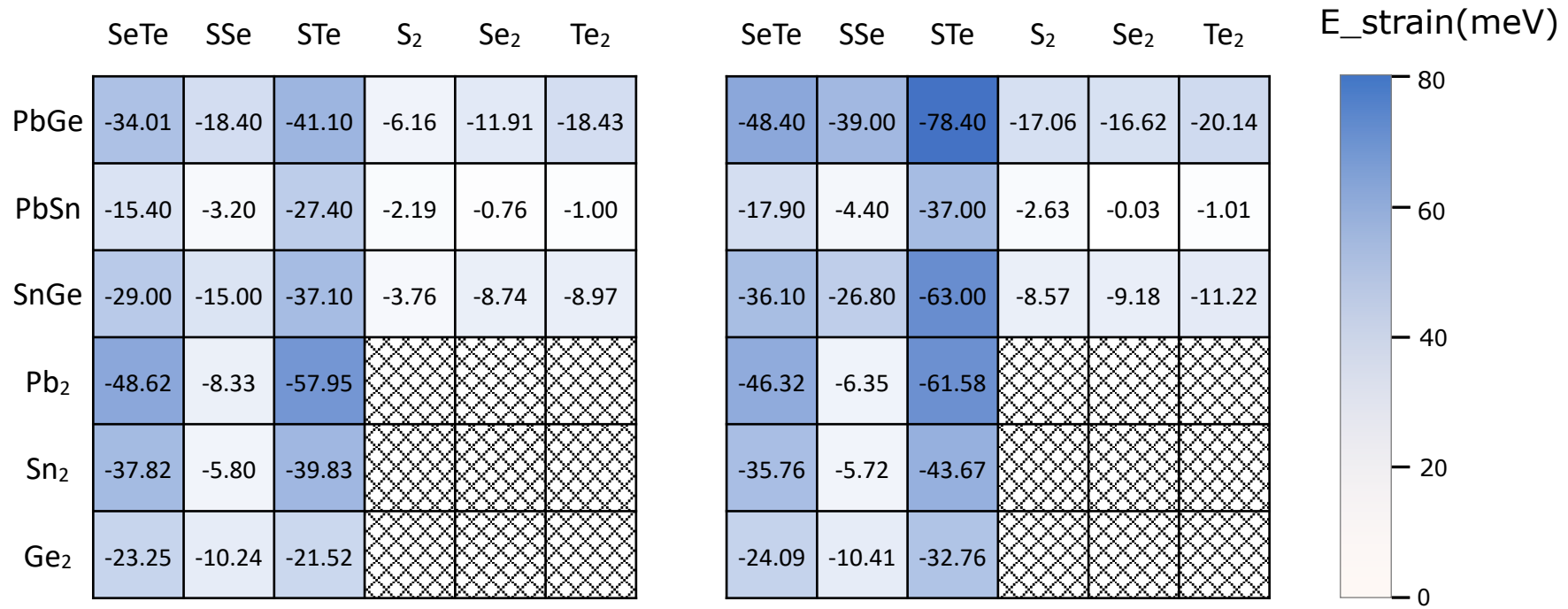


Heatmaps of radiuses of Janus nanotubes



- The GeSPbTe nanotube shows the smallest R_{\min} .

Heatmaps of E_strain of Janus nanotubes



- The GeSPbTe nanotube with the zigzag direction shows the most negative E_strain.